

# Isotopic analyses of the Middle Miocene evaporite assemblage and its fluid inclusions (Praid, Transylvanian Basin, Romania)

Orsolya Gelencsér(1,2), László Palcsu(2), István Futó(2), and Csaba Szabó(1)

*1Lithosphere Fluid Research Lab, Eötvös University Budapest, Hungary*

*2Isotope Climatology and Environmental Research Centre, Institute for Nuclear Research, Debrecen, Hungary*

(\*correspondence: [gecso@caesar.elte.hu](mailto:gecso@caesar.elte.hu))

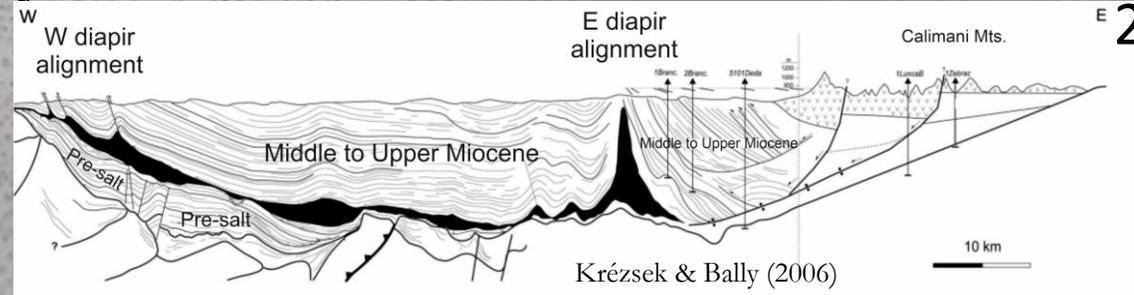


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# About the project

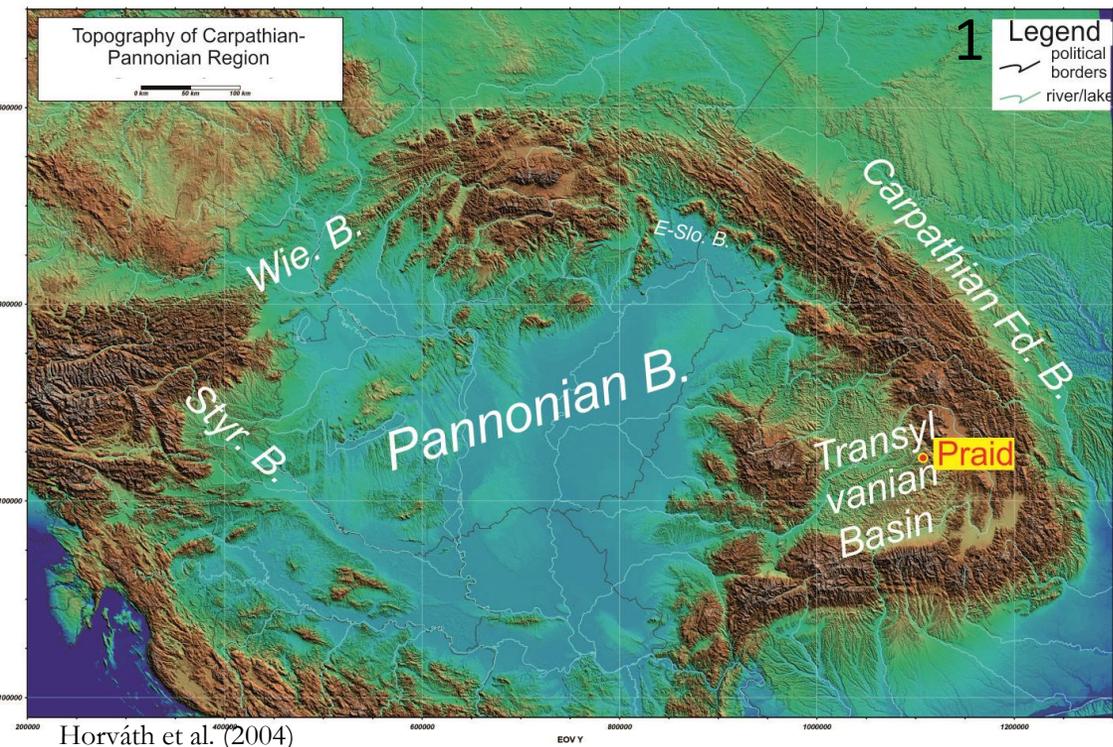
What are the geochemical signatures of the salt during the Badenian Salinity Crisis?



