



## Peculiarities of CO<sub>2</sub> sequestration in the Permafrost area

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Natural gas and gas-condensate accumulations in North of Western Siberia contain an admixture of CO<sub>2</sub> (about 0.5-1.0 mol.%). Recently, the development and transportation of natural gas in the Yamal peninsula has become of interest to Russian scientists. They suggest liquifaction of natural gas followed by delivery to consumers using icebreaking tankers. The technique of gas liquefaction requires CO<sub>2</sub> to be absent from natural gas, and therefore the liquefaction technology includes the amine treatment of gas. This then leads to a problem with utilization of recovered CO<sub>2</sub>.

It is important to note, that gas reservoirs in the northern part of Russia are situated within the Permafrost zone. The thickness of frozen sediment reaches 500 meters. That is why one of the promising places for CO<sub>2</sub> storage can be gas-permeable collectors in under-permafrost horizons. The favorable factors for preserving CO<sub>2</sub> in these places are as follows: low permeability of overlying frozen sediments, low temperatures, the existence of a CO<sub>2</sub> hydrate stability zone, and the possibility of sequestration at shallow depths (less then 800-1000 meters). When CO<sub>2</sub> (in liquid or gas phase) is pumped into the under-permafrost collectors it is possible that some CO<sub>2</sub> migrates towards the hydrate stability zone and hydrate-saturated horizons can be formed. This can result on the one hand in the increase of effective capacity of the collector, and on the other hand, in the increase of isolating properties of cap rock. Therefore, CO<sub>2</sub> injection sometimes can be performed without a good cap rock.

In connection with the abovementioned, to elaborate an effective technology for CO<sub>2</sub> injection it is necessary to perform a comprehensive experimental investigation with computer simulation of different utilization schemes, including the process of CO<sub>2</sub> hydrate formation in porous media. There are two possible schemes of hydrate formation in pore medium of sediments: from liquid CO<sub>2</sub> or the gas. The pore water in the sediment may be either in frozen or liquid states.

To study these processes, an experimental investigation of hydrate formation kinetics from liquid and gaseous CO<sub>2</sub> has been performed using the method of NMR imaging\*. Experiments were made with samples of quartz sand (particles' diameter 0,21-0,297mm) with different water saturation in the range of temperatures between -3 and +8oC and pressures between 3 and 6 MPa. The experiments performed revealed the main regularities of hydrate accumulation from liquid CO<sub>2</sub> in sediment.

The influence of temperature on the rate of pore hydrate growth was analyzed. For example, the rate of hydrate growth at +7.2oC was 6 times smaller then at -3 . Fast hydrate formation from liquid CO<sub>2</sub> was observed in sand samples with water saturation below 20-30%. With an increase in water saturation to 50%, the rate of hydrate formation decreased significantly, and when water saturation was 60% or more, nucleation was not observed during the time of the experiment (1-3 days). Experimental results revealed that pressure variation in the range between 4 and 6 MPa does not have any influence on the kinetics of hydrate formation from liquid CO<sub>2</sub>. Comparison of kinetics of hydrate formation from liquid and gas CO<sub>2</sub> showed that hydrate accumulation is faster from gas CO<sub>2</sub> then from liquid CO<sub>2</sub>. Thus, 50% of pore water that reacted with liquid CO<sub>2</sub> transformed into hydrate in 0.8 hours after nucleation, and when reacted with CO<sub>2</sub>-gas, it transformed in 0.3 hours.

The completed experiments allowed us to consider the peculiarities of hydrate formation and filtration of liquid and gaseous CO<sub>2</sub> towards the hydrate stability zone, which is important to take into account during the elaboration of industrial techniques of CO<sub>2</sub> injection in under-permafrost collectors.

\* Experiments have been made in the laboratory of NRC of Canada.