

Potential of on-line continuous leach ICP-MS for linking trace elements to mineralogy: PRELIMINARY RESULTS

Introduction

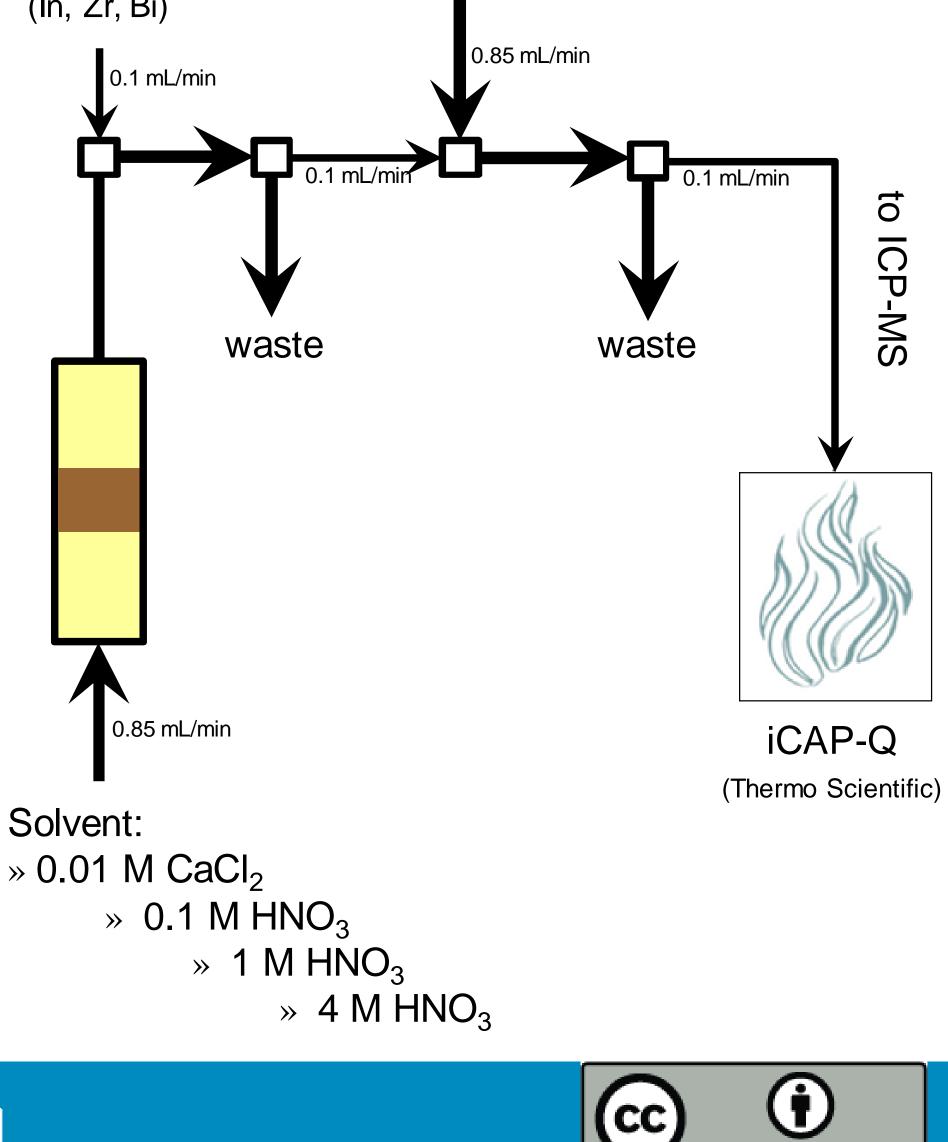
Continuous leach experiments have the advantage of real time (every 2 seconds) full elemental analysis. Mineral breakdown reactions can be monitored via the major elements, with simultaneous associated trace elements analysis, thus eliminating the uncertainties of host mineraltrace element combinations in traditional off-line sequential extractions.

The soil samples used were collected at 5 sites in the Koiliaris River watershed, Crete, Greece). The selection of the sites was based on variability in bedrock (limestone and alluvial sediments) and current land use (grape farming, olive trees). More information, see poster EGU2014-14894.

