



Ti or Sn doping as a way to increase activity and sulfur tolerance of Mn/CeO₂ catalyst for low temperature NH₃ selective catalytic reduction of NO

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Abstract Text

Mn/CeO₂ catalysts modified by doping of Ti or Sn were investigated for low temperature selective catalytic reduction (SCR) of NO by NH₃ with the aim of studying the effects of Ti, Sn doping on the catalytic performance. Ceria-based solid solutions (Ce_{0.8}Ti_{0.2}O₂ and Ce_{0.8}Sn_{0.2}O₂) were synthesized via inverse co-precipitation, and used as supports to prepare MnO_x/Ce_{0.8}M_{0.2}O₂ (M = Ti⁴⁺, Sn⁴⁺) catalysts through wetness impregnation method. The results showed that doping of Ti or Sn to the CeO₂ support increase the NO removal efficiency. A NO conversion of more than 90 % was obtained over the Mn/CeTi catalyst at the temperature window of 175 ~ 300 °C under a gas hourly space velocity (GHSV) of 60,000 mL•g⁻¹•h⁻¹. Catalysts modified by Ti and Sn were also found to obtain higher SO₂ resistance than Mn/CeO₂ catalyst. More than 90% NO conversion and 95% N₂ selectivity could be provided by Mn/CeTi catalyst in the presence of 100 ppm SO₂ at 250 °C for 10 h. A series of characterization techniques, namely XRD, BET, H₂-TPR, XPS, NH₃-TPD and in situ DRIFTS were used to elucidate the structure and surface properties of the obtained supports and catalysts. The results indicate that doping of Ti or Sn brings about catalysts with favorable properties such as higher BET surface area, better oxygen storage capacity and stronger surface acidity. The relative amount of Mn⁴⁺, Ce³⁺, adsorbed oxygen species and oxygen vacancies on the surface of catalysts are in the order of Mn/CeTi [U+FF1E] Mn/CeSn [U+FF1E] Mn/CeO₂, which is thought to make positive a contribution to the low-temperature SCR activity. The promoted SCR activity is considered as well to be related to the dual redox cycles in Mn/CeTi (Mn⁴⁺ + Ce³⁺ $\xrightleftharpoons[]{} \text{Mn}^{3+} + \text{Ce}^{4+}$, Ce⁴⁺ + Ti³⁺ $\xrightleftharpoons[]{} \text{Ce}^{3+} + \text{Ti}^{4+}$) and Mn/CeSn (Mn⁴⁺ + Ce³⁺ $\xrightleftharpoons[]{} \text{Mn}^{3+} + \text{Ce}^{4+}$, Ce⁴⁺ + Sn²⁺ $\xrightleftharpoons[]{} \text{Ce}^{3+} + \text{Sn}^{4+}$) catalysts.