Monitoring of CO,-induced geochemical changes in a shallow aquifer by time domain spectral induced polarization

Introduction

- Geological carbon sequestration is a promising technique for reducing CO₂ release into the atmosphere
- Potential risk: CO₂ or brine leakage into potable aquifers

Monitoring of dissolved CO₂ in shallow aquifers

- No direct CO₂ signal, only geochemical changes induced by the CO₂ can be observed
- Decreasing resistivity due to more dissolved ions, but effect highly site dependent (e.g., Dafflon et al., 2013, ES&T 47, 314)

Questions addressed here

- Can CO₂-induced geochemical changes be imaged with surface direct current (DC) and time domain induced polarization (IP) measurements?
- Is there any IP effect and if so, what causes it and can it be imaged?
- How reliable and stable is IP monitoring?

2. CO, injection experiment

Field site in western Denmark

- Geology consisting of
- \cdot Aeolian sand (0-5 m depth)
- · Glacial sands (5-10 m depth)
- · Marine sands (below 10 m depth)
- Groundwater table at 2 m depth

CO₂ injection

- 4 injection points at 5 and 10 m depth
- Total of 1600 kg CO₂ injected in 72 days
- Injection started on May 14th 2012, 12 L/min, reduced to 6 L/min after 14 days
- Pilot study showed consistent decrease of resistivity with increasing CO₂ concentrations

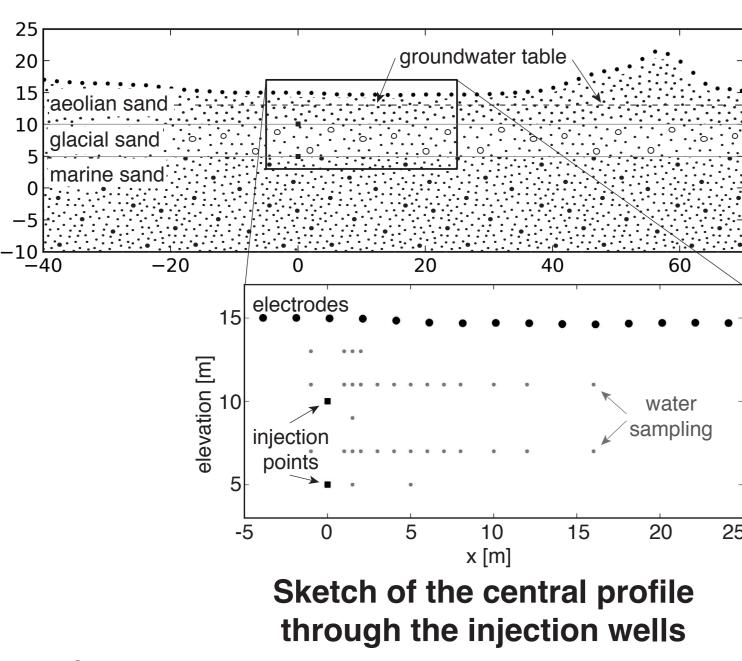
3. DC/IP and geochemical monitoring

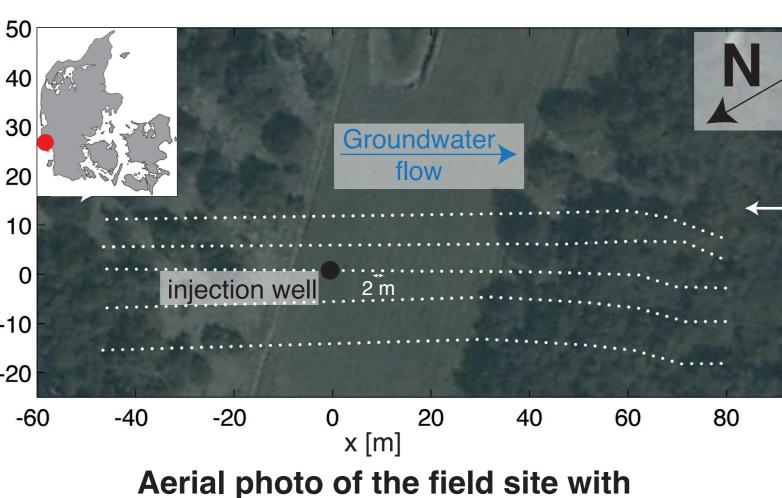
DC/IP monitoring

- 64 electrodes at 2 m spacing on profile through injection wells
- Fully automated acquisition
- High-quality measurements before CO₂ injection and continuous acquisition until day 113 (counted from injection start)

