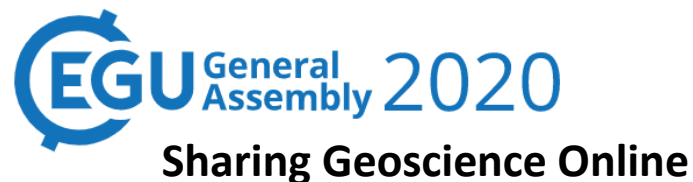


Modelling hygroscopicity-induced gas–aerosol partitioning, organic surface enrichment and cloud droplet formation

Andreas Zuend¹ & Kyle Gorkowski^{1,2}

¹ Department of Atmospheric and Oceanic Sciences, McGill University, QC, Canada

² Earth Systems Observations, Los Alamos National Laboratory, Los Alamos, New Mexico, United States



This project was undertaken with the financial support of:
Ce projet a été réalisé avec l'appui financier de :

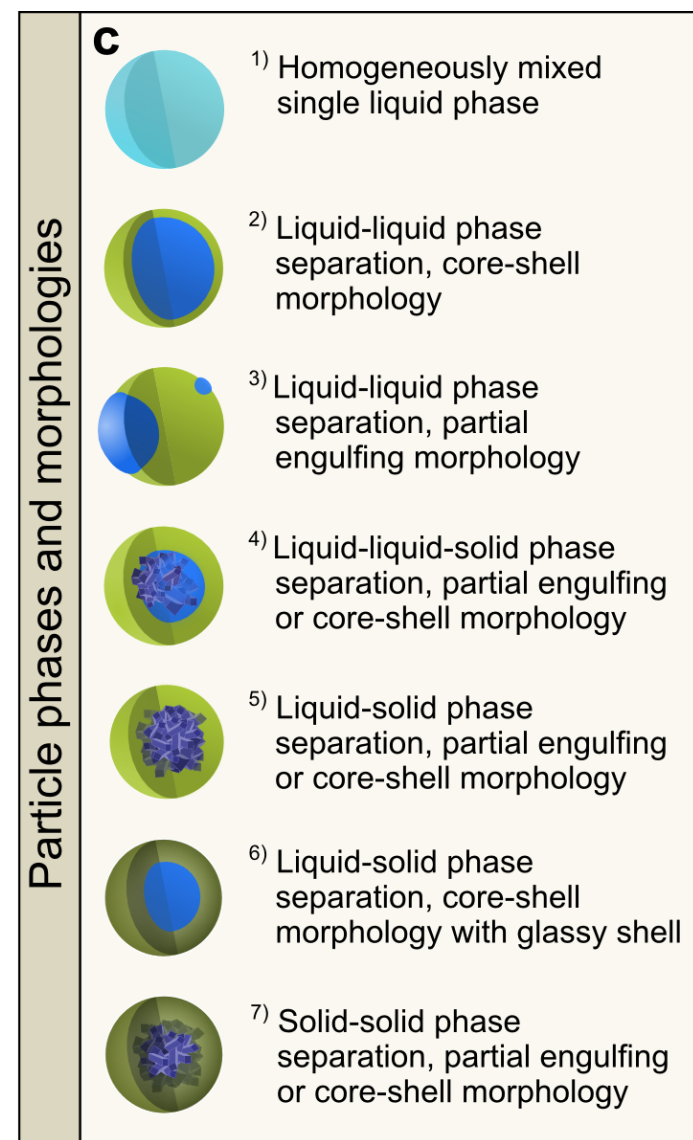
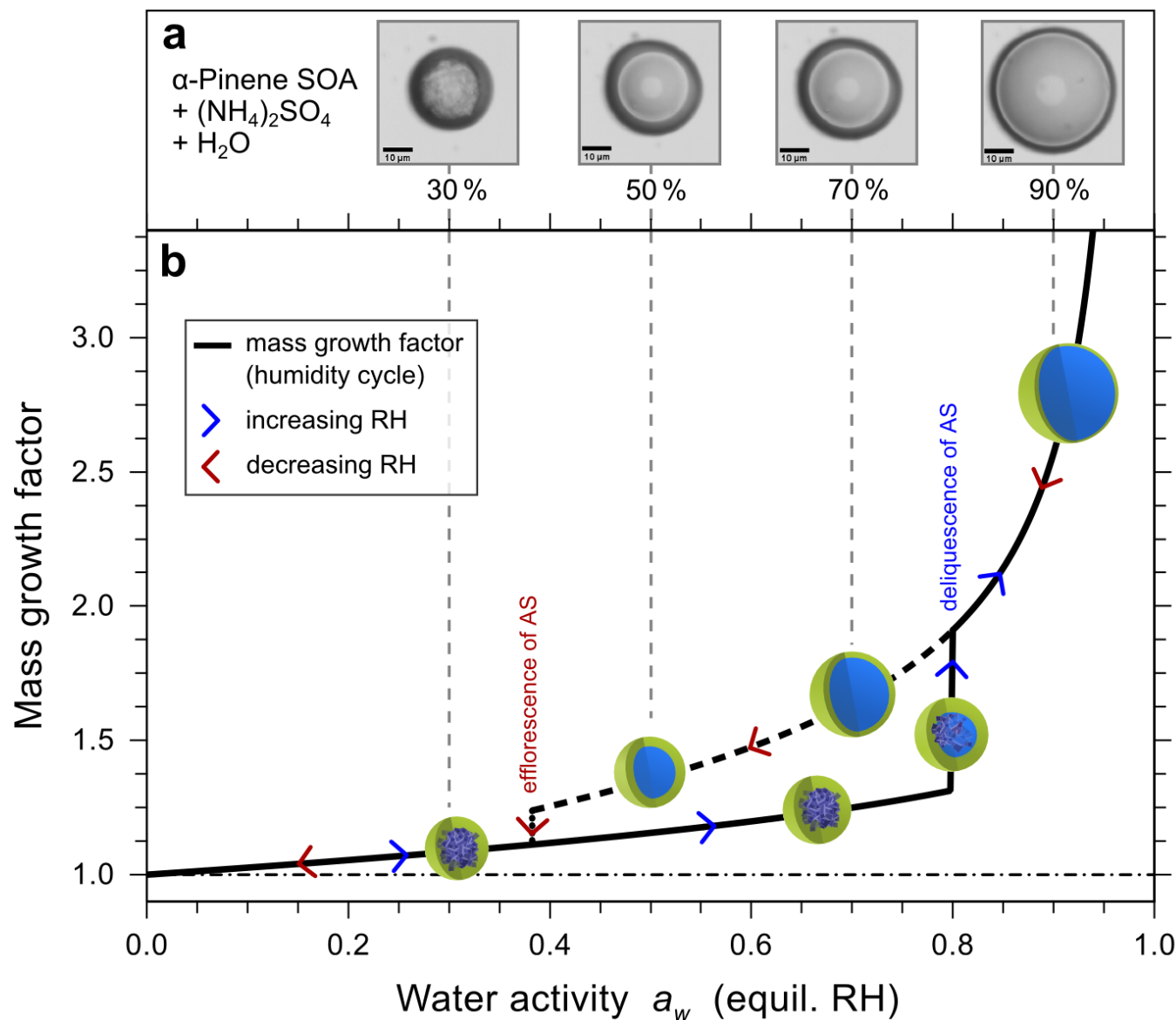