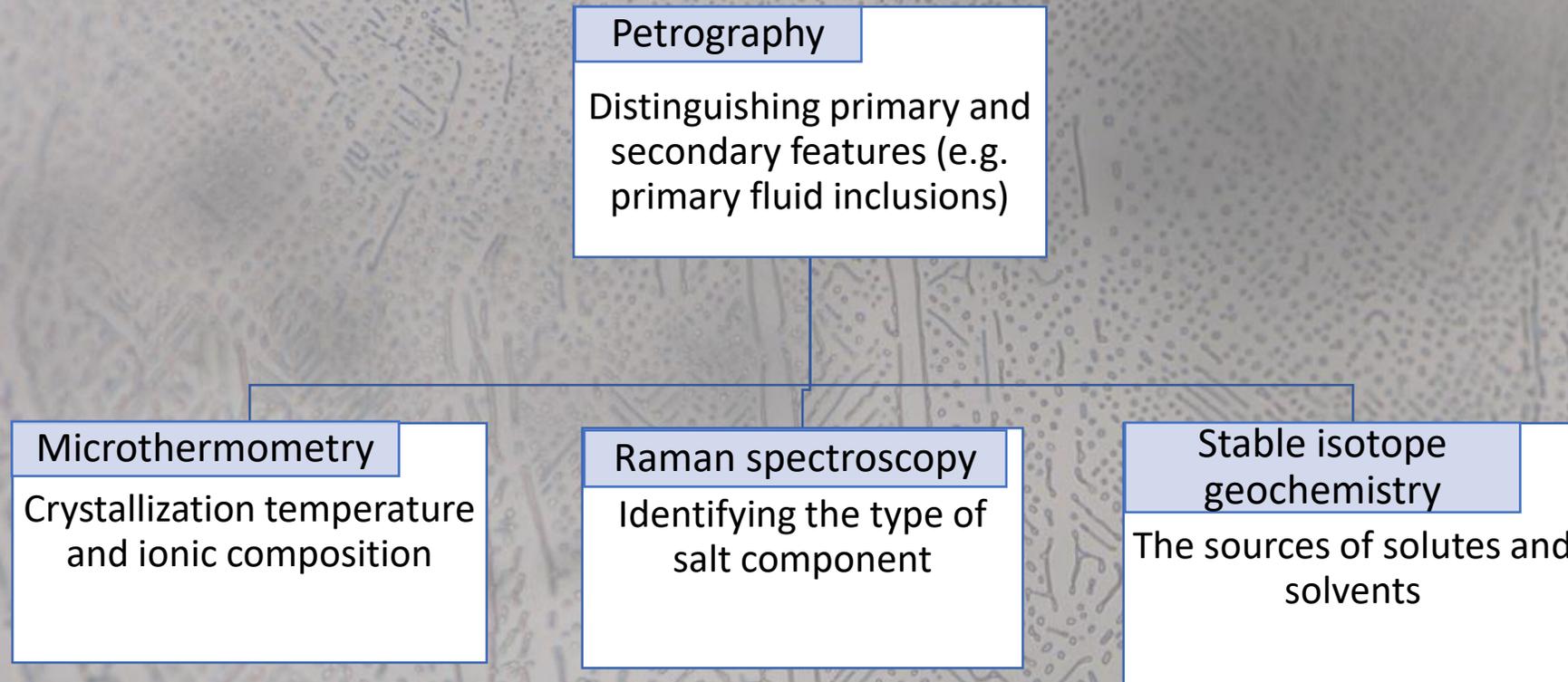
(1) The objective of this study is to constrain the formation conditions of deposition of the Middle Miocene marine evaporite in the Transylvanian Basin (TB). The salt rock, formed during the Badenian Salinity Crisis (BSC), consists principally of halite (> 90 %).

(2) BSC is resulted in the deposition of large amounts of marine salt rocks which recently forms diapirs in the eastern margin of the Transylvanian Basin.

(3) Representative samples were collected from mine in Praid Salt Mine.



