



Synergistic Study of Hydrocarbon Photochemistry in the Laboratory and Planetary Atmospheres

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A synergistic study of hydrocarbon photochemistry in the laboratory and planetary atmospheres has been carried out using the Caltech/JPL KINETICS photochemical model and laboratory measurements from Adamkovics and Boering (2003). The laboratory simulations provide the data for the time-evolution of gaseous species such as H₂, C₂H₂, C₂H₄, C₂H₆, C₃H₄, C₄H₂ and C₄H₁₀ during UV irradiation of CH₄. We apply forward and adjoint models to analyze the experiments. Different photochemical schemes (e.g., Moses et al. 2000, 2005) are compared and modified to reproduce the laboratory results. We first test the full sensitivity of the model results to *all* chemical kinetics using the adjoint model and show that the abundances of C₂H₂, C₂H₄, C₂H₆, and C₄H₁₀ can be well reproduced while that of C₄H₂ is underestimated by 1-2 orders of magnitude. The abundance of C₃H₄ is underestimated with Moses et al. (2000) kinetics but overestimated with Moses et al. (2005) kinetics. This suggests a major gap in our understanding of chemical pathways to higher hydrocarbons. We next examine higher order hydrocarbon chemistry (>C₂). In this case, we assume that all rate coefficients for the chemistry of C₁ and C₂ hydrocarbons remain invariant in the adjoint optimization. Better agreement is achieved, but complete agreement remains elusive. Further laboratory measurements are urgently needed to constrain the pathways. The implications for modeling the atmospheres of Titan and the giant planets (e.g., Jupiter) are discussed.