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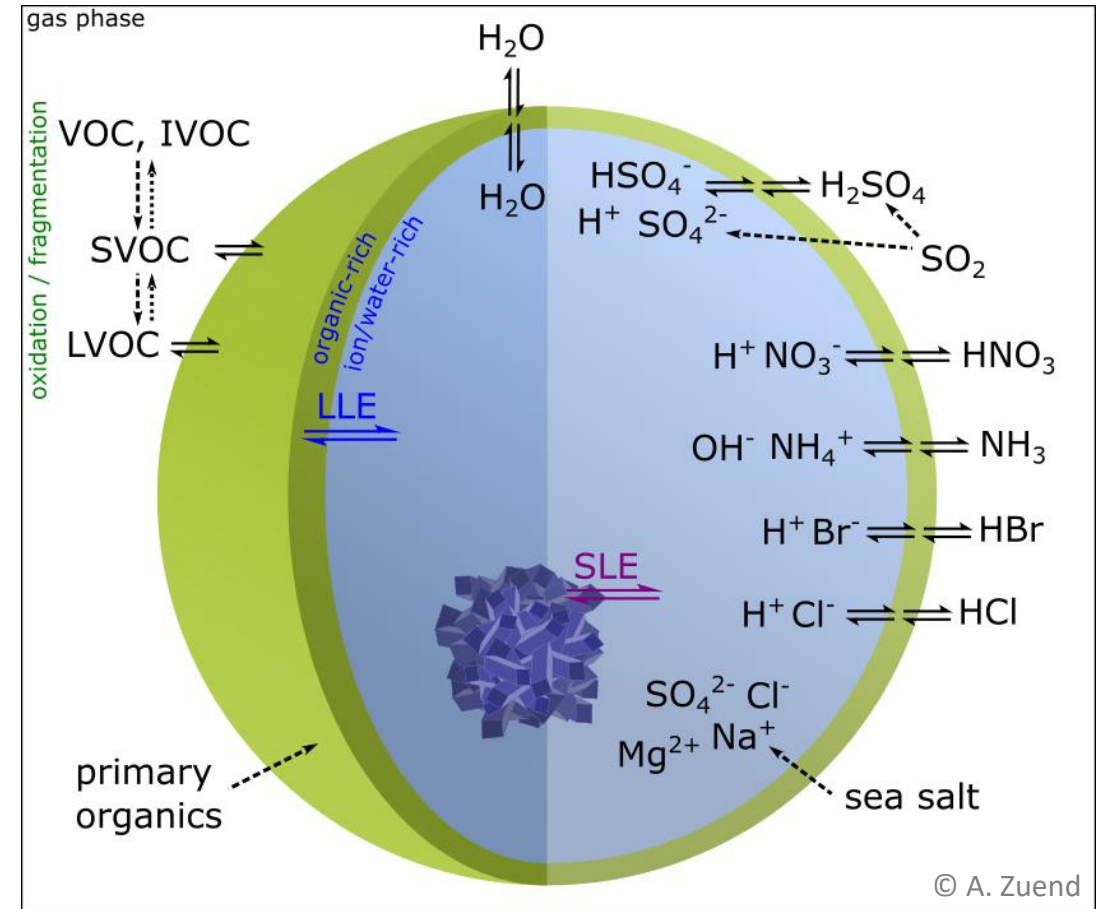
Introduction: organics & water-driven phase transitions



Organic–inorganic aerosol thermodynamics

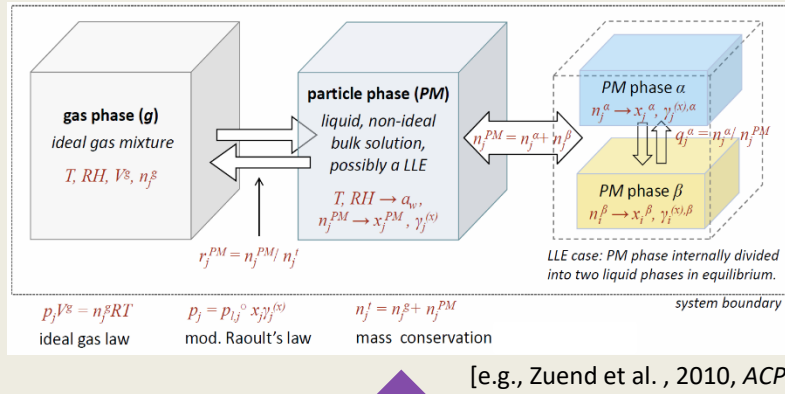
AIOMFAC-based modeling framework

- Gas–particle partitioning at different RH
- Aerosol phases, physical states, viscosity
- Non-ideal mixing, liquid–liquid equilibrium (LLE)
- Aerosol acidity (multiphase; e.g. Pye et al, 2020, *ACP*)
- Hygroscopicity & cloud condensation nucleus (CCN) properties



