



How much CO₂ is trapped in carbonate minerals of a natural CO₂ occurrence?

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Carbon Capture and Storage (CCS) is a transitional technology to decrease CO₂ emissions from human fossil fuel usage and, therefore, to mitigate climate change. The most important criteria of a CO₂ geological storage reservoir is that it must hold the injected CO₂ for geological time scales without its significant seepage. The injected CO₂ undergoes physical and chemical reactions in the reservoir rocks such as structural-stratigraphic, residual, dissolution or mineral trapping mechanisms. Among these, the safest is the mineral trapping, when carbonate minerals such as calcite, ankerite, siderite, dolomite and dawsonite build the CO₂ into their crystal structures. The study of natural CO₂ occurrences may help to understand the processes in CO₂ reservoirs on geological time scales. This is the reason why the selected, the Mihályi-Répcelak natural CO₂ occurrence as our research area, which is able to provide particular and highly significant information for the future of CO₂ storage. The area is one of the best known CO₂ fields in Central Europe. The main aim of this study is to estimate the amount of CO₂ trapped in the mineral phase at Mihályi-Répcelak CO₂ reservoirs. For gaining the suitable data, we apply petrographic, major and trace element (microprobe and LA-ICP-MS) and stable isotope analysis (mass spectrometry) and thermodynamic and kinetic geochemical models coded in PHREEQC.

Rock and pore water compositions of the same formation, representing the pre-CO₂ flooding stages of the Mihályi-Répcelak natural CO₂ reservoirs are used in the models. Kinetic rate parameters are derived from the USGS report of Palandri and Kharaka (2004). The results of petrographic analysis show that a significant amount of dawsonite (NaAlCO₃(OH)₂, max. 16 m/m%) precipitated in the rock due to its reactions with CO₂ which flooded the reservoir. This carbonate mineral alone traps about 10-30 kg/m³ of the reservoir rock from the CO₂ at Mihályi-Répcelak area, which is an unexpectedly high proportion of total amount of CO₂. Further results enlightened that other carbonates, ankerite, calcite and siderite have precipitated in two generations, the first before and the second after the CO₂ flooding. Further laboratory analysis and geochemical models allow us to estimate the ratio of these two generations and also to understand how far the reservoir rock is in the CO₂ mineral trapping process.