Geochemical monitoring

- Frequent monitoring at 33 locations on the profile
- Analysis for electrical conductivity (EC), pH and dissolved element concentrations (Al, Ca, Mn, K, ...)



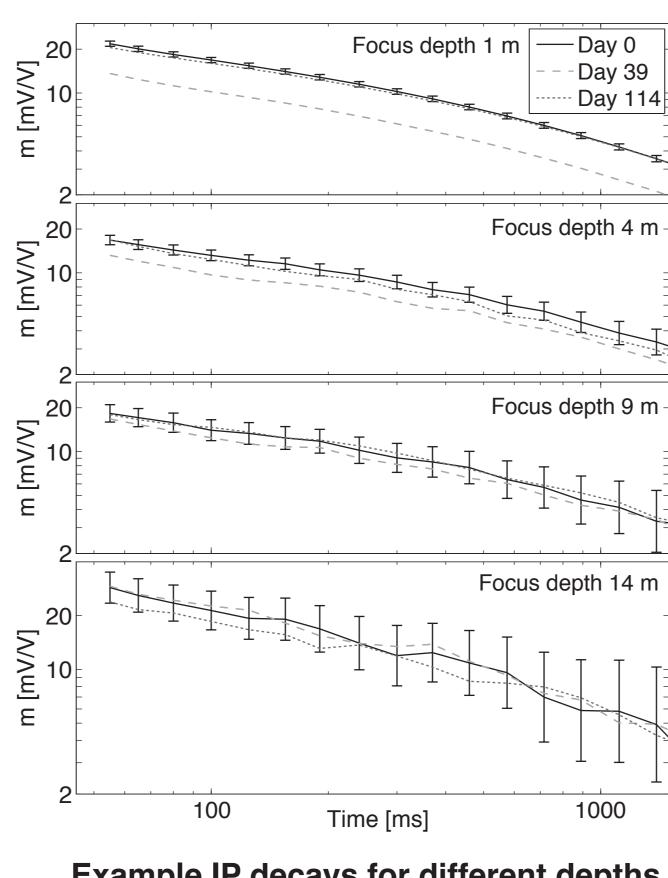


Aerial photo of the field site with electrodes marked as white dots

4. DC/IP data processing

Pre-processing

- 886 gradient-type four electrode configurations, $2 \text{ s } T_{on}$, $2 \text{ s } T_{off}$ time
- Remove IP decays that were
- negative
- above 500 mV/V
- curve increasing
- Error estimation
- absolute voltage error of 0.2 mV
- 4% relative error for DC and IP
- time-lapse differencing for DC (2% rel. error), but not for IP
- Significant change in IP decay curves in shallow configurations around injection wells