Modeling non-ideal mixing thermodynamics & gas-particle partitioning

Gas-particle equilibrium model:

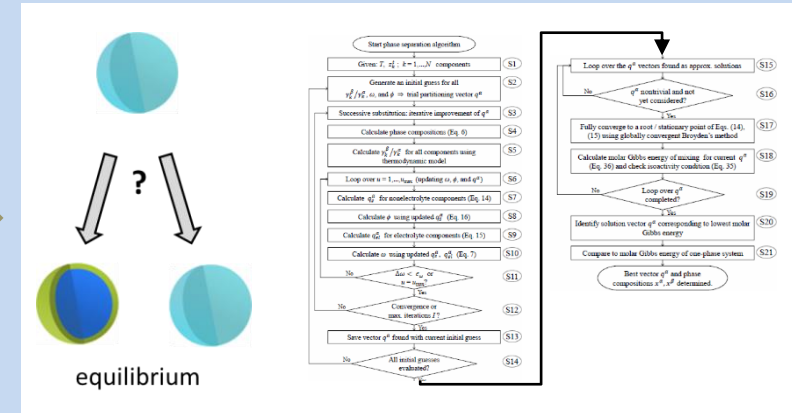


Volatility: Pure component vapour pressures as function of T (measured or predicted)

Output: (for specified input system)

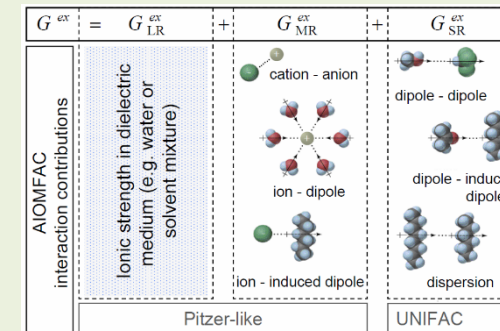
- # liquid phases; (& optionally solid phases)
- chemical compositions of all phases (bulk prediction)

LLE algorithm: Liquid-liquid phase separation?



AIOMFAC model:

Organic-inorganic mixing in a liquid phase



[Zuend et al., 2008; 2011, ACP; <https://aiomfac.lab.mcgill.ca>]

Next 3 slides outline examples of applications of the AIOMFAC model and the AIOMFAC-based equilibrium framework

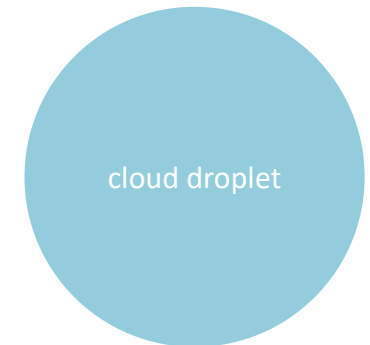
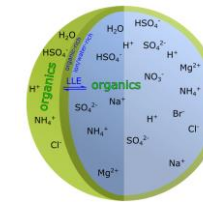
Application of AIOMFAC framework: CCN activation & the role of organics

Köhler theory:

$$S = a_w \exp \left[\frac{4 \sigma M_w}{RT \rho_w D_0 \text{HGF}} \right] \quad \text{with} \quad \text{HGF} = \frac{D}{D_0}$$

What is the role of organic aerosol constituents & liquid–liquid phase separation in affecting CCN properties?

