



First real-time isotopic characterization of N₂O from chemodenitrification

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Chemodenitrification is the main abiotic nitrous oxide (N₂O) source in soils, its N₂O isotopic signatures could be applied for source attribution, but are currently not known in sufficient detail. To overcome this limitation, the intramolecular isotopic composition of N₂O: ¹⁴N¹⁴N¹⁶O, ¹⁵N¹⁴N¹⁶O (¹⁵N^β-N₂O), ¹⁴N¹⁵N¹⁶O (¹⁵N^α-N₂O), and ¹⁴N¹⁴N¹⁸O, produced by the reaction of sodium nitrite with lignin, five lignin derivatives, and three types of soils, was analyzed online with a quantum cascade laser absorption spectrometer (QCLAS). Thereby, we present the first continuous dataset of $\delta^{15}\text{N}^{bulk}$ ($\delta^{15}\text{N}^{bulk} \equiv (\delta^{15}\text{N}^{\alpha} + \delta^{15}\text{N}^{\beta})/2$), $\delta^{18}\text{O}$, and SP (site preference, $\text{SP} \equiv \delta^{15}\text{N}^{\alpha} - \delta^{15}\text{N}^{\beta}$) of N₂O from chemodenitrification in both aqueous solutions and soils. Contrary to the previous assumption that SP values are constant for distinct abiotic N₂O source processes, our results reveal a considerable shift along time in SP from 5.6 to 29.0 ‰ in the reactions of sodium nitrite with organic substances. The large SP variability of N₂O might be explained by the multiple reaction pathways involved. These findings provide important new information for N₂O source partitioning using intramolecular N₂O isotopic signatures, e.g. end-member maps of SP vs. $\delta^{15}\text{N}^{bulk}$ and $\delta^{18}\text{O}$.