Example IP decays for different depths around the injection wells

5. Time domain spectral IP inversion

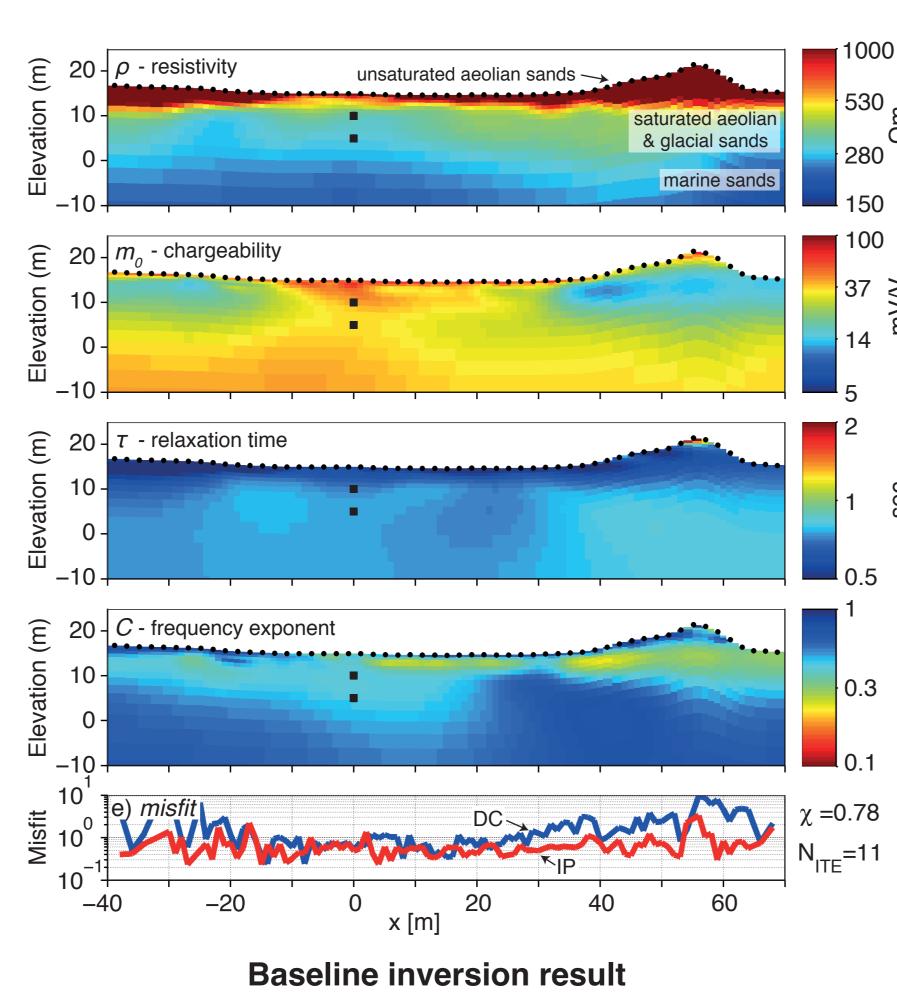
Full-decay inversion for spectral information

- Simultaneous inversion for resistivity and Cole-Cole parameters using AarhusInv (Fiandaca, G. et al., 2013, GJI **192**, 631)
- Cole-Cole parameters
- resistivity
- m_o chargeability
- relaxation time
- *C* frequency exponent

complex
$$\zeta$$
 resistivity

$$(\boldsymbol{\omega}) = \rho \left[1 - m_0 \left(1 - \frac{1}{1 + (i\omega\tau)^C} \right) \right]$$

- frequency ω
- Full modeling of current waveform and instrument filters
- Spectral information contained in Cole-Cole parameters