- Thermodynamic perspective
- Aerosol population / air parcel dynamics



Field study at Mace Head: effective ultrafine CCN in marine air

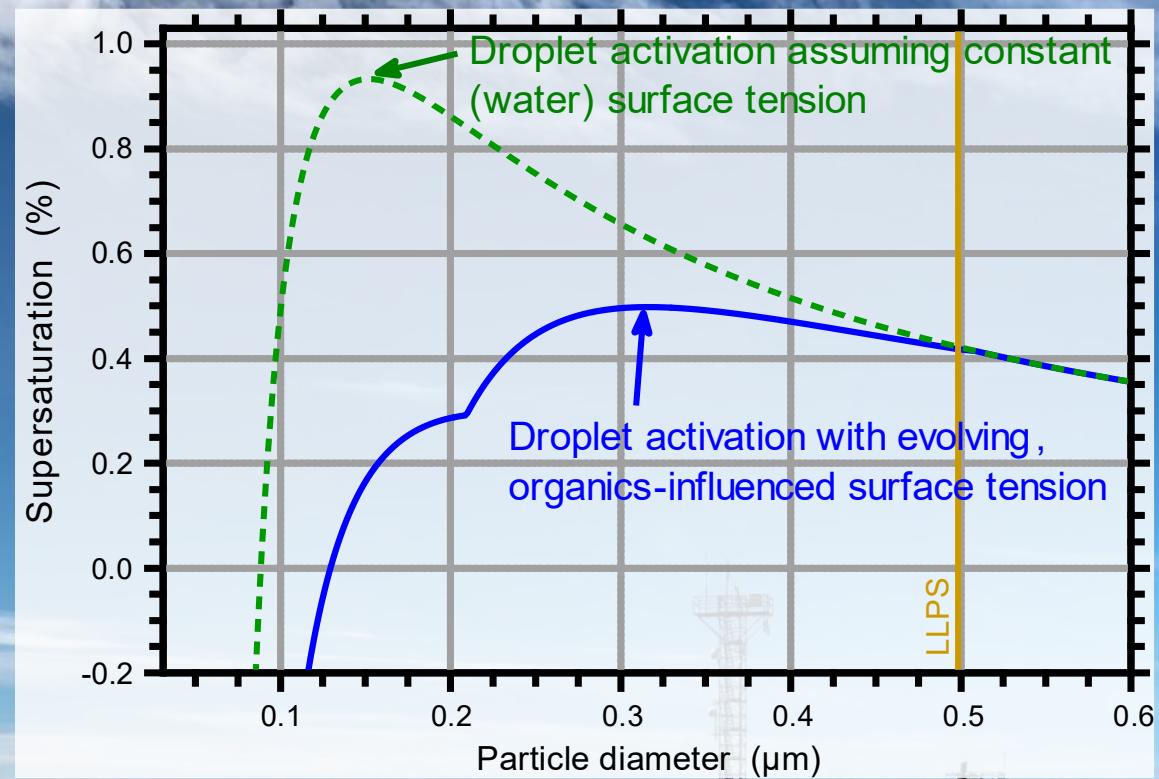
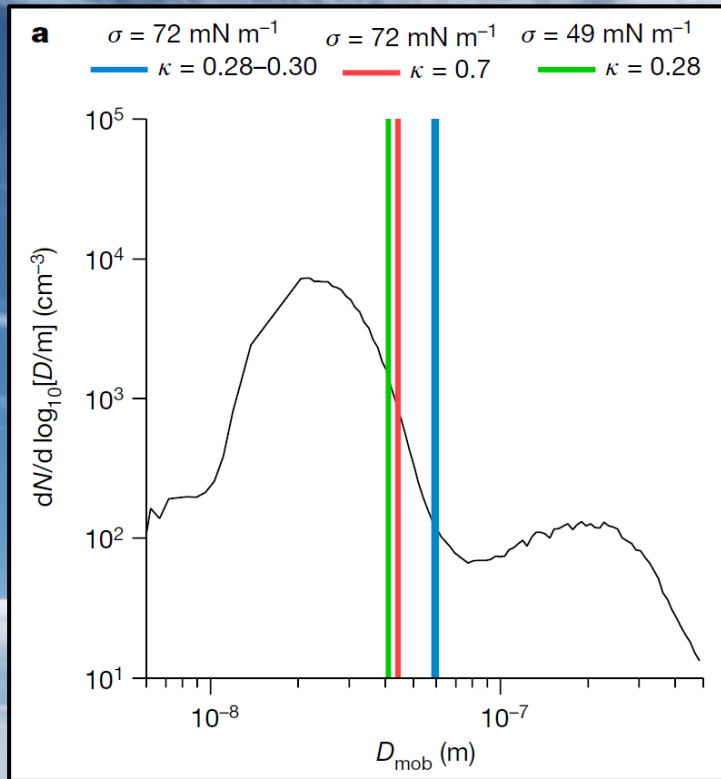
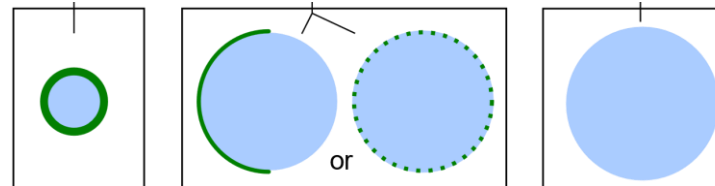


