

Effect of Various Fog Processing Stages on the Oxidation Process of Organic Aerosol

Anil Kumar Mandariya¹; Tarun Gupta^{1,2}; S. N. Tripathi^{1,2}

¹Department of Civil Engineering, Indian Institute of Technology, Kanpur, India ²Centre for Environmental Science and Engineering, IIT, Kanpur, India



2. Effect of various fog processing stages on OA, composition, and oxidation processes

Poster X5.307 EGU2019-1609



Introduction

- Estimates of global secondary organic aerosol (SOA) budget are still largely uncertain due to understanding of its oxidation processes, especially in the heavily polluted region in which biomass burning (BB) sources dominate under fog periods (high relative humidity (RH) and lower temperature (T) conditions).
- > Organic aerosol (OA) exhibit various phases or amorphous solid) ambient prevailing temperature and RH.
- > In fog processing scenario, bulk diffusion coefficients of OH and levoglucosan (a marker of biomass burning OA) enhance as compared to dry atmospheric conditions which subsequent effects on the SOA production due to changes in the diffusivity of oxidant and organics.
- > Fog duration varies from minutes to several hours, highly depending on local metrological parameters which change with time of the day like RH, temperature and wind speed. Therefore, to thoroughly understand the fog processing phenomena, as well as its effect on the OA oxidation processes, relatively number of stages during the fog-cycle, are required.

Hypothesis

- Activation fog period and dissipating fog periods represent the period just before (fog formation) and after (evaporation of fog droplets) the fog period, where number concentration of fine droplets, as well as ALWC, are higher, are crucial in any fog life cycle.
- Activation-fog-period and dissipating-fogperiod are crucial for this study to separate the combined effect of high ALWC and LWC during pre-fog-period and contrast it with post-fog-period having low ALWC and zero LWC.

Methods and Materials

Site: IIT Kanpur (26.5°N, 80.3°E, and 142 m above mean sea level), located in the center of Indo-Gangetic Plain (IGP)

Study Period: 30th December 2015 to 23rd January 2016

Instrumentation: HR-ToF-AMS, CDP, RH & T Sensor

Selection Criteria for Fog Processing Periods: LWC: Liquid Water Content of fog droplets

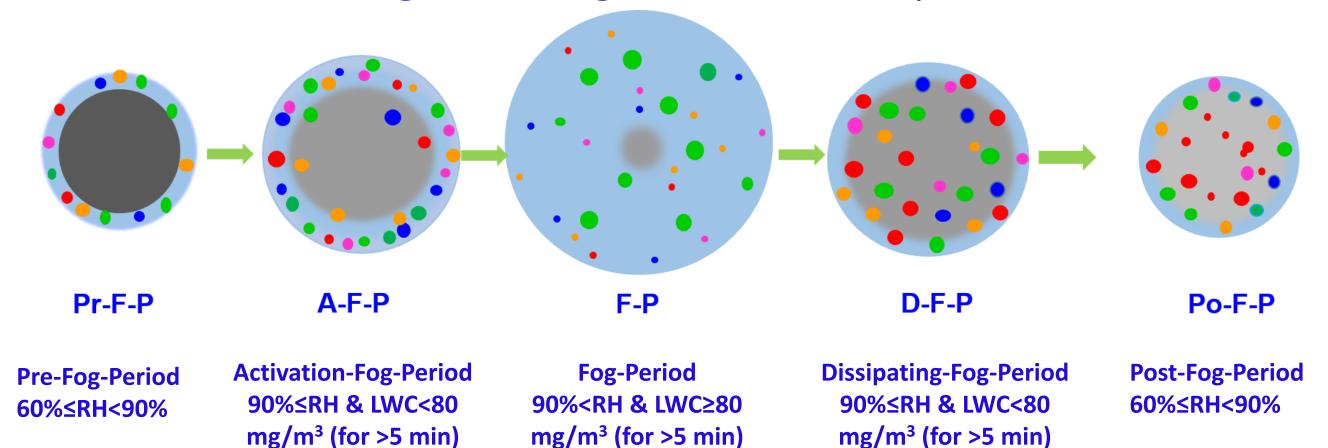


Figure 1. Various stages of fog life cycles.