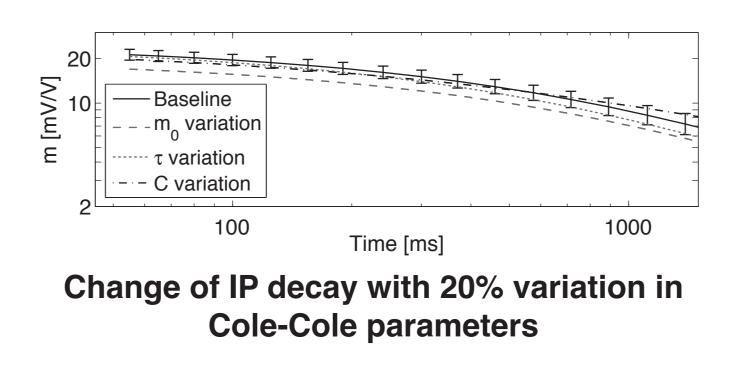


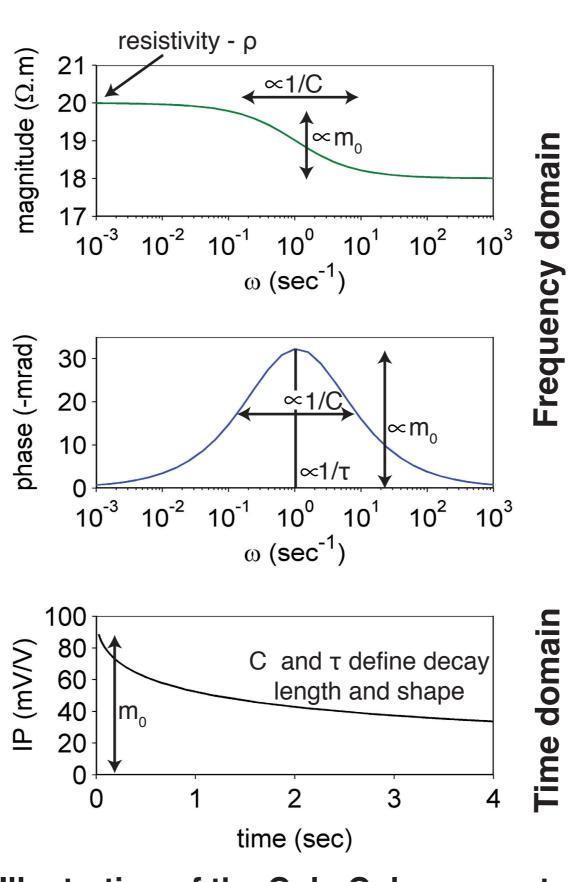
- **Baseline inversion**
- Inversion of pre-injection data
- Sequential inversion . DC only (4 iterations) 2. combined DC & IP (7 iterations)
- saturated sands
- Lower chargeability in dune, high monear surface
- τ and C show variation around water table



Sensitivity of IP decays

- Analyze change of IP decays to 20% variation in m_o , τ and C
- Normalized RMS with 20% variation and typical error estimates for 4 m focus depth configuration
 - *m*_o 2.16 - 0.73
 - *C* 0.75
- Only change in m_{o} can be reliably detected