# Applied techniques

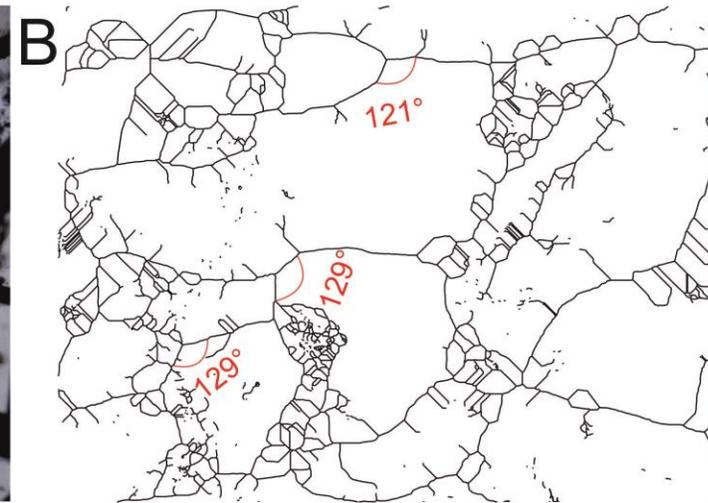
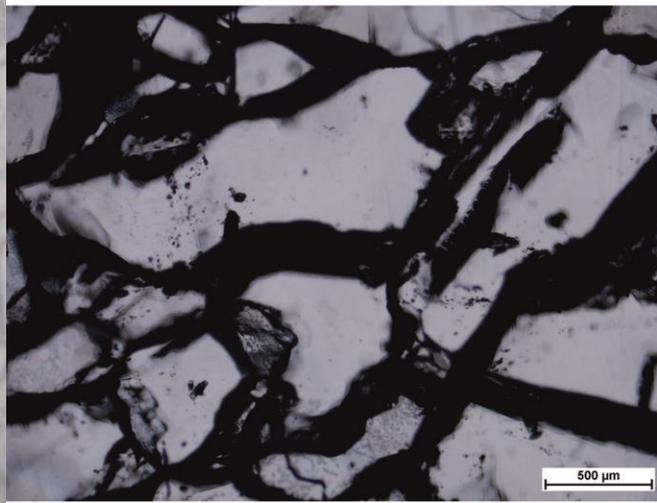
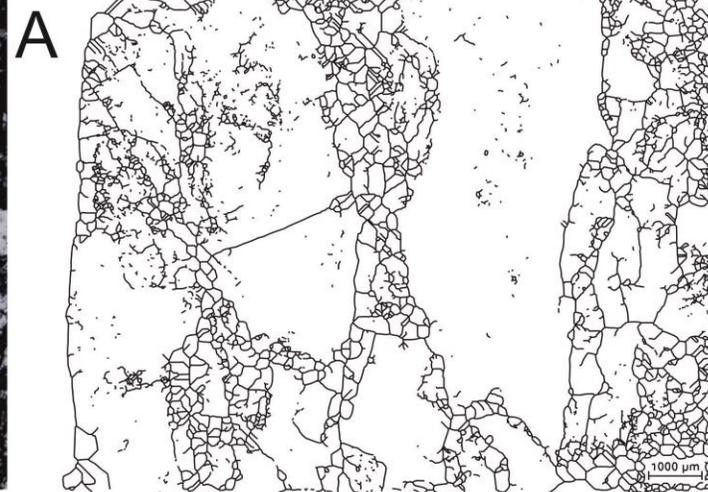
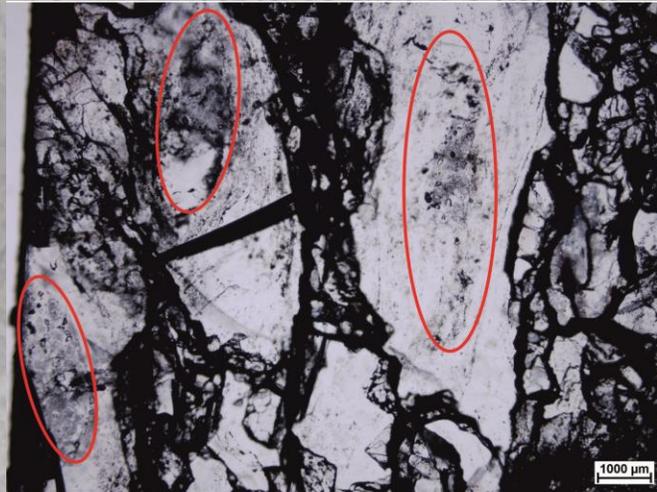
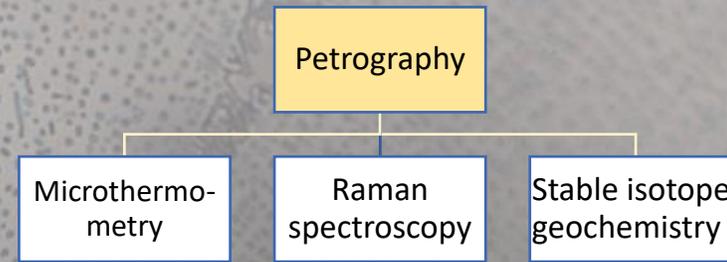


