



## Hydroperoxyl radical measurements at the canopy-atmosphere interface in the Amazon rainforest

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Hydroperoxyl radicals ( $\text{HO}_2$ ) account for bulk of the  $\text{HO}_x$  number density and together with the hydroxyl radicals (OH) maintain the self-cleaning capacity of the atmosphere. The reaction of  $\text{HO}_2$  with NO forming  $\text{NO}_2$  is an important chemical source of tropospheric  $\text{O}_3$ , which is a greenhouse gas detrimental to the health of humans and plants. The oxidation of CO and hydrocarbons emitted from various anthropogenic and biogenic processes lead to the formation of  $\text{HO}_2$ . The chemical lifetime of  $\text{HO}_2$  is about a few hundred seconds, sufficient to feed oxidation processes and interact with turbulences in the microscale e.g.  $\text{HO}_2$  formed due to oxidation of biogenic hydrocarbons above forest canopies can be transported down into the canopies and react with NO and  $\text{O}_3$  to recycle OH.

The vast green stretches of Amazon are not only a storehouse of carbon but also the source of copious biogenic hydrocarbons, which impact the regional and global atmospheric chemistry. The export of primary hydrocarbons and their oxidized products from the Amazon to the global troposphere depends on the oxidation capacity at the canopy-atmosphere interface. To understand the chemical processes influencing the oxidation capacity over the Amazons, measurements of OH,  $\text{HO}_2$  along with related chemical, meteorological and photochemical parameters were carried out during an intensive field campaign SEGAM (SEGregation experiment in the AMazon) in November 2015. Fast 5 Hz measurements of OH and  $\text{HO}_2$  using laser induced fluorescence were conducted on a tower at 40 m altitude at the canopy-atmosphere interface to understand the sources, sinks and chemistry of atmospheric oxidants in relation to the variation in NO and isoprene fluxes down in the forest. This presentation deals with the measured variability of  $\text{HO}_2$  during SEGAM and its chemical sources and sinks estimated with a photochemical box model.