

C isotope fractionation during heterotrophic activity driven carbonate precipitation

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Stable carbon isotopic fractionation during carbonate precipitation induced by environmentally enriched heterotrophic halophilic microorganisms was experimentally investigated under various salinity (4.5, 8, 15) conditions at 30 °C. Halophilic heterotrophic microorganisms were enriched from a hypersaline Lake Acıgöl located in SW Turkey (Balci et al., 2015) and later used for the precipitation experiments (solid and liquid medium). The carbonate precipitates had relatively high $\delta^{13}\text{C}$ values (-4.3 to -16.9 ‰ compared to the $\delta^{13}\text{C}$ values of the organic compounds that ranged from -27.5 to -25.4 ‰. At salinity of 4.5 ‰ $\delta^{13}\text{C}$ values of carbonate ranged from -4.9 ‰ to -10.9 ‰ with a ^{13}C -enrichment factor of +20 to +16 ‰ higher than the $\delta^{13}\text{C}$ values of the associated DOC (-27.5) . At salinity 8 ‰ $\delta^{13}\text{C}$ values of carbonate ranged from -16.3 ‰ to -11.7 ‰ with a ^{13}C -enrichment factor of +11.3 to +15.9 ‰ higher than the $\delta^{13}\text{C}$ values of the associated DOC. The respected values for 15 ‰ salinity ranged from -12.3 ‰ to -9.7 ‰ with a ^{13}C -enrichment factor of +15.2 to +16.8 ‰ higher than the $\delta^{13}\text{C}$ values of the associated DOC. The carbonate precipitates produced in the solid medium are more enriched in ^{13}C relative to liquid culture experiments. These results suggest that the carbon in the solid was derived from both the bacterial oxidation of organic compounds in the medium and from the atmospheric CO_2 . A solid medium used in the experiments may have suppressed convective and advective mass transport favouring diffusion-controlled system. This determination suggests that the rate and equilibration of CO_2 exchange with the atmosphere is the major control on C isotope composition of carbonate minerals precipitated in the experiments.

Key words: Lake Acıgöl, halophilic bacteria, carbonate biomineralization, C isotopes

References

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