



In-situ carbonatization of CO₂ - The Carbfix project in Iceland

D. Wolff-Boenisch (1), S.R. Gislason (1), E.H. Oelkers (2), W.S. Broecker (3), J.M. Matter (3), M. Stute (3), A. Stefansson (1), E. Gunnlaugsson (4), G. Björnsson (4), and G. Axelsson (5)

(1) Institute of Earth Sciences, University of Iceland, Askja, Sturlugata 7, 101 Reykjavik, Iceland, (2) LMTG-Université de Toulouse-CNRS-IRD-OMP, 14 Avenues Edouard Belin, 31400 Toulouse, France, (3) Lamont-Doherty Earth Observatory, 61 Route 9W, Palisades, NY 10964, USA, (4) Reykjavik Energy, Baejarhalsi 1, 110 Reykjavik, Iceland, (5) ISOR, Iceland Geosurvey, Grensasvegur 9, 108 Reykjavik, Iceland

Among commonly proposed CO₂ storage techniques, the injection of anthropogenic CO₂ into deep geological formations is quite promising due their large potential storage capacity. Finding a storage solution that is long lasting, thermodynamically stable and environmentally benign would be ideal. In-situ mineralization and storage of carbon dioxide as solid carbonate in basaltic rocks may provide such an ideal solution. In nature, the carbonatization of basaltic rocks occurs in a variety of well-documented settings, e.g. geothermal and deep ocean vent systems. In order to assess the feasibility of in-situ CO₂ mineralization in basaltic rocks, a field-scale project, CarbFix, is scheduled to inject CO₂ in spring of 2009 at Hellisheiði, Iceland (www.carbfix.com). The injection site is adjacent to a new geothermal power plant, which will supply up to 30,000 tons of CO₂ per year for injection into subsurface basalts at a depth of 400–700 m and a temperature of ca. 30°C. The Hellisheiði site was chosen for several reasons, including a local source of CO₂, the availability of several strategically located wells for monitoring the chemical evolution of the groundwater, and the proximity to infrastructure through Reykjavik Energy, a CarbFix partner. CO₂ will be injected at a partial pressure of 25 bars and will be dissolved in water during injection. Model simulations, natural analogues and experimental work suggest that the CO₂ charged waters will react with the basalt and form carbonate minerals. Results of this project will thus be used to optimize the in-situ carbonatization process of CO₂.