Photo: Mace Head Atmospheric Research Station, C-CAPS NUI Galway, Ireland

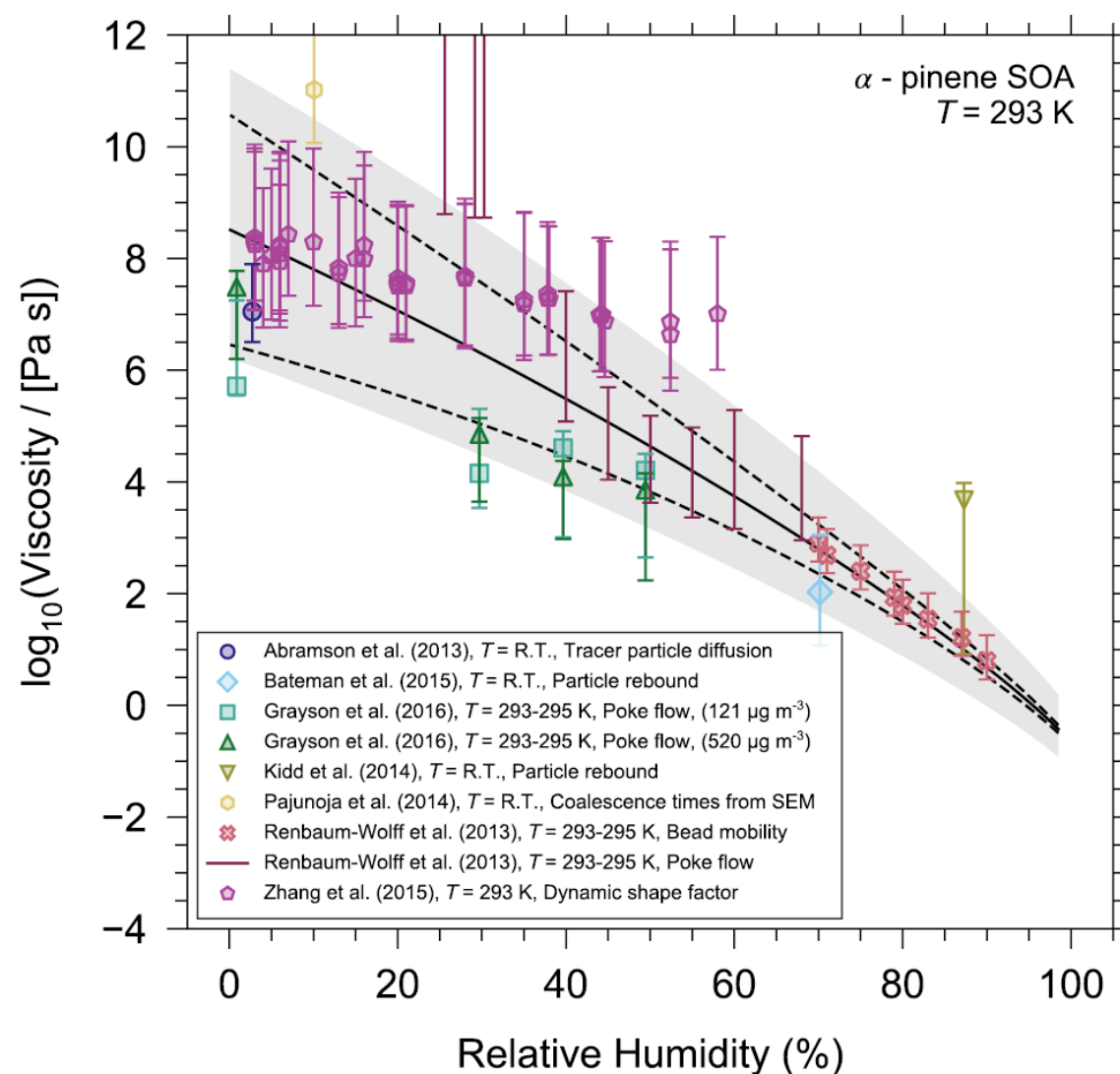
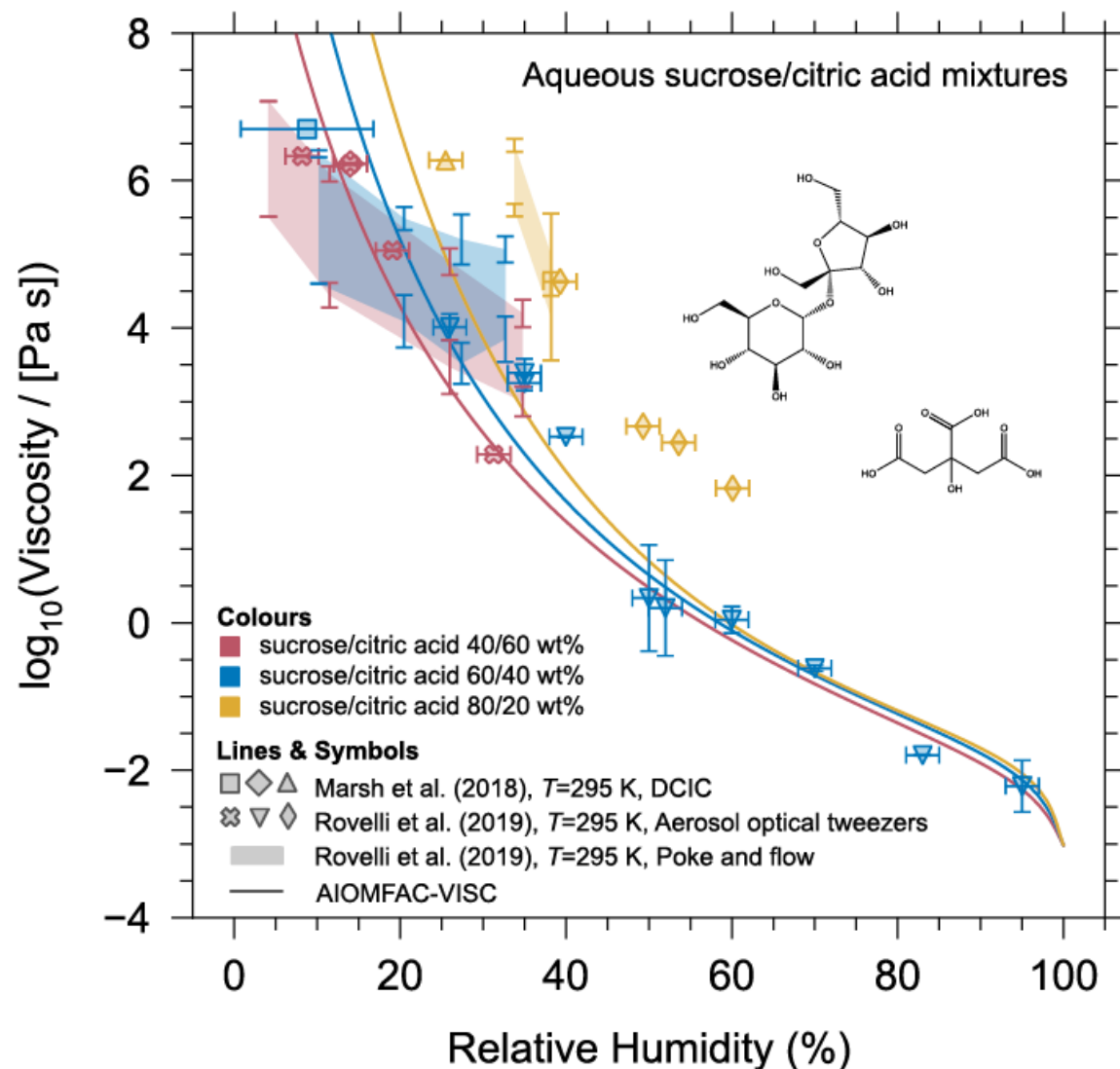
Proposed explanation:

