Geophysical Research Abstracts Vol. 12, EGU2010-10077-1, 2010 EGU General Assembly 2010 © Author(s) 2010



A super-aluminous phase D stable within subdacting oceanic crust in the Earth's lower mantle.

Tiziana Boffa Ballaran, Daniel J. Frost, Martha Pamato, and Nobuyoshi Miyajima Bayerisches Geoinsitut, Bayreuth, Germany (tiziana.boffa-ballaran@uni-bayreuth.de)

High-pressure experiments have identified a number of dense hydrous magnesium silicate (DHMS) phases that are potentially stable in the cool interiors of subducting slabs. These phases may be responsible for the subduction of H_2O deep into the Earth's interior. The highest pressure DHMS phase yet identified has been termed phase D, with the ideal formula $MgSi_2H_2O_6$. Phase D is stable at pressures equivalent to those in the lower mantle but dehydrates at temperatures at or above approximately 1400 °C and would therefore not be stable in the ambient lower mantle. This, combined with the recognized low solubility of H_2O in nominally anhydrous minerals that make up the bulk of the lower mantle, implies that H_2O might be forced into fluid or melt phases in the lower mantle.

A new Al-rich form of the dense hydrous magnesium silicate phase D has been synthesized at $1600\,^{\circ}\text{C}$ and $26\,\text{GPa}$. Initial multianvil experiments employing a simplified hydrous MORB bulk composition in the system H-Fe-Mg-Si-Al-O produced silicate perovskite with the formula $Mg_{0.59}Fe_{0.42}Al_{0.34}Si_{0.65}O_3$ coexisting with a minor amount of an Al-rich hydrous phase. The composition of this phase was used to derive a second starting composition comprised of oxides and hydroxides, which was equilibrated at $1600\,^{\circ}\text{C}$ and $26\,^{\circ}\text{GPa}$ in a Pt capsule. The run produced single crystals with the formula $H_{1.9}Mg_{0.2}Fe_{0.2}Al_{1.8}SiO_6$ of up to $100\,^{\circ}\text{microns}$ in largest edge length. The unit-cell lattice parameters determined with a Huber four-circle diffractometer using $20\,^{\circ}\text{reflections}$ are the following: $a = 4.7704(2)\,^{\circ}\text{Å}$, $c = 4.2896(2)\,^{\circ}\text{Å}$, $c = 4.2896(2)\,^{\circ}\text{Å}$, $c = 4.2896(2)\,^{\circ}\text{Å}$, $c = 4.2896(2)\,^{\circ}\text{A}$, $c = 4.2896(2)\,^{\circ}$

The high temperature stability of this phase and its coexistence with phases in a simplified MORB bulk composition are strong grounds to argue that it could be a host for H_2O in the Earth's lower mantle. H_2O in the ambient lower mantle may therefore become preferentially concentrated in recycled portions of oceanic crust.