Petrographic studies (on texture and fluid inclusions) were carried out according to Hardie et al. (1985)

The identification of the solid phases by Raman spectroscopy is based on the databases of Uriarte et al. (2015), Baumgartner and Bakker (2015) and Dubessy et al. (1982).

Three constituents of the salt rock were analyzed isotopically. Halite fluid inclusions (FI) for O and H, anhydrite for S and O, dolomite for C and O.

# Textural observations

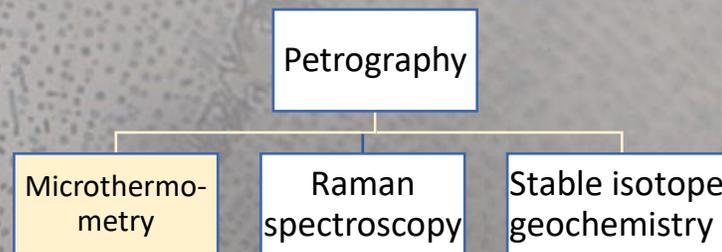


Two types of fabric can be distinguished.

A-type: Elongated halite crystals placed in a submillimeter size crystal mash. Large grains are remnants of chevron halite formed directly from the concentrated paleo-seawater.

B-type: Grainsize distribution is more consistent compared to A-type, 120° joints -> mosaic structure. Lack of primary fluid inclusions.

