Geophysical Research Abstracts Vol. 17, EGU2015-6832, 2015 EGU General Assembly 2015 © Author(s) 2015. CC Attribution 3.0 License.



Determination of the Sources of Chlorinated Hydrocarbons Detected During MSL Missions

Arnaud Buch (1), Imène Belmahdi (1), Cyril Szopa (2), Caroline Freissinet (3), Danny P. Glavin (3), Kirsten Miller (4), Roger Summons (4), Pascaline Francois (5), Patrice Coll (5), Jennifer Eigenbrode (3), Samuel Teinturier (3), Rafael Navarro-Gonzalez (6), Jennifer Stern (3), Amy McAdam (3), Tristan Dequaire (5), Maeva Millan (2), Jean Yves Bonnet (2), David Coscia (2), Michel Cabane (2), and Paul Mahaffy (3)

(1) Ecole centrale Paris, LGPM, Chatenay-Malabry, France (arnaud.buch@ecp.fr), (2) LATMOS, Univ. Pierre et Marie Curie, Univ. Versailles Saint-Quentin & CNRS, 75005 Paris, France, (3) NASA Goddard Space Flight Center, Greenbelt, MD 20771, USA, (4) Department of Earth, Atmospheric and Planetary Sciences, Massachussetts Institute of Technology, Cambridge, Massachussetts, USA, (5) LISA, Univ. Paris-Est Créteil, Univ. Denis Diderot & CNRS 94010 Créteil, France, (6) Universidad Nacional Autónoma de México, México, D.F. 04510

Sample Analysis at Mars (SAM) is one of the instruments of the MSL mission. Three analytical instruments composed SAM: the Tunable Laser Spectrometer (TLS), the Gas Chromatography (GC) and the Mass Spectrometer (QMS). It includes a sample preparation and gas processing system implemented with a pyrolysis system, wet chemistry (MTBSTFA and TMAH) and the hydrocarbon trap (silica beads, Tenax[®] TA and Carbosieve G), employed to concentrate volatiles released from the sample prior to GC-MS analysis [1].

This study investigates several propositions for chlorinated hydrocarbon formation, detected by SAM. Here we report on the detection of chlorohydrocarbon compounds and their potential origin.

To date, SAM has detected a range of diverse chlorinated hydrocarbons from GCMS analysis of samples collected at the several locations explored by Curiosity (Rocknest, John Klein, Cumberland and Confidence Hill). Some of these chlorohydrocarbons are produced during pyrolysis by the reaction of Martian oxychlorine compounds in the samples with terrestrial carbon from a derivatization agent (MTBSTFA) present in SAM [2, 3]. Chlorobenzene cannot be formed by the reaction of MTBSTFA and perchlorates (2) and two other reaction pathways were therefore proposed : (i) reactions between the volatile thermal degradation products of perchlorates (e.g. O_2 , Cl2 and HCl) and Tenax[®] and (ii) the interaction of perchlorates (T > 200 °C) with OM from Martian soil such as benzenecarboxylates [4, 5].

References: 1. P. R. Mahaffy et al. (2012) Space Sci. Rev. 170, 401–478. 2. D. P. Glavin et al.(2013) JGR 118, 1955–1973. 3. L. a Leshin et al. (2013) Science 341, 1238937. 4. C. Freissinet et al. (2014) LPSC XXXXV Abstract 2796. 5. H. Steininger et al. (2012) Planet. Space Sci. 71, 9–17.