PREDICTING THE FORMATION AND THE DISPERSION OF TOXIC COMBUSTION PRODUCTS FROM THE FIRES OF DANGEROUS SUBSTANCES

VÁCLAV NEVRLÝ ^{1,2}, PETR BITALA ¹, PAVEL DANIHELKA ¹, PAVEL DOBEŠ ¹, JAKUB DLABKA ¹, TOMÁŠ HEJZLAR¹, BARBORA BAUDIŠOVÁ¹, DALIBOR MÍČEK¹, ZDENĚK ZELINGER³



 \bigcirc

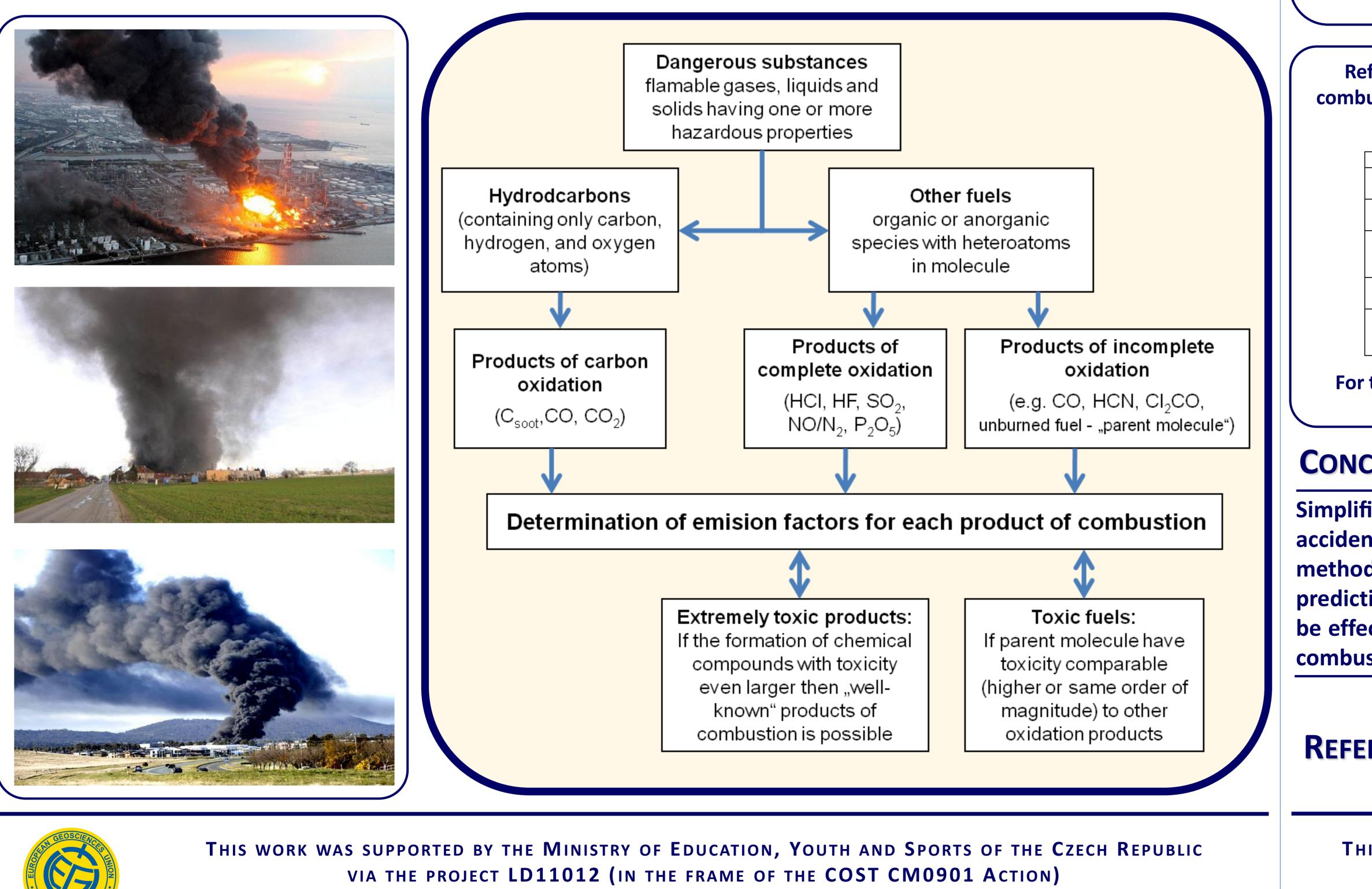
 (\mathbf{i})

¹ VŠB – TECHNIKAL UNIVERSITY OF OSTRAVA, FACULTY OF SAFETY ENGINEERING, LUMÍROVA 13, CZ-700 30 OSTRAVA-VÝŠKOVICE, CZECH REPUBLIC (VACLAV.NEVRLY@VSB.CZ)