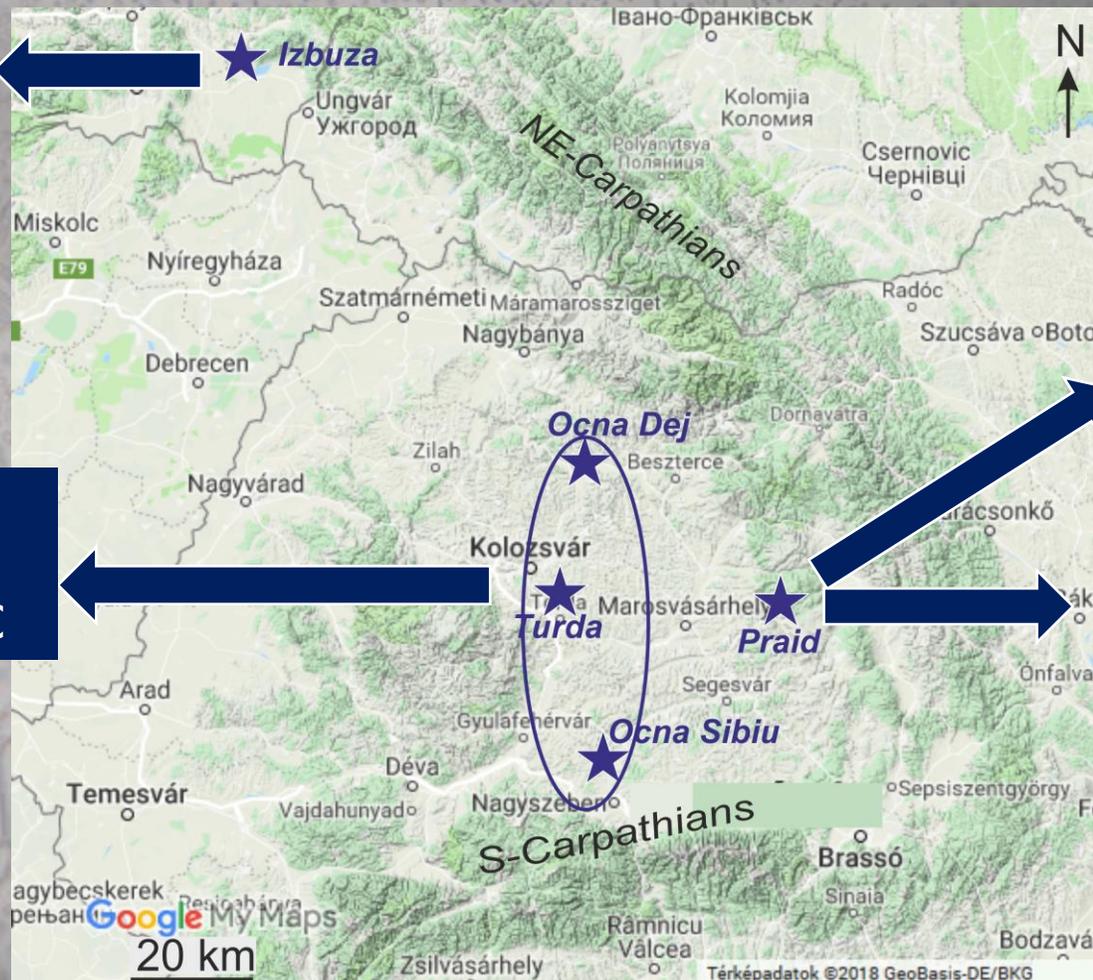
# Microthermometry



Homogenization temperatures ( $T_h$ ) measured on primary fluid inclusion are similar to the other Badenian salts.

Eutectic temperatures ( $T_e$ ) show complex chemical composition.

Galamay et al. (2004):  
 $T_h$ : 10 – 36 °C



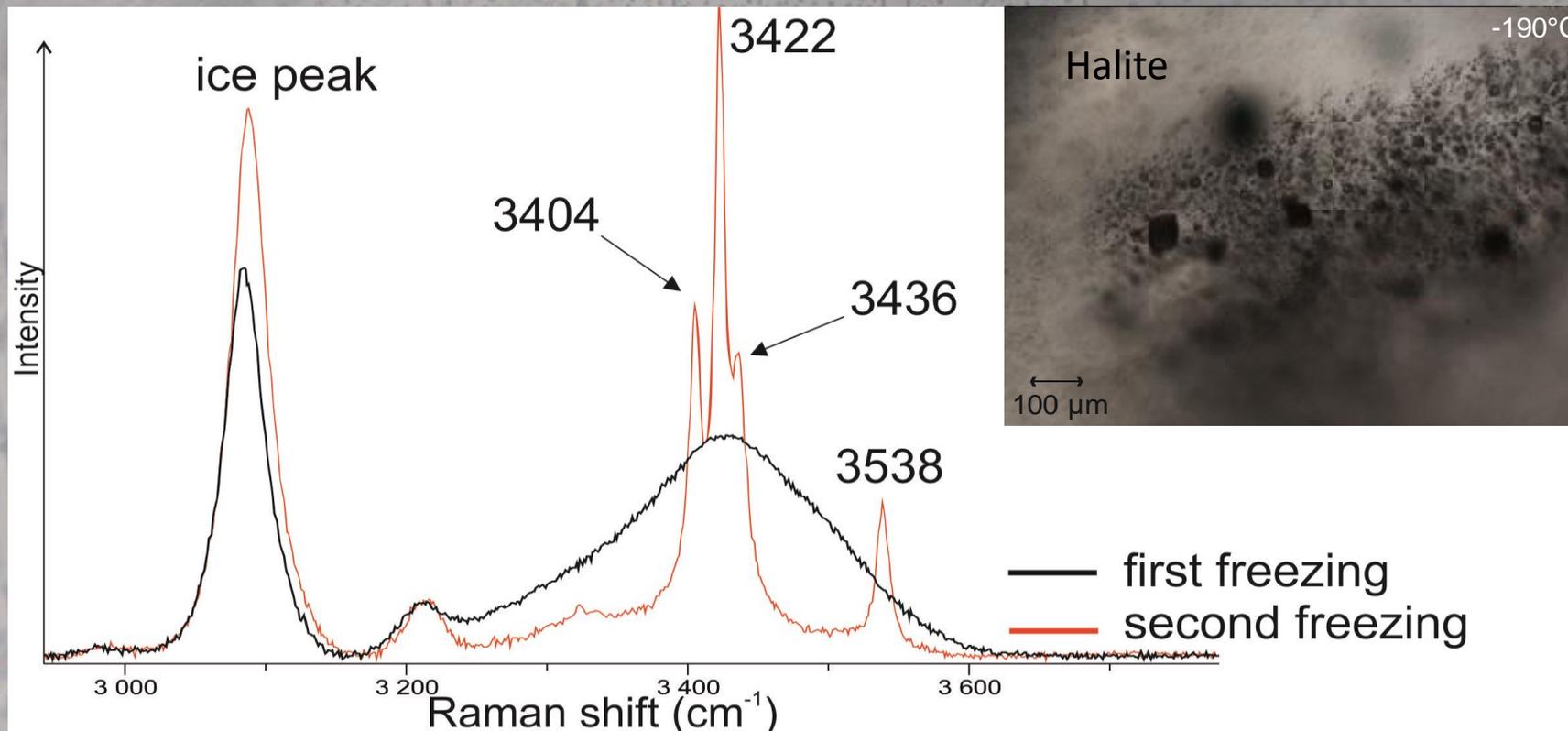
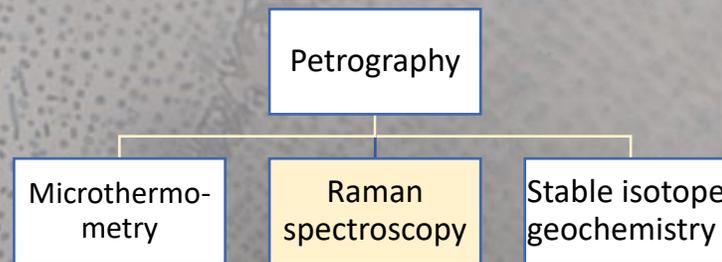
**This study:**  
 $T_h$ : 10 – 24 °C  
 $T_e$ : - 55 – -42 °C