Fog Events: 04 Number

Fog Processing Periods: 05 Numbers

Data Analysis: Wind speed (WS) and wind direction (WD) were obtained from NOAA ARL dataset. Aerosol liquid water content (ALWC) associated with inorganic species was predicted using E-AIM-IV model.

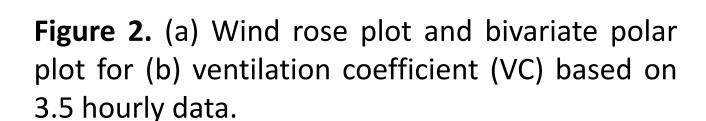
Results

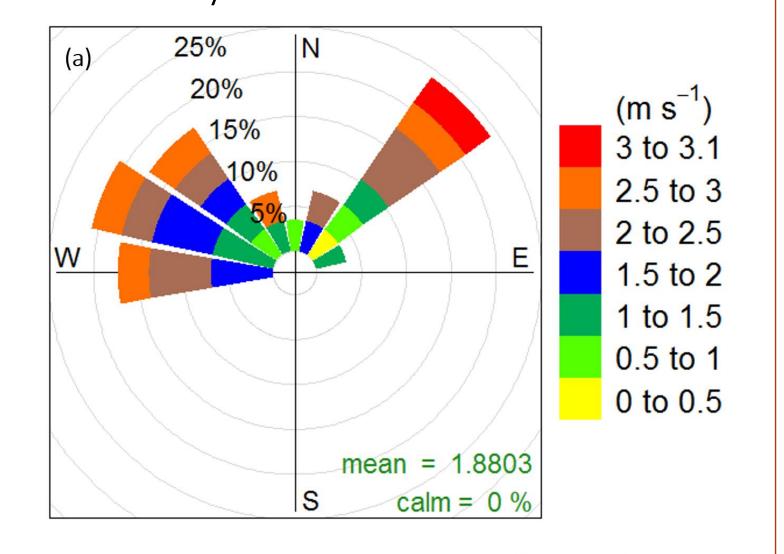
1. Average meteorological parameters for fog processing periods

Table 1. Fog processing period's average±standard deviation meteorological parameters: solar radiation (SR), planetary boundary layer (PBL), F (fine fog droplets, diameter: 4-16 μm), M (Medium fog droplets, diameter: 16-22 μm), C (Coarse fog droplets, diameter: >22 µm), LWC (fog droplets liquid water content), ALWC (aerosol liquid water content).

Meteorology	Fog Processing Period				
Parameter	Pr-F-P	A-F-P	F-P	D-F-P	Po-F-P
RH (%)	83.7±6.7	93.1±1.2	94.4±1.3	94.1±2.1	75.5±8.8
T (°C)	13.0±2.9	10.1±2.7	6.8±2.4	8.8±3.8	16.3±3.8
WD (° from N)	232.4±134.4	147.4±140.2	216.2±134.4	210.1±124.0	218.1±113.5
WS (m/s)	2.1±0.8	1.9±0.7	1.5±0.6	1.5±0.4	1.9±0.5
PBL (m)	7.5±15.0	25.2±3.5	45.3±34.2	399.6±339.3	858.8±426.5
SR (W/m²)	0	0	19.9±23.2	156.4±109.1	479.4±157.6
Fractional of Fog Droplets Number Concentration (%)		83.2±6.0 (F) 6.3±2.4 (M) 10.5±4.1 (C)	69.7±8.1 (F) 11.1±3.0 (M) 19.2±6.4 (C)	84.3±15.8 (F) 5.1±5.5 (M) 10.6±11.9 (C)	
LWC (mg/m³)		29.9±15.9	173.7±74.8	15.3±20.5	
ALWC (μg/ m³)	207.9±111.1	512.8±128.9	440.0±283.2	541.9±220.4	144.1±108.8

- **➢ Wind speeds were found to be ranging from 0.5 -3.1 m/s** (1.9±0.7 m/s), which results in a stable atmospheric condition.
- > VC≤ 2000 m²/s indicates the "bad" category from pollution dispersion point of view during various fog life cycles.
- > Suggest that OOA formation is most likely of local origin





Frequency of counts by wind direction (%)

