



Detection of reduced sulfur and other S-bearing species evolved from Rocknest sample in the Sample Analysis at Mars (SAM) experiment

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The SAM instrument suite onboard the Mars Science Laboratory (MSL) Curiosity Rover detected sulfur-bearing compounds during pyrolysis of soil fines obtained from aeolian material at Rocknest in Gale Crater. SO₂ and H₂S were identified by the quadrupole mass spectrometer (QMS) both in direct evolved gas analysis mass spectrometry (EGA-MS) and after gas chromatograph separation (GC-MS) [1].

In EGA-MS, the 34 Da trace shows at least 3 peaks. The first peak is evolved at relatively low temperature (T), near 400°C, and the other peaks evolved as part of a “hump” at higher T, between ~500°C and ~800°C. The higher T releases at 34 Da occur at T close to, but not at exactly the same, as an evolution of SO₂ from the samples. We hypothesize that these 34 Da releases are due to H₂S. This assertion is supported by peaks in 35 and 36 Da traces at the same T. The lower T release of 34 Da species corresponds to a large O₂ release from the Rocknest samples, and can be attributed for the most part to an isotopologue of O₂. However, the GCMS analysis of the temperature cut involving this first evolved peak displays evidence of H₂S based on a comparison of the mass spectrum to a NIST library. Therefore, we propose that H₂S must be contributing to the 400°C peak. The quantification of H₂S from GCMS shows an amount of this species of less than 1 nmol. It is unclear what the source of this lower T H₂S is and how sulfur remains in its reduced form instead of undergoing oxidation to SO₂ at the temperature where O₂ is evolved; laboratory work with relevant analogs to inform these questions is ongoing. An initial hypothesis for the low temperature H₂S source is the product of a reaction between an S-bearing phase and a hydrogen-bearing phase, such as the abundant water evolved at less than 500°C from the sample. Potential sources of this water are adsorbed water or mineral structural water.

There is also EGA-MS evidence of reaction of reduced S with CO₂ in the pyrolysis oven to form OCS (main mass 60 Da) and possibly CS₂ (main mass 76 Da). Sulfur in all detected compounds is highly likely to have a Martian origin, as from the analysis of SAM background, the S-bearing species are quantitatively very limited and no known chemical process enables the formation of H₂S from the background. Pyrolysis experiments on SNC meteorites and most recently the Tissint meteorite show a large SO₂ peak evolved at temperatures above 600°C, which was not observed at lower temperature, and may be from sulfate thermal degradation. GC shows that a large quantity of CO₂ is evolved along with OCS and CS₂. However, the absence of H₂S confirms a high oxidation of the molecules in the sample. This Tissint meteorite is believed to have resemblances with Rocknest soil, and a cross-analysis of these results, in addition to the lab work, are essential steps for a complete understanding of these Martian sulfur-bearing compounds.