



## **Uptake of NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub> to Saharan dust, ambient urban aerosol and soot: A relative rate study**

Mingjin Tang, Jim Thieser, Gerhard Schuster, and John N. Cowley

Max Planck Institute for Chemistry, Atmospheric Chemistry Department, Germany (mingjintang@gmail.com)

The uptake of NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub> to Saharan dust, ambient aerosols and soot was investigated using a novel and simple relative rate method with simultaneous detection of both NO<sub>3</sub> and N<sub>2</sub>O<sub>5</sub>. The use of cavity ring down spectroscopy to detect both trace gases enabled the measurements to be carried out at low mixing ratios (< 500 pptv or  $1 \times 10^{10}$  molecules cm<sup>-3</sup>). The uptake coefficient ratio,  $\gamma(\text{NO}_3) / \gamma(\text{N}_2\text{O}_5)$ , was determined to be  $0.9 \pm 0.3$  for Saharan dust, independent of relative humidity and exposure time. Ambient (urban) aerosols showed a very limited capacity to take up N<sub>2</sub>O<sub>5</sub> but were reactive towards NO<sub>3</sub> with  $\gamma(\text{NO}_3) / \gamma(\text{N}_2\text{O}_5) > 15$ . A value of  $\gamma(\text{NO}_3) / \gamma(\text{N}_2\text{O}_5) \approx 1.5 - 3$  was obtained when using candle generated soot. The relative rate obtained for Saharan dust can be placed on an absolute basis using our recently determined value of  $\gamma(\text{N}_2\text{O}_5) = 1 \times 10^{-2}$  to give  $\gamma(\text{NO}_3) = 9 \times 10^{-3}$ , which is significantly smaller than the present literature value. With the present uptake coefficient, reaction with mineral dust will generally not contribute significantly to NO<sub>3</sub> loss in the atmosphere or to the nitration of mineral dust.