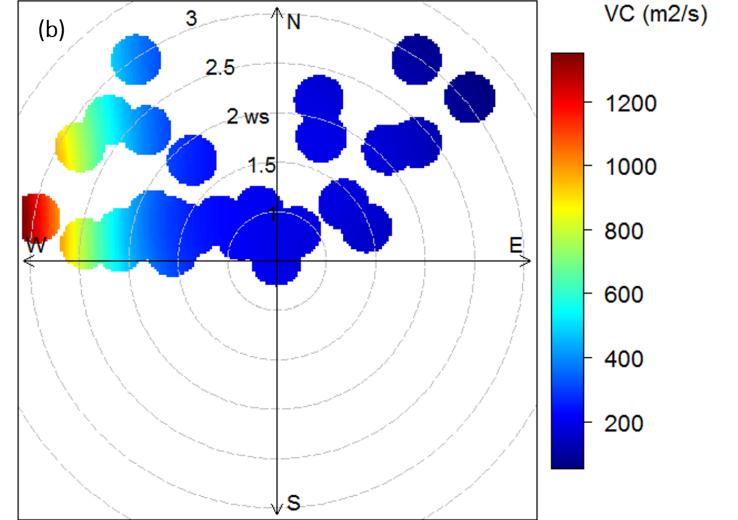


Figure 5. Evaluation OA oxidation processes in terms of formation processes and f43, f44, f60.

♀ 0.57 **⊢**

0.17 -

တ္တံ -0.15 -

course of fog life cycles

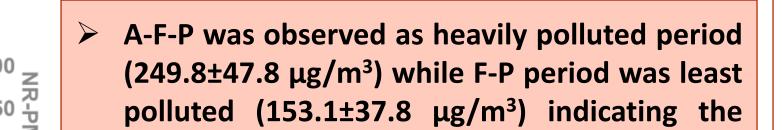
Figure 3. The overview of the variation in overall value of various

-0.15

• 0.12 ·

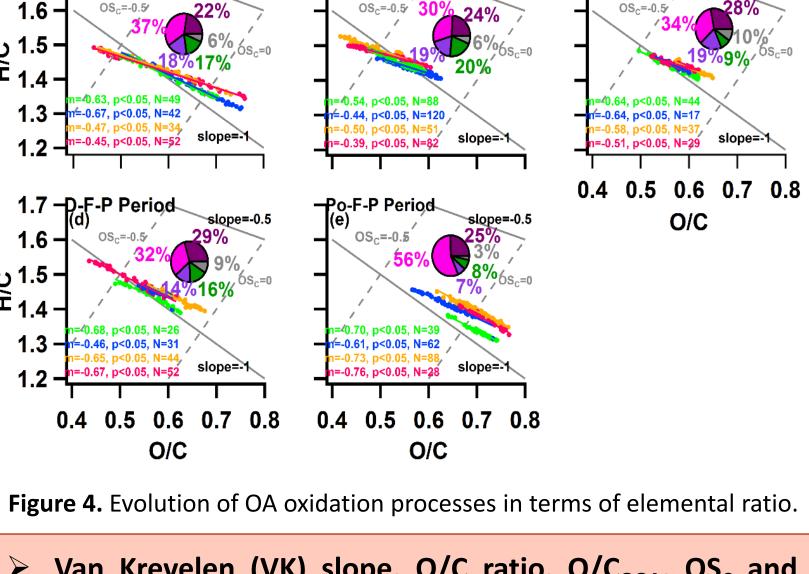
15.0 14.0 13.0 12.0 11.0

parameters for the various fog processing periods during the



wet removal through grown fog droplets.

- Aerosol in A-F-P and D-F-P were observed acidic as well as with high ALWC.
- OOA-1 processed mainly through biomass (BB) emission suggesting is a good enhanced significantly under high biomassburning emissions during A-F-P and D-F-P under acidic aerosol conditions.
- O/C of OA and OOA varied during the all fog processing periods with a maximum during Po-F-P, however low O/C during A-F-P and D-F-P was observed to associate with high OA loading.



- Van Krevelen (VK) slope, O/C ratio, O/C_{OOA}, OS_C and $(OS_c)_{OOA}$, substantially varies throughout all the fog processing periods.
- Oligomerization mechanism possibly significant for the formation of OOA along with functionalization of -OH and carbonyl (aldehyde/ketone) moieties during A-F-P and D-F-P, respectively accompanied by acidic aerosol as well as high aerosol liquid water content (ALWC) condition.
- > Fragmentation process can be dominant along with functionalization of (aldehyde/ketone) and -RCOOH moieties during F-P and Po-F-P periods, respectively.