Pintea (2008):  
 $T_h$ : 10 – 15 °C  
 $T_e$ : - 36 – -18 °C

Kátai (2017):  
 $T_h$ : 15 – 32,7 °C  
 $T_e$ : -52 – -47 °C

*Microthermometry measurements were performed on samples in the Lithosphere Fluid Research Lab (Eötvös University) using LINKAM THMSG600 stage.*

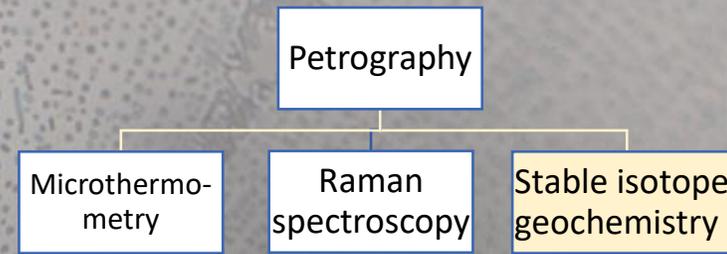
# Cryogenic Raman spectroscopy



Characteristic Raman spectra of primary fluid inclusions (black cubes on the photomicrograph) on -190 °C after first (black) and second (red) freezing in the stretching regions of structural H<sub>2</sub>O.

