



HO_x recycling via HO₂ + isoprene-derived RO₂: Direct experimental evidence

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Recent results from field, lab and theoretical work suggest that the atmospheric oxidation of isoprene and other important NHMCs is still inadequately represented in atmospheric models. In low NO_x conditions organic peroxy radicals (RO₂) – the key intermediates in these processes – react mainly with HO₂. Depending on the substitution pattern of the RO₂ these reactions either terminate or propagate radical chemistry. The propagating reaction channels recycle HO_x by direct production of OH and thus contribute to the maintenance of the atmosphere's oxidative capacity.

In this work for the first time OH yields for reactions of HO₂ with RO₂ formed during isoprene oxidation were measured directly. We deployed a laser-based experiment with multidagnostic spectroscopic tools including direct detection of OH, HO₂ and RO₂ as well as on-line determination of precursor concentrations. Branching ratios were derived by numerical simulations of the photochemical systems. For the reactions of HO₂ with CH₃C(O)O₂ and HOCH₂C(O)O₂ substantial OH yields were determined.