



## Mg incorporation in $\text{Fe}_5\text{O}_6$ and its relevance as a possible mantle phase

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Recent studies on the simple Fe-O system demonstrate that the list of possible high-pressure phases has to be expanded by adding new oxide phases such as  $\text{Fe}_4\text{O}_5$ ,  $\text{Fe}_5\text{O}_6$ ,  $\text{Fe}_7\text{O}_9$ ,  $\text{Fe}_9\text{O}_{11}$  [1,2,3,4,5]. These new stoichiometries highlight that our understanding of such a simple binary system was far too limited.  $\text{Fe}_4\text{O}_5$  and  $\text{Fe}_5\text{O}_6$  has been proven to be stable over a wide range of pressure and temperature corresponding to those of the Earth's deep upper mantle and transition zone. Based on their ability to incorporate  $\text{Fe}^{3+}$  as well as  $\text{Fe}^{2+}$ , the stability of these mixed-valenced Fe-oxides are related to the prevailing oxygen fugacity in the Earth's interior.

Since  $M_4\text{O}_5$  stoichiometries become stable as post-spinel phases and can coexist with Si-bearing phases [6,7,8], this stoichiometry needs to be considered as a potential constituent of the mantle assemblage. However, based on the fact that  $\text{Fe}_4\text{O}_5$  has a higher  $\text{Fe}^{3+}/\text{Fe}_{\text{tot}}$  ( $\approx 0.5$ ) than  $\text{Fe}_5\text{O}_6$  with  $\text{Fe}^{3+}/\text{Fe}_{\text{tot}} \approx 0.4$ , the latter phase might be more relevant for the expected redox conditions in the Earth. Since  $\text{Fe}_5\text{O}_6$  and  $\text{Fe}_4\text{O}_5$  share the same orthorhombic *Cmcm* space group and  $\text{Fe}_4\text{O}_5$  forms a complete solid solution along the  $\text{Fe}_2\text{Fe}_2\text{O}_5$ - $\text{Mg}_2\text{Fe}_2\text{O}_5$  binary join [8], the same behavior might be also expected for  $\text{Fe}_5\text{O}_6$ . But, the ability of  $\text{Fe}_5\text{O}_6$  to form  $\text{Mg}^{2+}$ - $\text{Fe}^{2+}$  solid solutions (i.e.  $\text{Fe}_5\text{O}_6$ - $\text{Mg}_3\text{Fe}_2\text{O}_6$ ) remains essentially unknown, which is why the aim of this experimental study is to investigate (i) the solubility of Mg in the  $\text{Fe}_5\text{O}_6$  structure and (ii) its phase relations at relevant mantle conditions.

Experiments were carried out at 8-20 GPa and temperatures 1000-1600°C using the multi-anvil apparatus. Starting materials were stoichiometric mixtures of MgO,  $\text{Fe}^0$  and  $\text{Fe}_3\text{O}_4$  components. Run products were analysed by electron microprobe, powder XRD and in some cases by TEM.

Preliminary results indicate that a solid solution along the  $\text{Fe}_5\text{O}_6$ - $\text{Mg}_3\text{Fe}_2\text{O}_6$  binary is incomplete; the  $\text{Mg}_3\text{Fe}_2\text{O}_6$  is not stable. The phase diagram for  $\text{Mg}_{0.5}\text{Fe}_{2.5}\text{Fe}_2\text{O}_6$  indicates that the phase relations strongly deviate from that of the  $\text{Fe}^{2+}$ -endmember composition. The low-pressure stability of  $\text{Fe}_5\text{O}_6$  is limited down to  $\sim 9$  GPa over a wide temperature range (900-1400°C), where the assemblage  $M_3\text{O}_4 + (\text{Mg,Fe})\text{O}$  becomes stable. The stability field of  $\text{Fe}_5\text{O}_6$  pinches out with increasing pressure, but is still stable at 28 GPa (at 1500°C). In contrast, in the Mg-bearing compositions an intervening  $M_4\text{O}_5 + (\text{Mg,Fe})\text{O}$  assemblage is stable between 9-18 GPa. With increasing pressure,  $\text{Mg}_{0.5}\text{Fe}_{2.5}\text{Fe}_2\text{O}_6$  becomes stable relative to the assemblage  $(\text{Mg,Fe})_2\text{Fe}_2\text{O}_5 + (\text{Mg,Fe})\text{O}$  and seems to be limited to  $\sim 20$  GPa.  $\text{Mg}_{0.5}\text{Fe}_{2.5}\text{Fe}_2\text{O}_6$  can coexist with  $(\text{Mg,Fe})\text{O}$  and/or  $M_9\text{O}_{11}$ .

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