Conclusions

- > Our results suggest that current SOA models should consider crucial A-F-P and D-F-P fog processing periods separately from pre-fog and post-fog, respectively to evaluate oxidative properties and mass concentration of SOA.
- > Oligomerization and functionalization mechanisms dominating the oxidation processes of OA during A-F-P and D-F-P.
- > Fragmentation and functionalization mechanisms govern the oxidation process of OA during F-P and Po-F-P
- > Both activation as well as dissipating fog processing periods have acidic aerosol and high aerosol liquid water content

Contact Information

Anil Kumar Mandariya

Email: mandaria@iitk.ac.in, anil.mandarias@gmail.com Phone: +918960541440

> **Prof. Tarun Gupta** Email: tarun@iitk.ac.in Prof. S.N. Tripathi Email: snt@iitk.ac.in

References

Mandariya, A.K., Gupta, T., Tripathi, S.N., 2019. Effect of Aqueous-Phase Processing on the Formation and Evolution of Organic Aerosol (OA) Under Different Stages of Fog Life Cycles. Atmos. Environ. 206, 60-71

2. Xu, W., Han, T., Du, W., Wang, Q., Chen, C., Zhao, J., Zhang, Y., Li, J., Fu, P., Wang, Z., Worsnop, D.R., Sun, Y., 2017. Effects of aqueous-phase and photochemical processing on secondary organic aerosol formation and evolution

- Beijing, China. Environ. Sci. Technol. 51, 762-770. https://doi.org/10.1021/acs.est.6b04498 Gilardoni, S., Massoli, P., Paglione, M., Giulianelli, L., Carbone, C., Rinaldi, M., Decesari, S., Sandrini, S., Costabile, F., Gobbi, G.P., Pietrogrande, M.C., Visentin, M., Scotto, F., Fuzzi, S., Facchini, M.C., 2016. Direct observation of aqueous secondary organic aerosol from biomass-burning emissions. Proc. Natl. Acad. Sci. 113, 10013–10018. https://doi.org/10.1073/pnas.1602212113
- McNeill, V.F., 2015. Aqueous organic chemistry in the atmosphere: Sources and chemical processing of organic aerosols. Environ. Sci. Technol. 49, 1237–1244. https://doi.org/10.1021/es5043707
- Gupta, T., Mandariya, A., 2013. Sources of submicron aerosol during fog-dominated wintertime at Kanpur. Environ. Sci. Pollut. Res. 20, 5615–5629. https://doi.org/10.1007/s11356-013-1580-6 6. Kroll, J.H., Donahue, N.M., Jimenez, J.L., Kessler, S.H., Canagaratna, M.R., Wilson, K.R., Altieri, K.E., Mazzoleni, L.R., Wozniak, A.S., Bluhm, H., Mysak, E.R., Smith, J.D., Kolb, C.E., Worsnop, D.R., 2011. Carbon oxidation state as a
- metric for describing the chemistry of atmospheric organic aerosol. Nat. Chem. 3, 133–139. https://doi.org/10.1038/nchem.948 7. Ng, N.L., Canagaratna, M.R., Jimenez, J.L., Chhabra, P.S., Seinfeld, J.H., Worsnop, D.R., 2011. Changes in organic aerosol composition with aging inferred from aerosol mass spectra. Atmos. Chem. Phys. 11, 6465–6474. https://doi.org/10.5194/acp-11-6465-2011
- 8. DeCarlo, P.F., Kimmel, J.R., Trimborn, A., Northway, M.J., Jayne, J.T., Aiken, A.C., Gonin, M., Fuhrer, K., Horvath, T., Docherty, K.S., Worsnop, D.R., Jimenez, J.L., 2006. Field-deployable, high-resolution, time-of-flight aerosol mass spectrometer. Anal. Chem. 78, 8281-8289. https://doi.org/10.1021/ac061249n

Acknowledgements

We acknowledge the support of IIT Kanpur for providing us with HR-ToF-AMS for PG research and teaching.