Method

A set of 5 soil samples was subjected to an on-line continuous leach inductively coupled plasma mass spectrometry experiment, with progressively reactive solvents $(0.01M CaCl_2, 0.1 M HNO_3, 1M)$ HNO_3 , 4M HNO_3). Each sample was diluted 1:5 with acid washed quartz to prevent clogging and packed in a quartz tube (\emptyset = 1 cm, length 2 cm) sandwiched by layers of quartz.

internal standard (In, Zr, Bi)

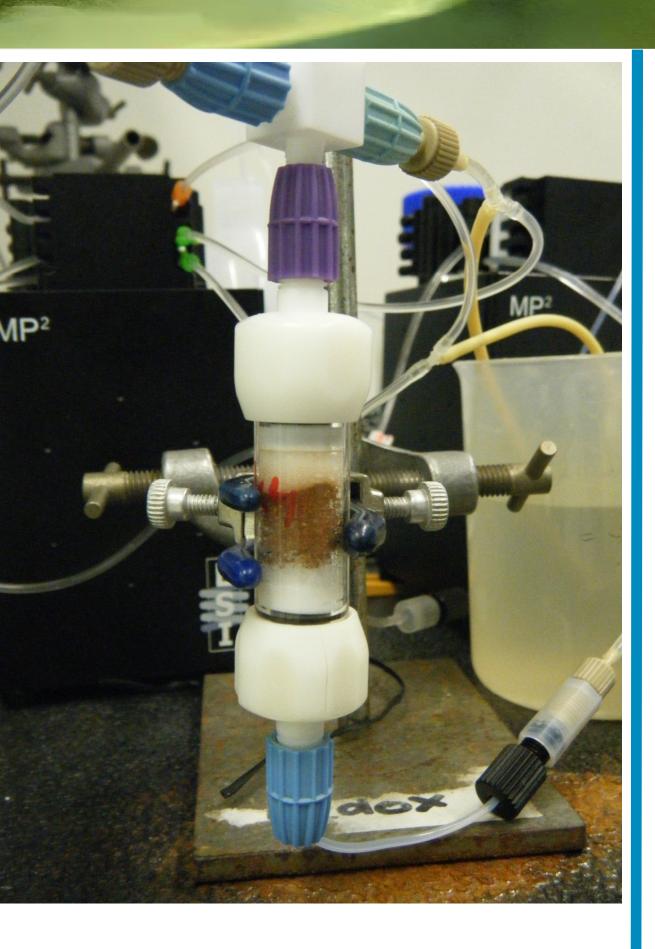


 $0.1 \text{ M} \text{HNO}_3$



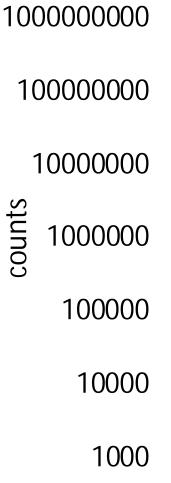


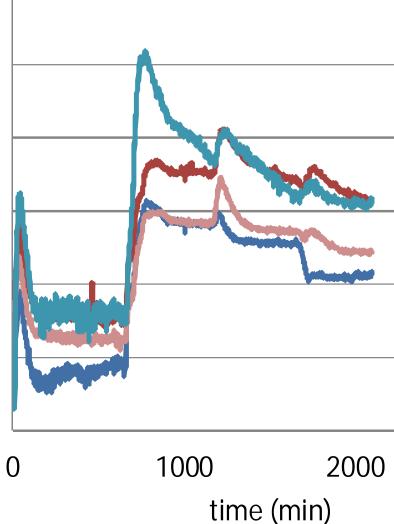
Giannakis and Pauline van Gaans





Despite the 10-fold dilution with 0.1M HNO₃, there is a clear matrix effect. Counts were corrected for the ¹¹⁵Insignal. Due to the generally limited leaching during the 0.01M CaCl₂ step and the instability of the signals in this solvent, this step was considered a wash step.







- The data can be used for mineral differentiation.
- Large amounts of data are generated, target specific questions!

