



Isotope effect in the formation of H₂ from H₂CO studied at the atmospheric simulation chamber SAPHIR

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Formaldehyde of known, near-natural isotopic composition was photolyzed in a large photochemical reactor under ambient conditions. The isotopic composition of the product H₂ was used to determine the isotope effects in formaldehyde photolysis. The experiments are sensitive to the molecular photolysis channel, and the radical channel has only a second order effect and can thus not be derived with high precision. The molecular channel kinetic isotope effect, the ratio of photolysis frequencies $j(\text{HCHO} \rightarrow \text{CO} + \text{H}_2)/j(\text{HCDO} \rightarrow \text{CO} + \text{HD})$ under tropospheric conditions is determined to be $\text{KIE}_{\text{mol}}=1.63 \pm 0.03$. Combining this result with the total KIE from a recent relative rate experiment, it is likely that KIE_{mol} and KIE_{rad} are not as different as described previously in the literature.