



Simultaneous measurement of CO₂ concentration and isotopic ratios in gas samples using IRMS

Eun-Ji Yu (1), Dongho Lee (2), Yeon-Sik Bong (2), Kwang-Sik Lee (1,2)

(1) Graduate School of Analytical Science and Technology, Chungnam National University, Daejeon, Republic of Korea, (2) Korea Basic Science Institute, Division of Earth & Environmental Sciences, Ochang, Republic of Korea (bonggeo@gmail.com)

Isotopic methods are indispensable tools for studies on atmosphere-biosphere exchanges of CO₂ and environmental monitoring such as CO₂ leakage detection from subsurface carbon storages. CO₂ concentration is an important variable in interpreting isotopic composition of CO₂ especially in atmospheric applications (e.g., 'Keeling plot'). Optical methods such as CRDS (Cavity Ring Down Spectroscopy) are gaining attention recently because of its capability to simultaneously measure CO₂ concentration and isotopic ratios with a short measurement interval (up to 1 sec.). On the other hand, IRMS (Isotope Ratio Mass Spectrometer) has been used only for isotopic measurements.

In this study, we propose a method to measure CO₂ concentration from gas samples along with isotopic ratios using conventional IRMS system. The system consists of Delta V Plus IRMS interfaced with GasBench II (Thermo Scientific, Germany). 12-mL vials with open top screw cap and rubber septum were used for both gas sampling and analysis. For isotopic analysis, gases in the vials were transferred into GasBench II by He carrier flow and CO₂ was trapped by a single cryotrap (-180 °C) after passing a water trap (Mg(ClO₄)₂). Upon release of the cryotrap, liberated CO₂ was separated from N₂O using gas chromatography column inside the GasBench II and introduced online into the IRMS. Isotopic ratios were measured for the masses of 44, 45 and 46, and the peak intensity (mV of mass 44 and peak area) was recorded for the concentration calculation.

For the determination of CO₂ concentration, a calibration curve relating the peak intensity with molar concentration of CO₂ was constructed. By dissolving NaHCO₃ in de-ionized water, solutions containing 0.05, 0.1, 0.25 and 0.5 μmol of inorganic carbon were prepared in 12 mL vials. Phosphoric acid was injected through rubber septum of the vials to acidify the solution and released CO₂ was analyzed for the isotopic ratios and the corresponding peak intensity was recorded using the same procedure with gas samples. The peak area (Vs) linearly correlated with the molar concentration of inorganic carbon ($r^2 > 0.99$) with sufficient stability, thereby enabling the determination of CO₂ concentration in gas samples. The calibrated range of CO₂ concentration corresponds to 200 ~ 1000 ppm of atmospheric pCO₂. Accuracy of the method is continuously being improved by analyzing standard gas samples with known concentration and isotopic ratios. This method provides a simple, cost-effective technique that can be applied for various studies based on the isotopic composition and concentration of CO₂.