Future work

- Investigate the reliability of the obtained results
- Compare leached amounts to total content;
- Vary in eluents;
- Adjust experiment so a gradual increase in the HNO₃concentration is applied;
- Link elements to mineral phases.
- Compare subsoil to topsoil samples

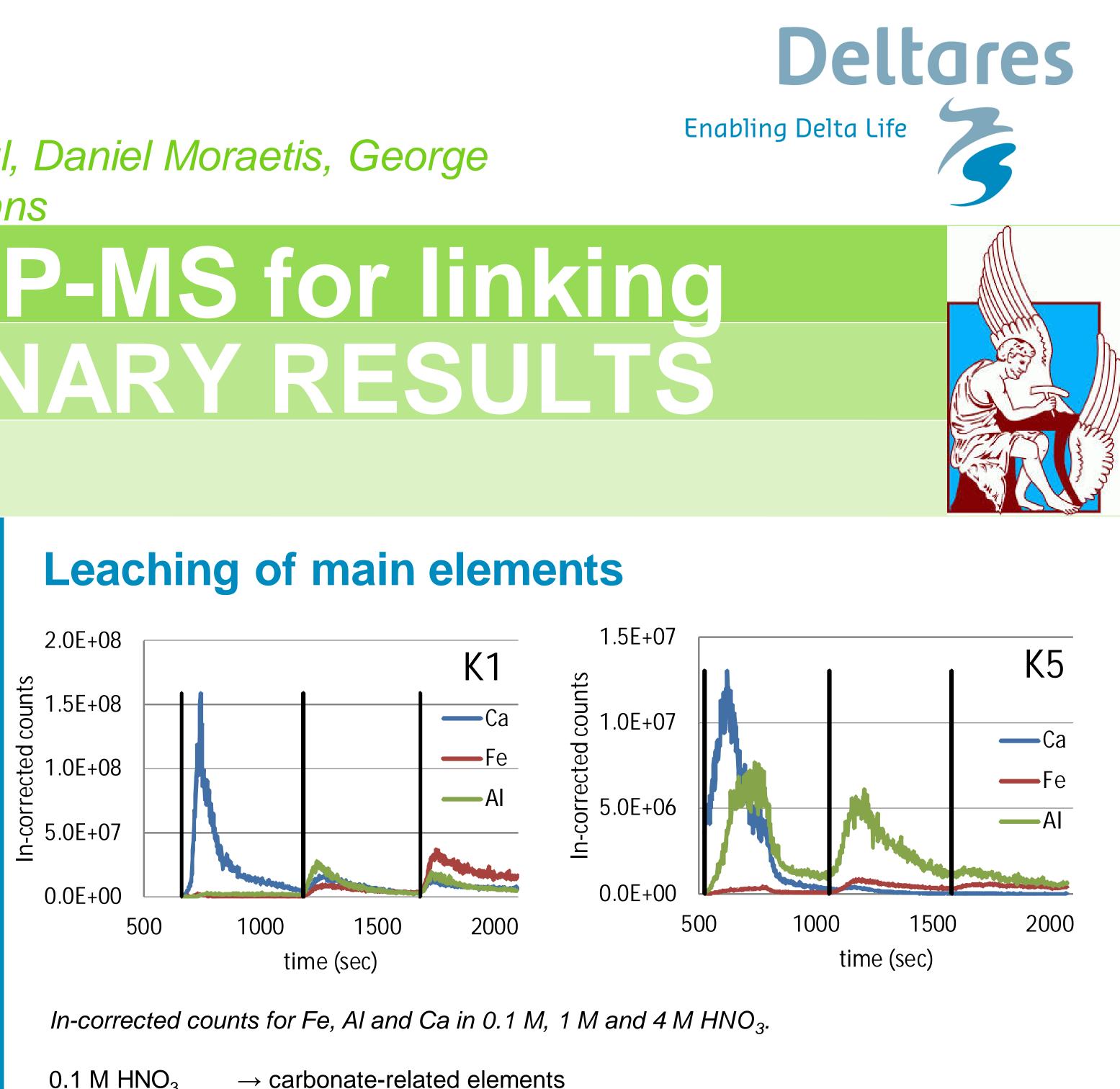
Gerlinde Roskam, Marc Verheul, Daniel Moraetis, George

