

Pernitric acid in a coastal Antarctic boundary layer – a wintertime study



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INTRODUCTION

1) PERNITRIC ACID (PNA, i.e. HO₂NO₂)

PNA couples HOx and NOx chemistry through its production and loss reactions:

 $HO_2 + NO_2 + M \leftrightarrow HO_2NO_2 + M$ $HO_2 NO_2 + hv \rightarrow HO_2 + NO_2 \text{ or } OH + NO_3$ $HO_2NO_2 + OH \rightarrow H_2O + NO_2 + O_2$

Further, PNA can strongly adsorb to ice/snow surfaces, so has the potential to act as NOx/HOx reservoirs, either short or longer-term.

 $HO_2NO_2 \rightarrow deposition$

The only previous measurements in Antarctica were at South Pole during the summer. Then, significant amounts of boundary layer PNA were detected (<5 to 54 pptv, mean 25 pptv), with lifetime controlled predominantly by dry deposition and thermal decomposition (Slusher et al., 2002). No wintertime observations have previously been made.

2) MEASUREMENT LOCATION AND INSTRUMENTATION

PNA was measured during the 2007 overwinter campaign at the British Antarctic Survey station, Halley, in coastal Antarctica (see Fig. 1).

Observations were made at the Clean Air Sector Laboratory (CASLab) (Fig. 2) using a Chemical ionisation Mass Spectrometer (CIMS) using the SF_6 technique described by Slusher et al. (2001).

The CIMS concurrently measured nitric acid (HNO₃) and measurements of NO and NO₂ were obtained by a chemiluminescence analyser.

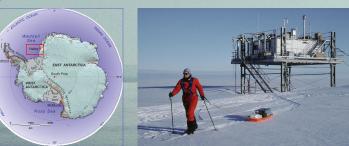
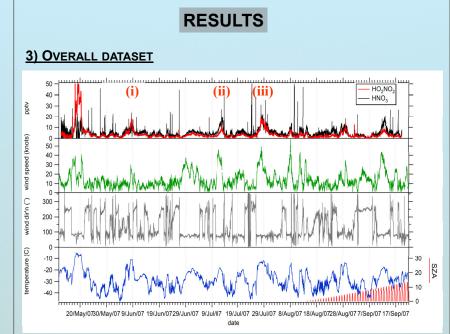
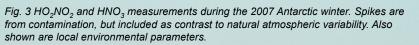


Fig. 1 Map of Antarctica Fig. 2 The Clean Air Sector Laboratory (CASLab)





- PNA is present in the wintertime boundary layer at Halley!
- · Considerable variability is evident, with mixing ratios varying between instrumental detection limits and ~20 pptv;
- Generally, the same variability patterns reflected in HNO₃ mixing ratios;
- The majority of these measurements were made during polar night, so little or no sunlight, as shown by SZA trace;

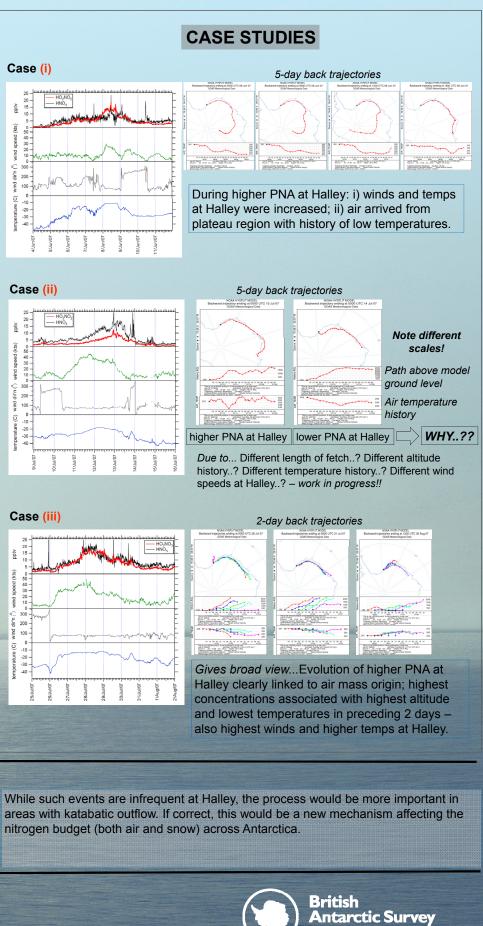
• No in situ production of HO₂NO₂, or loss via photolysis or OH are possible. ([NO] and [NO₂] were below detection limit throughout). [PNA] must be controlled by transport and physical exchange processes;

