



Assessment of CO₂ discharge in a spring using time-variant stable carbon isotope data as a natural analogue study of CO₂ leakage

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CO₂-rich springs have been studied as a natural analogue of CO₂ leakage through shallow subsurface environment, as they provide information on the behaviors of CO₂ during the leakage from geologic CO₂ storage sites. For this study, we monitored the $\delta^{13}\text{C}$ values as well as temperature, pH, EC, DO, and alkalinity for a CO₂-rich spring for 48 hours. The water samples (N=47) were collected every hour in stopper bottles without headspace to avoid the interaction with air and the CO₂ degassing. The $\delta^{13}\text{C}$ values of total dissolved inorganic carbon (TDIC) in the water samples were analyzed using a cavity ring-down spectroscopy (CRDS) system (Picarro). The values of $\delta^{13}\text{C}_{TDIC}$, temperature, pH, EC, DO, and alkalinity were in the range of $-9.43 \sim -8.91 \text{ ‰}$, $12.3 \sim 13.2^\circ\text{C}$, $4.86 \sim 5.02$, $186 \sim 189 \mu\text{S}/\text{cm}$, $1.8 \sim 3.4 \text{ mg}/\text{L}$, and $0.74 \sim 0.95 \text{ meq}/\text{L}$, respectively. The concentrations of TDIC calculated using pH and alkalinity values were between 22.5 and 34.8 mmol/L. The $\delta^{13}\text{C}_{TDIC}$ data imply that dissolved carbon in the spring was derived from a deep-seated source (i.e., magmatic) that was slightly intermixed with soil CO₂. Careful examination of the time-series variation of measured parameters shows the following characteristics: 1) the $\delta^{13}\text{C}_{TDIC}$ values are negatively correlated with pH ($r = -0.59$) and positively correlated with TDIC ($r = 0.58$), and 2) delay times of the change of pH and alkalinity following the change of $\delta^{13}\text{C}_{TDIC}$ values are 0 and -3 hours, respectively; the pH change occurs simultaneously with the change of $\delta^{13}\text{C}_{TDIC}$, while the alkalinity change happens before 3 hours. Our results indicate that the studied CO₂-rich spring is influenced by the intermittent supply of deep-seated CO₂. [Acknowledgment] This work was financially supported by the fundamental research project of KIGAM and partially by the "Geo-Advanced Innovative Action (GAIA) Project (2014000530003)" from Korea Ministry of Environment (MOE).