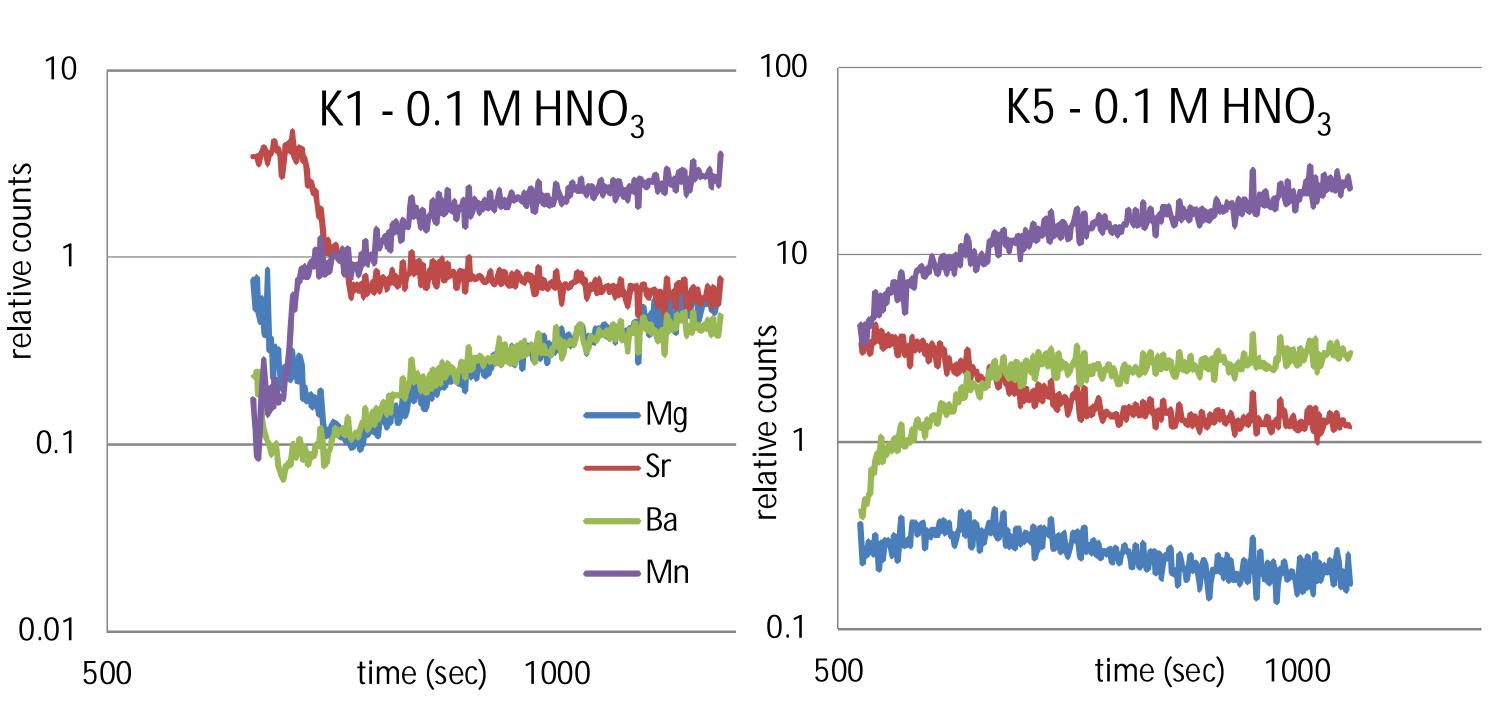
 1 M HNO_3

 4 M HNO_3

— In
— Zr
 — Bi
 Ca

3000





Counts of several elements in 0.1 M HNO₃ normalised on the counts for calcium.

In K1, the carbonates that initially dissolve have a higher Sr and Mg content than carbonates that are harder to dissolve;

 \rightarrow clay-related elements

 \rightarrow iron-containing minerals

Though a high percentage of the leachable Ba and Mn is released in the first extraction step, these elements are not or just partly related to Ca.



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