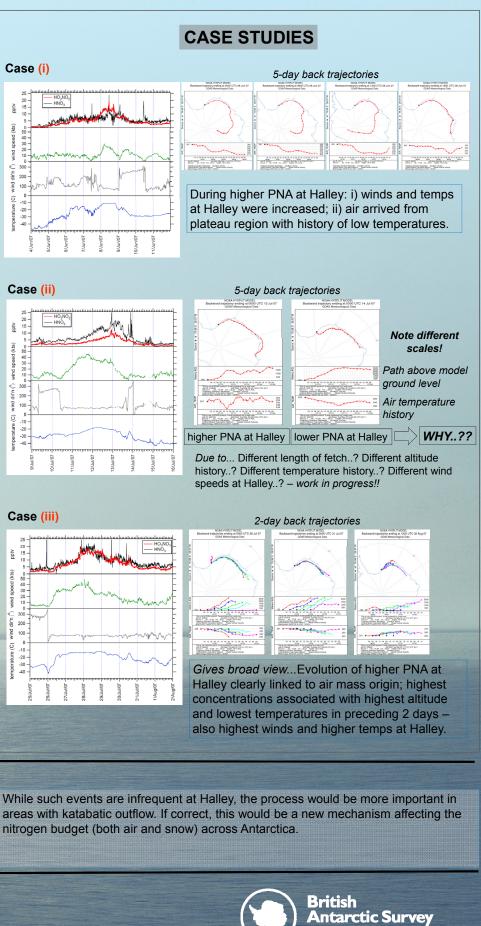
• At first glance, higher mixing ratios of both HO₂NO₂ and HNO₃ associated with strong winds and warmer temperatures (but see next panel);

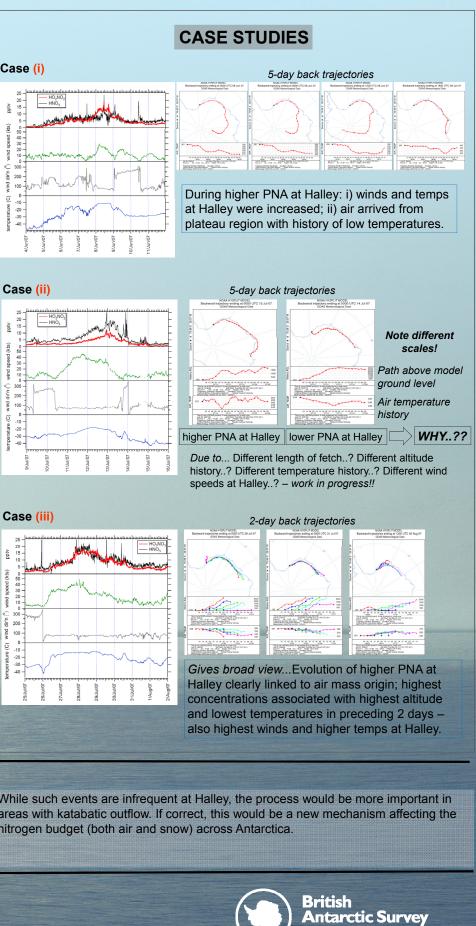
• 3 case studies during polar night examined in more detail with air parcel history explored using Hysplit trajectory model (Draxler and Rolf, 2003)

References:

Draxler, R.R. and Rolph, G.D., 2003. NOAA Air Resources Laboratory, Silver Spring, MD.; Li et al., JGR (101), 6795, 1996; Slusher et al., GRL (28), 3875, 2001; Slusher et al., GRL (29), doi:10.1029/2002GL015703, 2002







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CONCEPTUAL MODEL....

All case studies shown here imply a link between enhanced coastal PNA and air transport from the Antarctic Plateau. But air temperatures at the coast increase with increasing PNA which is counter-intuitive given that a major loss of PNA is thermal decomposition.

However, we know that PNA can adsorb onto ice surfaces and desorb as temperatures rise (Li et al., 1996); thermal desorption from the snowpack would thus be associated with increasing temperatures. Our data suggest that PNA from Plateau snow is feeding into coastal areas under specific environmental conditions.