



Trifluoroacetic Acid Level in the Atmosphere of Beijing and Its Relationship with PM_{2.5}

Junyu Guo (1) and Jianbo Zhang (2)

(1) Peking University, College of Environmental Sciences and Engineering, Peking University, China (guojy16@163.com),

(2) Peking University, College of Environmental Sciences and Engineering, Peking University, China (jbzhang@pku.edu.cn)

Atmospheric concentrations of Trifluoroacetic Acid (TFA), one of the main degradation products of HCFC-123, HCFC-124 and HFC-134a, were detected in Beijing, China between 2013 and 2014. By analyzing the 137 atmospheric samples, the results showed that the annual mean atmospheric concentration of TFA was 1459 ± 223 $\text{pg} \cdot \text{m}^{-3}$. TFA was mainly distributed in gaseous phase, for the concentration was 1396 ± 225 $\text{pg} \cdot \text{m}^{-3}$, while that in particle phase was 62 ± 8 $\text{pg} \cdot \text{m}^{-3}$. Considering the frequent occurrence of hazy weather in Beijing, the relationship between TFA and PM_{2.5} in atmosphere was analyzed. The correlation analysis shows that the proportion of particle phase in atmosphere concentration of TFA and mass concentration of PM_{2.5} are positively correlated with each other ($P < 0.001$), indicating the particles have an absorption effect on TFA. At the same time, when mass concentration of PM_{2.5} in atmosphere is high, atmospheric concentration of TFA is relatively low. According to the correlation analysis, mass concentration of PM_{2.5} and atmospheric concentration of TFA are negatively correlated with each other ($P = 0.005$). The main reason is very likely that particle's extinction for light can be enhanced as particle level rises, which causes TFA precursors photolysis to weaken. The results indicate that PM_{2.5} has a significant impact on TFA.