

## **Combining CO<sub>2</sub> sequestration and CH<sub>4</sub> production by means of guest exchange in a gas hydrate reservoir: two pilot scale experiments**

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Methane (CH<sub>4</sub>) hydrates are considered as a player in the field of energy supply and – if applied as such – as a possible sink for the greenhouse gas carbon dioxide (CO<sub>2</sub>). Next to the more conventional production methods depressurization and thermal stimulation, an extraction of CH<sub>4</sub> by means of CO<sub>2</sub> injection is investigated. The method is based on the chemical potential gradient between the CH<sub>4</sub> hydrate phase and the injected CO<sub>2</sub> phase. Results from small-scale laboratory experiments on the replacement method indicate recovery ratios of up to 66% CH<sub>4</sub> but also encounter major discrepancies in conversion rates. So far it has not been demonstrated with certainty that the process rates are sufficient for an energy and cost effective production of CH<sub>4</sub> with a concurrent sequestration of CO<sub>2</sub>.

In a co-operation of GFZ and GEOMAR we used LARS (Large Scale Reservoir Simulator) to investigate the CO<sub>2</sub>-CH<sub>4</sub>-replacement method combined with thermal stimulation. LARS accommodates a sample volume of 210 l and allows for the simulation of in situ conditions typically found in gas hydrate reservoirs. Based on the sample size, diverse transport mechanisms could be simulated, which are assumed to significantly alter process yields. Temperature and pressure data complemented by a high resolution electrical resistivity tomography (ERT), gas chromatography, and flow measurements serve to interpret the experiments.

In two experiments 50 kg heated CO<sub>2</sub> was injected into sediments with CH<sub>4</sub> hydrate saturations of 50%. While in the first experiment the CO<sub>2</sub> was injected discontinuously in a so called “huff’n puff” manner, the second experiment saw a continuous injection. Conditions within LARS were set to 13 MPa and 8°C, which allow for stability of pure CO<sub>2</sub> and CH<sub>4</sub> hydrates as well as mixed hydrates. The CO<sub>2</sub> was heated and entered the sediment sample with temperatures of approximately 30° C. In this presentation we will discuss the results from the large-scale experiments and compare them with data from small-scale experiments.