



## Triple oxygen isotope analysis of nitrate using cavity ringdown laser spectroscopy

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Triple oxygen isotopes in nitrates are a valuable tool to ascertain the pathways of nitrate formation in the atmosphere and the fate of nitrate in ecosystems. Current analytical methods for determining  $^{17}\text{O}/^{16}\text{O}$ ,  $^{18}\text{O}/^{16}\text{O}$  and derived  $^{17}\text{O}$  anomalies ( $\Delta^{17}\text{O}$ ) are tedious, time-consuming and sometimes involve hazardous reagents. Here we present a new method for triple oxygen isotope analysis of nitrate, based on nitrate-water isotope equilibration (IE) and subsequent isotopic analysis of water using cavity ringdown laser spectroscopy (CRDS). First, oxygen of nitrate ( $\text{O-NO}_3$ ) is equilibrated with oxygen of water ( $\text{O-H}_2\text{O}$ ) at low pH (0.1) and  $80^\circ\text{C}$  in sealed borosilicate tubes for at least three days. After neutralizing the solution ( $\text{pH}\sim 7$ ), the isotopic composition of the equilibrated water is determined by CRDS on a Picarro L-2140i analyser.  $\delta^{17}\text{O}$  and  $\delta^{18}\text{O}$  of waters are scaled to V-SMOW and V-SLAP. International references USGS-34, USGS-35 and IAEA- $\text{NO}_3$  are used to calibrate in-house nitrate standards, that in turn are utilized for calibration of unknowns. We provide isotopic measurements of synthetic and natural nitrates with a wide range of  $\Delta^{17}\text{O}$  values. In addition, we demonstrate a direct inter-lab comparison between the results of  $\Delta^{17}\text{O}$  obtained by IE-CRDS and the classic method of thermal-decomposition of nitrate followed by isotope ratio mass spectrometry of  $\text{O}_2$  (TD-IRMS) at Louisiana State University. The precision of our method improves with sample size. This is  $0.8\text{‰}$  for  $\delta^{17}\text{O}$ ,  $1.8\text{‰}$  for  $\delta^{18}\text{O}$  and  $0.2\text{‰}$  for  $\Delta^{17}\text{O}$  when using a  $\text{O-NO}_3/\text{O-H}_2\text{O}$  of  $0.0112\pm 0.0001$  (e.g. 1 mg of  $\text{NaNO}_3$  in 50  $\mu\text{l}$  of the acid solution). This reproducibility is comparable to that from other methods. IE-CRDS and TD-IRMS methods yield similar isotopic results for the analysis of both synthetic and natural nitrate samples within analytical errors of the two methods. The IE-CRDS method is cheaper, safer, and requires less tedious sample preparation and analysis than IRMS-based methods, with a relatively high sample throughput ( $\sim 12$  samples/day).