² INSTITUT OF THERMODYNAMICS OF ACADEMY OF SCIENCE OF THE CZECH REPUBLIC, V. V. I., PRAHA, CZECH REPUBLIC ³ HEYROVSKY INSTITUTE OF PHYSICAL CHEMISTRY OF ACADEMY OF SCIENCE OF THE CZECH REPUBLIC, V. V. I., PRAHA, CZECH REPUBLIC

ABSTRACT:

Natural events, such as wildfires, lightning or earthquakes represent a frequent trigger of industrial fires involving dangerous substances. Dispersion of smoke plume from such fires and the effects of toxic combustion products are one of the reference scenarios expected in the framework of major accident prevention. Nowadays, tools for impact assessment of these events are rather missing. Detailed knowledge of burning material composition, atmospheric conditions, and other factors are required in order to describe quantitatively the source term of toxic fire products and to evaluate the parameters of smoke plume. Nevertheless, an assessment of toxic emissions from large scale fires involves a high degree of uncertainty, because of the complex character of physical and chemical processes in the harsh environment of uncontrolled flame. Among the others, soot particle formation can be mentioned as still being one of the unresolved problems in combustion chemistry, as well as decomposition pathways of chemical substances.





AND THE MINISTRY OF ENVIRONMENT OF THE CZECH REPUBLIC (PROJECT NO. SPII 1A10 45/70).

Acetamic

Acetone

Ace

Acety

Acety

Dichl

More then 150 dangerous chemicals identified (with qualitative assignement of possible combustion products):

	Phase (293K)					Number of atoms								Combustion products					
Name	Ι	g	S	Formula	Molar mass	С	Н	0	F	CI S	SI	NE	3r	 A) HCI	B) HBr	C) HF	D) HX	F) NO _x	
de, N-phenyl-	0	0	1	CH ₃ .CO.NH.C ₆ H ₅	135	8	9	1	0	0	0	1	0	0	0	0	0	1	
cyanohydrin	1	0	0	(CH ₃) ₂ .C(OH)(CN)	85	4	7	1	0	0	0	1	0	0	0	0	0	0	
etonitrile	1	0	0	CH ₃ .CN	41	2	3	0	0	0	0	1	0	0	0	0	0	0	
/l bromide	1	0	0	CH₃.CO.Br	123	2	3	1	0	0	0	0	1	0	1	0	0	0	
/l chloride	1	0	0	CH ₃ .CO.CI	78.5	2	3	1	0	1	0	0	0	1	0	0	0	0	
lorethene	1	0	0	CHCI:CHCI	97	2	2	0	0	2	0	0	0	1	0	0	0	0	
chloride	1	0	0	H2C:CHCI	62.4	2	3	0	0	1	0	0	0	1	0	0	0	0	

Emission factors for each combustion product were estimated (using simplifying assumptions) based on the balance of carbon and other elements (heteroatoms) present in the molecule of fuel.

Fuel:	C ₂ H ₃ CI				
Atoms in fuel molecule:	C		CI		
	2	3	1		
Molar weight of C in fuel molecule:	12		35,45 35,45		
Molar weight of fuel molecule:	62,45		55,45		
Products	C(s)+CO2+CO+HCI				
F _{carbon,fuel}	Mass fraction of carbon in fuel	0,384			
F _{carbon,soot}	Mass fraction of carbon in soot	1,000			
F _{carbon,CO}	Mass fraction of carbon in carbon monoxide	0,429			
F _{carbon,CO2}	Mass fraction of carbon in carbon dioxide	0,273			
F _{chlorine,fuel}	Mass fraction of chlorine in fuel	0,5676541			
F _{chlorine,HCl}	Mass fraction of chlorine in hydrogen chloride	0,9725652			
E _{soot}	Emission factor of soot particles	0,150	=>	150	g/kg
η_{soot}	E _{soot} /F _{carbon,fuel}	0,390			
G _{soot,C}	η_{soot} * $F_{carbon,soot}$	0,390			
ŋ co	$\eta_{CO} = 0.0014 + (0,37*\eta_{soot})$	0,146			
G _{CO,C}	$\eta_{CO} * F_{carbon,CO}$	0,062			
G _{CO2,C}	$1-(G_{CO,C}+G_{soot,C})$	0,547			
η _{CO2}	G _{CO2,C} /F _{carbon,CO2}	2,006			
G _{HCI,CI}	Yield of hydrogen chloride from chlorine in fuel	1,000			
ŋ нсі	G _{HCI,CI} /F _{chlorine,HCl}	1,028			
E _{co}	η _{CO} *F _{carbon,fuel}	0,056038	=>	56,03803	g/kg
E _{CO2}	η_{CO2} *F _{carbon,fuel}	0,7710675	=>	771,06754	g/kg
E _{HCI}	η_{HCl} *F chlorine, fuel	0,5836669	=>	583,66693	a/ka

Reference threshold (I and II) values for toxicity of individual combustion products were determined following the methodology for selection of acute exposure limits [3].