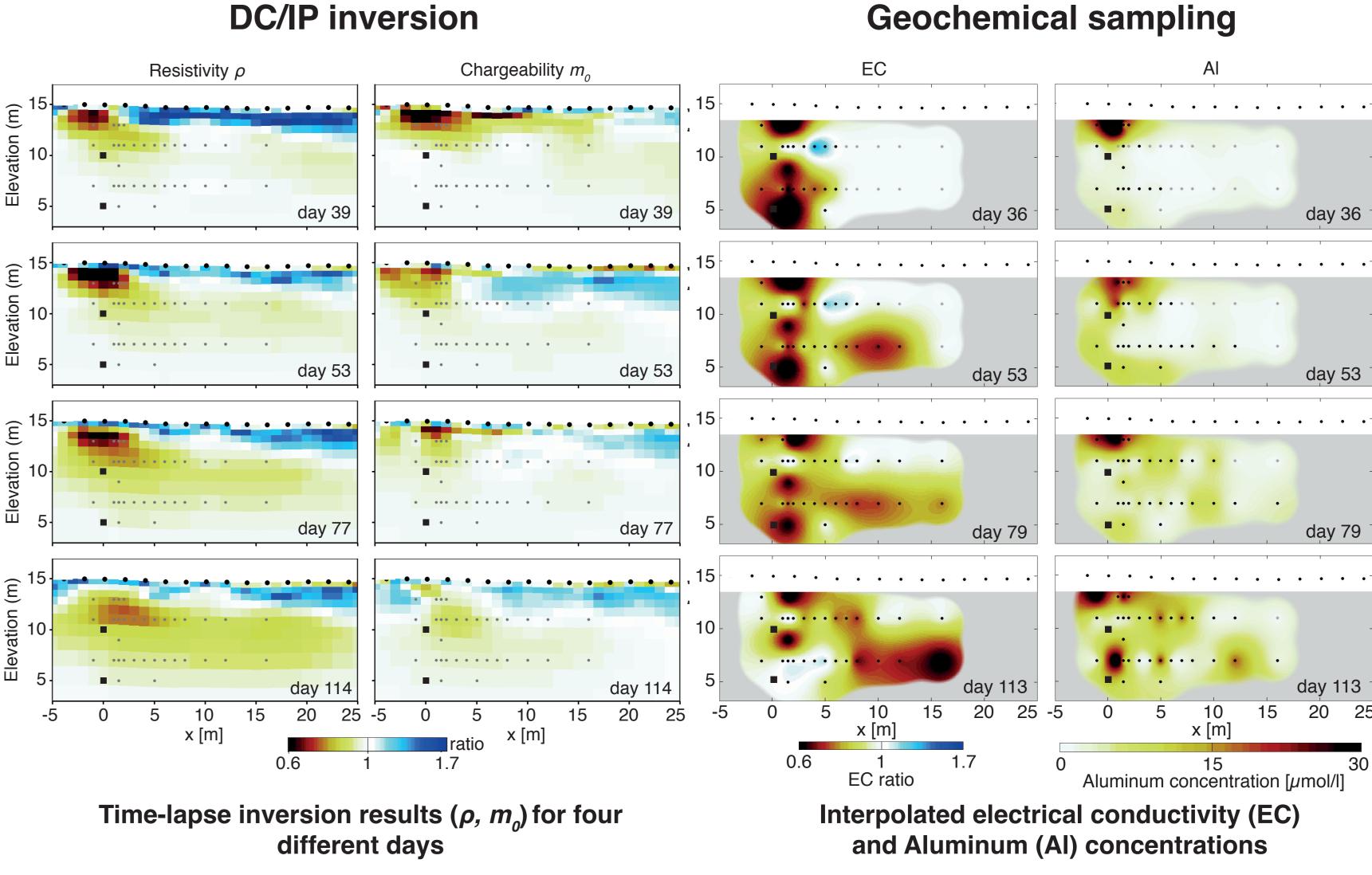




Resistivity section confirms sequence of unsaturated and

6. Time-lapse inversions

- Invert for differences in Cole-Cole parameters from baseline model
- Distinct CO₂ driven anomalies in ρ , m_0 ; τ and C variations are below detection limit



7. Conclusions

- chargeability
- Two different chemical mechanisms imaged:
- other cations by surface processes (primarily ion exchange) imaged by lower DC-resistivity
- 1. advective pulse from carbonic acid reacting with Al-hydroxide minerals forming bicarbonate and Al ions releasing 2. pH is lowered by the carbonic acid, changing surfaces directly (by dissolution) and indirectly by altering the surface charge, the combined effect being picked up in the IP signal.
- Good agreement with water chemistry samples from >30 sample locations (see Cahill et al., 2014, WRR 50, 1735)

Acknowledgements

- Danish Strategic Research Council.



- ρ images advective flow of dissolved CO₂; m_{ρ} images increasing persistent acidification, closely related to AI concentration
- IP anomaly moves much slower than resistivity

Geochemical sampling

• High-quality IP monitoring data can be acquired on the field scale over extended periods of time (120 days) • Surface DC and IP measurements are able to image geochemical changes induced by injected CO₂ • Time-lapse 2D full-decay IP inversions image the CO, plume as a decrease in resistivity and a decrease in

• We would like to thank the field crew for installing and continuously maintaining the ERT equipment. • J. Ramm and D.K. Løve performed initial processing of the data. This study was conducted as part of the CO2-GS project (http://co2gs.geus.net/) funded by the

Further information: Contact joseph.doetsch@geo.au.dk and see publication

Auken, E., Doetsch, J., Fiandaca, G., Christiansen, A.V., Gazoty, A., Cahill, A.G., Jakobsen, R., 2014, Imaging subsurface migration of dissolved CO2 in a shallow aquifer using 3-D time-lapse electrical resistivity tomography, J. Appli. Geophys. 101, 31-41.

