



## **H<sub>2</sub>S and SO<sub>2</sub> degassing from El Chichón (Chiapas, Mexico) and Víti (Iceland) volcanic lakes**

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Measurements of the composition of volcanic gas, especially S-species, have mainly been concentrated in fumaroles, plumes, bubbling/boiling waters and hyper-acidic lakes hosted by quiescent and active volcanoes. Here, we study for the first time the composition (CO<sub>2</sub>, H<sub>2</sub>S, H<sub>2</sub> and SO<sub>2</sub>) of gases released by two steam-heated volcanic lakes, El Chichón volcano (Chiapas, Mexico) and Víti (Askja volcano, Iceland). Both lakes have similar pH (2-3) and SO<sub>4</sub><sup>2-</sup> contents (~700 mg/l) but differ for their temperature (30.1±1 °C at El Chichón and 21.8±1 °C at Víti) and maximum depth (~11 m and ~60 m at El Chichón and Víti, respectively). The aim of this study is to test if sulfur species (H<sub>2</sub>S and SO<sub>2</sub>) are actively released at these chemical and physical lake-water conditions. Our results evidence the presence of relatively high (1-100 ppmv) H<sub>2</sub>S concentrations in the El Chichón and Víti lake gas plumes, with CO<sub>2</sub>/H<sub>2</sub>S and H<sub>2</sub>/H<sub>2</sub>S ratios (31.5-5685 and 0.77-35.1, respectively) higher than in the feeding volcanic gas, i.e. the offshore fumarolic gas composition (12.7-28.6 and 0.08-0.49, respectively). H<sub>2</sub>S degassing at the lake surface implies that only a fraction of the fumarolic gas entering the lake bottom is ultimately dissolved into the lakes. At El Chichón, by scaling our CO<sub>2</sub>/H<sub>2</sub>S ratios by the lake CO<sub>2</sub> output in 2016, we evaluate an H<sub>2</sub>S flux from the crater lake into the atmosphere at 0.02-0.06 t/d. Surprisingly, SO<sub>2</sub>, a highly soluble gas in water, is also detected (at trace levels, 0.003-0.3 ppmv) in the plume of both lakes. To explain SO<sub>2</sub> degassing from pH 2-3 volcanic lakes, we propose a formation mechanism via H<sub>2</sub>S oxidation reactions into lake waters, having dissolved sulphite as transient species. This explanation is favoured by higher SO<sub>2</sub> concentrations measured where higher dissolved sulphite were detected (i.e. in bubbling areas; Casas et al., 2016). The original H<sub>2</sub>S is therefore partially oxidized in the lake water (mainly as SO<sub>4</sub><sup>2-</sup>), and partially released as a gas phase, as H<sub>2</sub>S or SO<sub>2</sub>.

### Reference

Casas, A. S., Armienta, M. A., & Ramos, S. (2016). Sulfur speciation with high performance liquid chromatography as a tool for El Chichón volcano, crater lake monitoring. *Journal of South American Earth Sciences*, 72, 241-249. <https://doi.org/10.1016/j.jsames.2016.09.001>