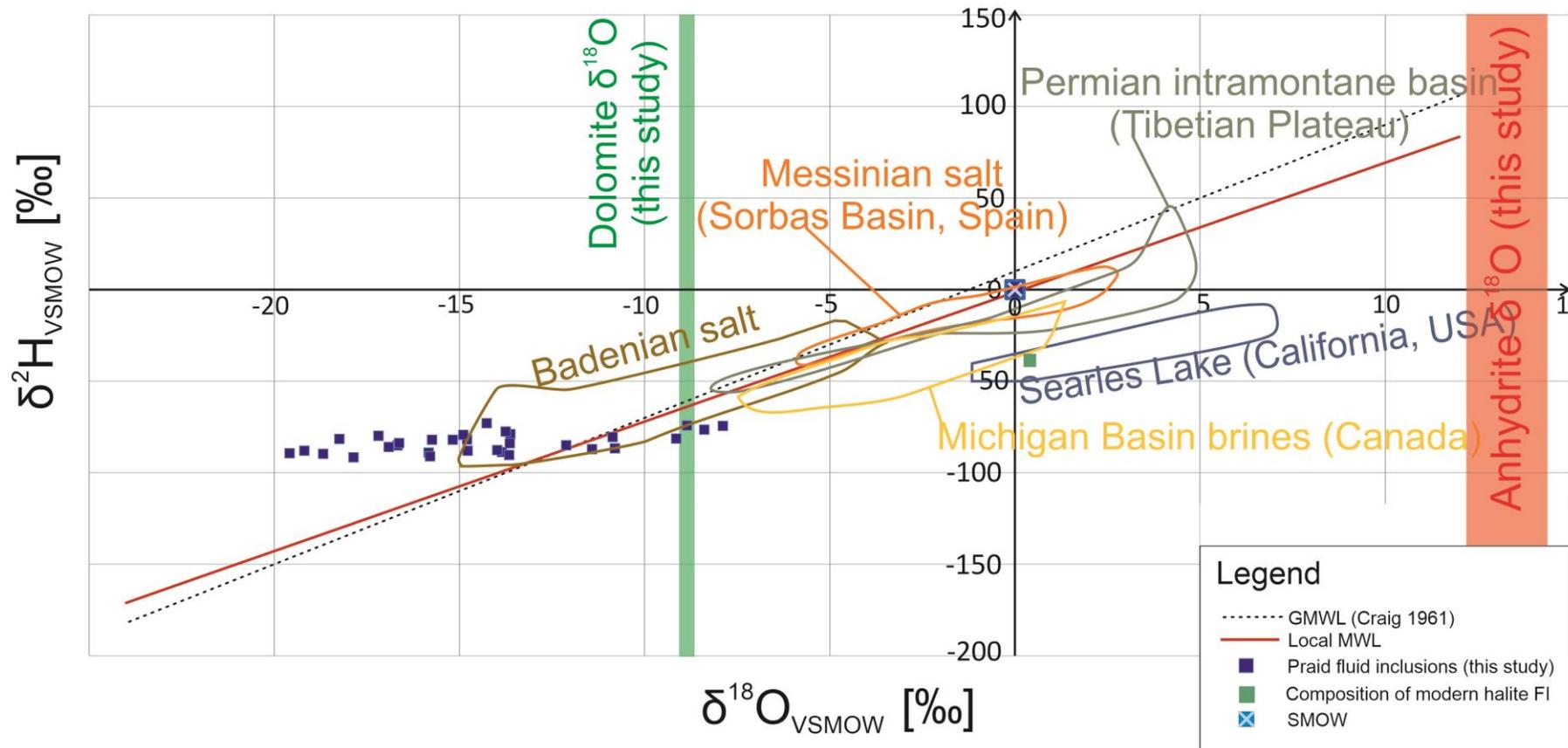
*Raman analyses were carried out in the Research and Instrument Core Facility at the Faculty of Science (Eötvös University) by a Horiba Jobin Yvon LabRam HR800.*

# Isotope geochemistry



Stable isotopic study was completed in the Isotope Climatology and Environmental Research Centre, Institute for Nuclear Research, Debrecen (Hungary).

Isotopic characteristics of primary FI draw very low-slope line. Delta values are partially overlapped with the other BSC data (Halas & Krouse, 1982).



Translucent dolomite has negative delta values ( $\delta^{18}\text{O}$  -7.07 - -4.55 ‰ and  $\delta^{13}\text{C}$  -9.03 - -8.31 ‰)

Sulfate isotope values measured in anhydrite are ranging  $\delta^{34}\text{S}$  20.4 – 22.4 ‰ and  $\delta^{18}\text{O}$  12.9 – 14.5 ‰ that coincide with the BSC deposits from the Carpathian foredeep (Peryt, 2010) and support seawater origin.

Literature of FI: (Koehler et al. 1991; Horita 1990), brines: (Horita 1990; Wilson & Long 1993), and gypsum hydration water: (Halas & Krouse 1981; Evans et al. 2015; Li et al. 2017)

# Summary

- Primary features of the Middle Miocene salt rock were detected.
- Although further studies are in process, it is clear that the geochemical characters of the evaporite minerals reveal a complex restricted hydrogeologic evolution.
- Negative isotopic characteristics of fluid inclusion H<sub>2</sub>O and carbonates suggest presence of organisms in the evaporating environment.

# Acknowledgements

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