

Laboratory spectra of N₂ and CH₄ mixtures: Applications to Observations of TNOs

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1. Introduction

CH₄-ice has multiple low temperature phases that have distinct spectra [1]. In the laboratory, infrared spectra CH₄-ice diluted in N₂ have absorptions that are different in shape, location, and strength compared to those of pure CH₄-ice [2]. However, the full range of dilutions and changes with temperature have not been consistently explored for α -N₂-ices. Although, β -N₂ ices are common throughout the outer solar system, there are also surfaces where the temperatures are low enough for α -N₂ to survive. Recent observations of Eris [3] demonstrate that the surfaces of TNOs may consist of a complex mixture of ices of different phases and dilutions.

2. Figures

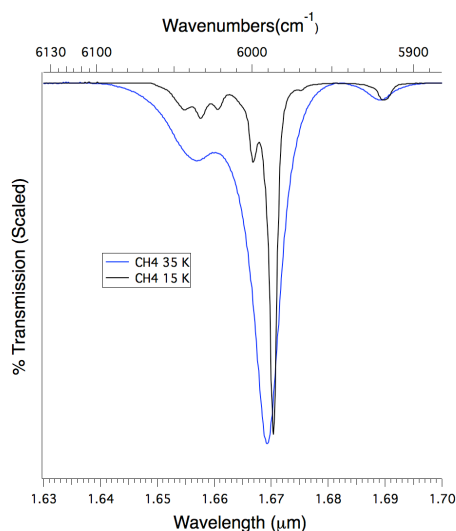


Figure 1. Transmission spectra of pure CH₄-ices at 15 and 35 K, demonstrating changes in the absorption near 1.67 μ m. The individual spectra were scaled to similar strength for clarity.

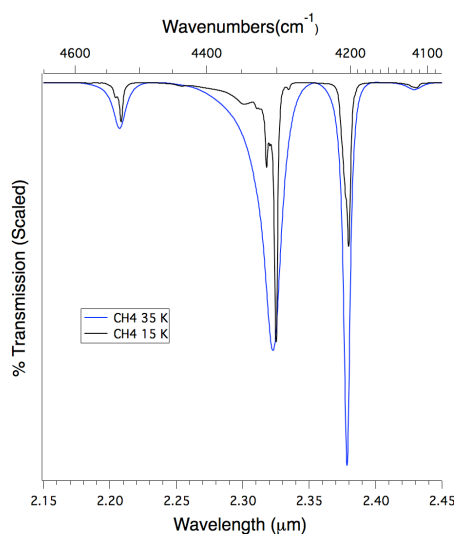


Figure 2. Transmission spectra of pure CH₄-ices at 15 and 35 K, depicting changes in the absorptions near 2.2, 2.32, and 2.38 μ m. The individual spectra were scaled to similar strength for clarity.

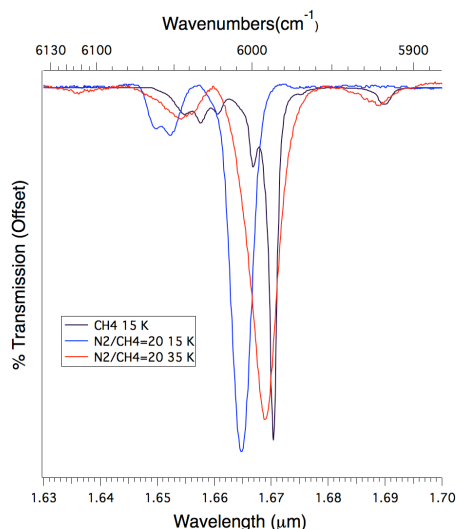


Figure 3. Transmission spectra of pure CH₄-ices at 15 and a mixture of N₂/CH₄=20 at 15 and 35 K, depicting changes in the absorptions near 1.67 μm. The individual spectra were scaled to similar strength for clarity.

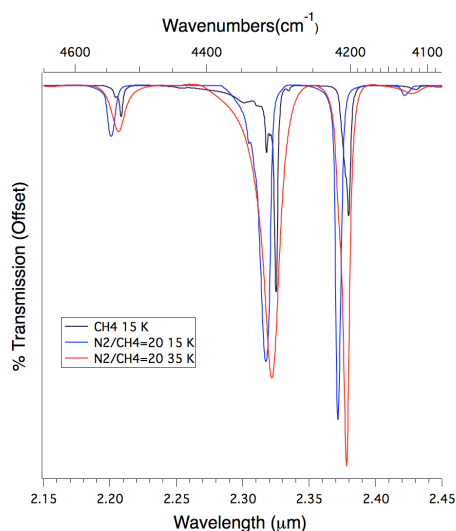


Figure 4. Transmission spectra of pure CH₄-ices at 15 and a mixture of N₂/CH₄=20 at 15 and 35 K, depicting changes in the absorptions near 2.2, 2.32, and 2.38 μm. The individual spectra were scaled to similar strength for clarity.

4. Summary and Conclusions

Figures 1 & 2 demonstrate shifts in absorptions in pure CH₄-ice. Note that all of the absorptions are shifted to shorter wavelength at higher temperature.

Figures 2 & 3 demonstrate the effect of diluting CH₄ in N₂-ice. There is a significant shift to shorter wavelength at 15 K, but that shift is lost at higher temperature.

We have demonstrated multiple changes in the location of CH₄ bands in both pure samples and dilutions in N₂. We will present results detailing the changes in absorption profile and location as a function of temperature and dilution for all wavelength ranges relevant to observations of TNOS.

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References

1. Grundy, W.M., B. Schmitt, and E. Quirico, *The Temperature-Dependent Spectrum of Methane Ice 1 between 0.7 and 5 μm and Opportunities for Near-Infrared Remote Thermometry*. *Icarus*, 2002. **155**: p. 486-496.
2. Quirico, E. and B. Schmitt, *Near-Infrared spectroscopy of simple hydrocarbons and carbon oxides in solid N₂ and as pure ices: Implications for Triton and Pluto*. *Icarus*, 1997. **127**: p. 354-378.
3. Cook, J.C., et al., *Analysis of High Resolution Spectra of Eris: Possible Evidence for Cold Phase CH₄ Ice*. This Meeting, 2011.