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Modeling the impact of heterogeneous reactions of chlorine on summertime nitrate formation in Beijing, China

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A comprehensive chlorine heterogeneous chemistry is incorporated into the Community Multiscale Air Quality (CMAQ) model to evaluate the impact of chlorine-related heterogeneous reaction on diurnal and nocturnal nitrate formation and quantify the nitrate formation from gas-to-particle partitioning of HNO₃ and from different heterogeneous pathways. The results show that these heterogeneous reactions increase the atmospheric Cl2 and ClNO₂ level, leading to an increase of the nitrate concentration by $\sim 10\%$ in the daytime. However, these reactions also lead to a decrease the nocturnal nitrate by $\sim 20\%$. Sensitivity analyses of uptake coefficients show that the empirical uptake coefficient for the O₃ heterogeneous reaction with chlorinated particles may lead to the large uncertainties in the predicted Cl2 and nitrate concentrations. The N2O5 uptake coefficient with particulate Cl- concentration dependence performs better to capture the concentration of ClNO2 and nocturnal nitrate concentration. The reaction rate of OH and NO₂ in daytime increases by \sim 15% when the heterogeneous chlorine chemistry is incorporated, resulting more nitrate formation from HNO3 gas-to-particle partitioning. By contrast, the contribution of the heterogeneous reaction of N2O5 to nitrate concentrations decreases by about 27% in the nighttime when its reactions with chloriated particles are considered. However, the generated gas-phase ClNO2 from the heterogeneous reaction of N2O5 and chlorine-containing particles further decompose to increase the nitrate by 6%. In general, this study highlights the potential of significant underestimation of daytime and overestimation of nighttime nitrate concentrations for chemical transport models without proper chlorine chemistry in the gas and particle phases.