Publisher	Reference treshold I	Reference treshold II			
EU	AETL- 3a AETL-3b	AETL-2			
United States Environmental Protection Agency	AEGL-3	AEGL-2			
Ministère de l'Écologie, de l'Energie, du Développement durable et de la Mer	SPEL	SEI			
The American Industrial Hygiene Association	ERPG-3	ERPG-2			
Subcommittee on Consequence Assessment and Protective Actions	TEEL-3	TEEL-2			

For the multi-component mixture of toxic gases the concept of fractional effective dose (FED) [4] can be used.

CONCLUSION:

Simplified approach for estimating the emission factors from outdoor fires of dangerous chemicals, utilizable for major accident prevention and preparedness, was developed and the case study illustrating the application of the proposed method was performed. ALOFT-FT software tool based on large eddy simulation of buoyant fire plumes was employed for predicting the local toxic contamination in the down-wind vicinity of the fire. The database of model input parameters can be effectively modified enabling the simulation of the smoke plume from pool fires or jet fires of arbitrary flammable (or combustible) gas, liquid or solid.

k"). Publication series on dangerous substances (PGS 2). The Hague 2005. ACGRATTAN, K.B.; BAUM, H.R. AND REHM, R.G. Numerical simulation of smoke plumes from larg AUDISOVA B.; DANIHELKA P.; MICEK D. Evaluation of acute toxicity limits for purposes of risk ana 0) 911-916. GANN, R.G.; BRYNER, N.P. Combustion Products and Their Effects on Life Safety. In A.E. Cote, C.C. G acy: National Fire Protection Association. 2008.

THIS WORK WAS SUPPORTED BY THE MINISTRY OF EDUCATION, YOUTH AND SPORTS OF THE CZECH REPUBLIC VIA THE PROJECT LD11012 (IN THE FRAME OF THE COST CM0901 ACTION) AND THE MINISTRY OF ENVIRONMENT OF THE CZECH REPUBLIC (PROJECT NO. SPII 1A10 45/70).

For liquid pool fires, fuel properties (heat of combustion, mass burning rate and radiative fraction) and pool surface area are required in order to determine the source term of toxic species and buoyant force. For gaseous fuels, release rate (in kg/s) needs to be estimated, e.g. according to [1].

3.10: C:\Documents and Settings\VN\Plocha\aloft-ft\egu\e 🔳 🗖 🗙	🕒 Database: C:\Pro	gram Files\NIST\A	LOFT-FT 3.10\fu	els.mdb 🛛 🗙			
Graphs Map Contours Help	Add New Fuel	Add <u>N</u> ew Fuel Bas	ed on This Fuel	Delete Fuel Close			
Temperature	Fuel: 5		F H				
Vinyl Chloride (release rate 1 kg/s) jet fire	Fuel Name:	Vinyl Chloride					
I Distance (maximum) 3 km	Heat Release Rate per Unit Area (Total): Radiative Fraction:	0,169	169 MW/m^2 0,1				
Over Water (initial) 0 km Fuels Database	Burning Rate per Unit Area:	0,01	kg/m^2-s	_			
e Vinyl Chloride	Notes:	100% Cl in fuel conver	ted to HCI				
oride Edit Fuel III ◀ Fuel: 5 ► III Properties	Emission Factors –		User Defined En	nission Factors			
Fuel Properites Select Fuel Database	PM10:	150 g/kg	User 1 Label:	Hydrogen Chloride			
	PM2.5:	105 g/kg	User 1 Value:	584			
	C02:	771 g/kg	User 1 Units:	microgram/m3			
e 2 Fire 3 Fire 4 Fire 5 Fire 6	CO:	56 g/kg	User 2 Label:				
100 sq.m	S02:	0 g/kg	User 2 Value:	0			
ion - Optional	VOC:	0 g/kg	User 2 Units:				
e N 📩 🚺 0 deg 🚺 0 min 🚺 0 sec			User 3 Label:				
			User 3 Value:	0			
ide win 0 deg 0 min 0 sec			User 3 Units:				

Atmospheric dispersion of smoke plume was subsequently modelled by the ALOFT-FT [2] tool (large-eddy simulation) for the given accidental scenario (type of fire, wind speed, etc.):

Hydrogen Chloride Concentration (microgram/m3 - one hr avg) Horizontal Plane, 0 m Elevation

Downwind (km)

Hydrogen Chloride Concentration (microgram/m3 - one hr avg) Vertical Plane, 0 m Crosswind

calculation of physical effects (3rd Edition, 2nd Revised Print). CPR 14E ("Yellow

rge oil fires. Atmospheric Environment 30 (1996) 4125-4136. alysis in the Czech Republic. Reliability, Risk And Safety: Theory and Applications 1-3

Grant, J. R., Jr. Hall, R. E. Solomon (Eds.) Fire Protection Handbook, 20th Edition.



