Geophysical Research Abstracts Vol. 17, EGU2015-11349, 2015 EGU General Assembly 2015 © Author(s) 2015. CC Attribution 3.0 License.



Fluorinated greenhouse gases in the troposphere and stratosphere

Johannes C. Laube (1), Eileen Gallacher (1), David E. Oram (1,2), Harald Bönisch (3), Carl A. M. Brenninkmeijer (4), Paul J. Fraser (5), Thomas Röckmann (6), and William T. Sturges (1)

(1) Centre for Ocean and Atmospheric Sciences, School of Environmental Sciences, University of East Anglia, Norwich, United Kingdom (j.laube@uea.ac.uk), (2) National Center for Atmospheric Science (NCAS), School of Environmental Sciences, University of East Anglia, UK, (3) Institute for Atmospheric and Environmental Sciences, Goethe University of Frankfurt, Germany, (4) Max Planck Institute for Chemistry, Air Chemistry Division, Mainz, Germany, (5) CSIRO Marine and Atmospheric Research, Australia, (6) Institute for Marine and Atmospheric Research, Utrecht University, The Netherlands

Fluorinated organic trace gases in the atmosphere are almost exclusively thought to be of anthropogenic origin. In the case of fully fluorinated alkane and cycloalkane-derivatives their IR absorption features and very long atmospheric lifetimes (on the order of thousands of years) make them very strong greenhouse gases. We here present measurements of 10 of these perfluorocarbons in the UT/LS and stratosphere as derived from deployments of regular passenger aircraft (CARIBIC project, http://www.caribic-atmospheric.com/) and the high-altitude research aircraft M55 Geophysica. In combination with long-term tropospheric records obtained from the Cape Grim observatory, Tasmania, we estimate their impact on radiative forcing expressed as CO₂-equivalents. As these gases have no significant sinks in the stratosphere they could also be suitable to derive an important transport diagnostic: the so-called mean age-of-air i.e. the average stratospheric transit time of an air parcel. We evaluate this possibility for all above-mentioned species and compare their characteristics with other inert species such as SF6, SF5CF3, and long-lived chlorofluorocarbons.