



The Chemistry of Isoprene Hydroxy Hydroperoxides (ISOPOOH)

Jean Rivera (1), John Crouse (2), Tran Nguyen (2), Werner Jud (3), Jason St. Clair (4,5), Tomas Mikoviny (6), Jessica Gilman (7,8), Brian Lerner (7,8), Alex Teng (2), Kelvin Bates (9), John Seinfeld (9,10), Joost DeGouw (7,8), Armin Wisthaler (3,6), Armin Hansel (3), Paul Wennberg (2,10), Frank Keutsch (11,12)

(1) Department of Chemistry, University of Wisconsin-Madison, Madison, Wisconsin, USA, (2) Division of Geological and Planetary Sciences, California Institute of Technology, Pasadena, California, USA, (3) Institute of Ion Physics and Applied Physics, University of Innsbruck, Innsbruck, Austria, (4) Joint Center for Earth Systems Technology, University of Maryland, Baltimore, Maryland, USA, (5) NASA Goddard Space Flight Center, Greenbelt, Maryland, USA, (6) Department of Chemistry, University of Oslo, Oslo, Norway, (7) NOAA Earth System Research Laboratory, R/CSD7, Boulder, Colorado, USA, (8) Cooperative Institute for Research in Environmental Sciences, University of Colorado, Boulder, Colorado, USA, (9) Division of Chemistry and Chemical Engineering, California Institute of Technology, Pasadena, California, USA, (10) Division of Engineering and Applied Science, California Institute of Technology, Pasadena, California, USA, (11) School of Engineering and Applied Sciences, Harvard University, Cambridge, Massachusetts, USA, (12) Department of Chemistry & Chemical Biology, Harvard University, Cambridge, Massachusetts, USA

Organic hydroperoxides are important oxidation products of volatile organic compounds, especially under low- NO_x conditions. Despite their large global importance, the low- NO_x pathways are not well understood. High- NO_x oxidation pathways that typically produce carbonyls have been studied extensively. The formation of organic hydroperoxides makes the study of low- NO_x pathways challenging, as this class of compounds is not commercially available and the synthetic methods used to prepare them are still underdeveloped. This poses challenges for both quantification of these low- NO_x products as well as characterization of instruments with them. Isoprene hydroxyhydroperoxides (ISOPOOH) are the main first-generation products of the low- NO_x isoprene oxidation pathway; it is estimated that globally over 50% of isoprene peroxy radicals form ISOPOOH. We present a study of the kinetics of the formation of several ISOPOOH isomers as well as their atmospheric sinks. We also present instrument characterization studies that demonstrate that ISOPOOH is an interference in both GC and PTR-MS measurements. In these instruments ISOPOOH isomers are observed as the corresponding products (carbonyls) of the high- NO_x pathway. We discuss the interference mechanism as well as the implications of this interference on studies of OH reactivity, O:C ratios, OH recycling and secondary organic aerosol formation.