



## Hydrochemistry and origin of CO<sub>2</sub> gas and noble gas of carbonated mineral water in the Gyeongbuk-Gangwon Province, South Korea

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Hydrochemical, carbon isotopic ( $\delta^{13}\text{CDIC}$ ) analyses of 11 samples, and noble gas isotopic analyses of 8 samples collected in the Gangwon and the Gyeongbuk area of South Korea were carried out to elucidate hydrochemical characteristics and to interpret the source of noble gases and CO<sub>2</sub> gas.

The carbonated mineral waters show a weak acidic pH between 5.59 and 6.04. An electrical conductivity of carbonated mineral waters ranges from 302 to 864  $\mu\text{S}/\text{cm}$ . The chemical composition of all carbonated mineral waters can be grouped into only one type such as Ca-HCO<sub>3</sub>. A high content of Fe and Mn in carbonated mineral waters exceeds a regulation limit of drinking water.

The  $\delta^{13}\text{CDIC}$  values of carbonated mineral waters show the range of -5.30 [U+FF5E]-2.84 ‰. This range indicates that the carbon of carbonated mineral waters is mainly supplied from a deep-seated source and partly from an inorganic carbonate source. The 3He/4He ratios of the carbonated mineral waters show the range of  $1.51 \times 10^{-6}$  to  $6.45 \times 10^{-6}$ .

The carbonated mineral waters on the 3He/4He and 4He/20Ne diagram are plotted into three groups: deep seated area such as mantle source, atmospheric area, and air-mantle mixing area. A wide range of 4He/20Ne ratios is observed ( $0.036 \times 10^{-6}$  to  $1.76 \times 10^{-6}$ ), showing evidence that while radiogenic 4He is dominant in these water samples, He of mantle-origin is also supplied to these waters. It is estimated that supply of CO<sub>2</sub> gas and noble gas of a deep-seated source into carbonated waters is closely related to geologic structures such as fault and geologic boundary.

Key words: carbonated mineral waters, hydrochemical composition, carbon isotope, 3He/4He, deep-seated origin