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Combined δ^{11} B, δ^{13} C, and δ^{18} O analyses of coccolithophore calcite constrains the response of coccolith vesicle carbonate chemistry to CO₂-induced ocean acidification

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Coccolithophorid algae play a central role in the biological carbon pump, oceanic carbon sequestration, and in marine food webs. It is therefore important to understand the potential impacts of CO₂-induced ocean acidification on these organisms. Differences in the regulation of carbonate chemistry, pH, and carbon sources of the intracellular compartments where coccolith formation occurs may underlie the diverse calcification and growth responses to acidified seawater observed in prior experiments. Here we measured stable isotopes of boron (δ^{11} B), carbon (δ^{13} C) and oxygen (δ^{18} O) within coccolith calcite, and δ^{13} C of algal tissue to constrain carbonate system parameters in two strains of *Pleurochrysis carterae* (*P. carterae*). The two strains were cultured under variable pCO₂, with water temperature, salinity, dissolved inorganic carbon (DIC), and alkalinity monitored. Notably, PIC, POC, and PIC/POC ratio did not vary across treatments, indicating that P. carterae is able to calcify and photosynthesize at relatively constant rates irrespective of pCO₂ treatment. The $\delta^{11}B$ data indicate that mean pH at the site of coccolith formation did not vary significantly in response to elevated CO₂. These results suggest that *P. carterae* regulates calcifying vesicle pH, even amidst changes in external seawater pH. Furthermore, δ^{13} C and δ^{18} O data suggest that P. carterae may utilize carbon from a single internal DIC pool for both calcification and photosynthesis, and that a greater proportion of dissolved CO₂ relative to HCO₃ enters the internal DIC pool under acidified conditions. These results suggest that P. carterae is able to calcify and photosynthesize at relatively constant rates across pCO2 treatments by maintaining pH homeostasis at their site of calcification and utilizing a greater proportion of aqueous CO_2 .