evolution of surface tension due to organic-rich shell phase (surface tension lowering), depending on surface coverage;



[Ovadnevaite et al., 2017, *Nature*]

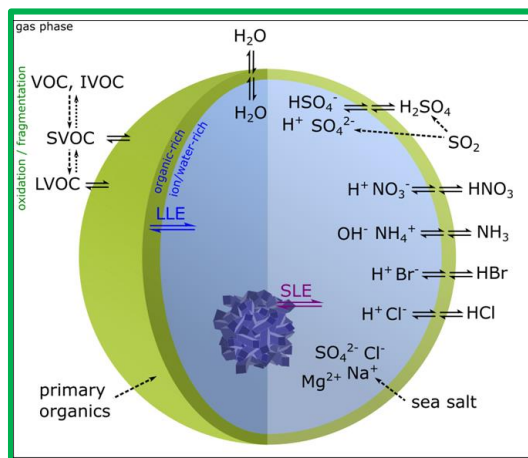
AIOMFAC-VISC: Predictive aqueous organic aerosol viscosity



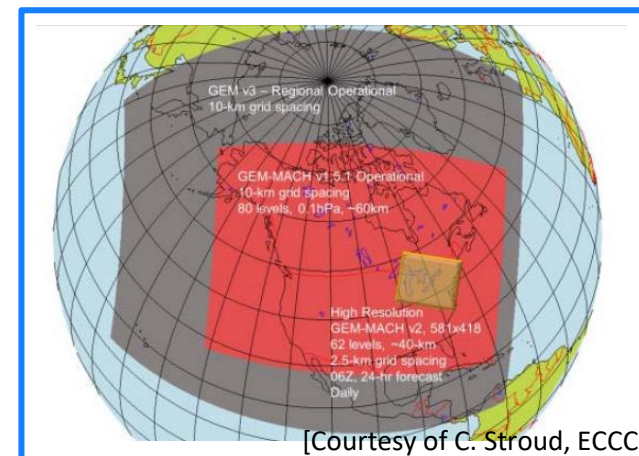
[Gervasi et al., 2020, ACP]

From process level to 3-D models

Process level



3-D atmospheric models



Chemical-structure-based models

- for detailed system characterization
- computationally expensive

Reduced-complexity models

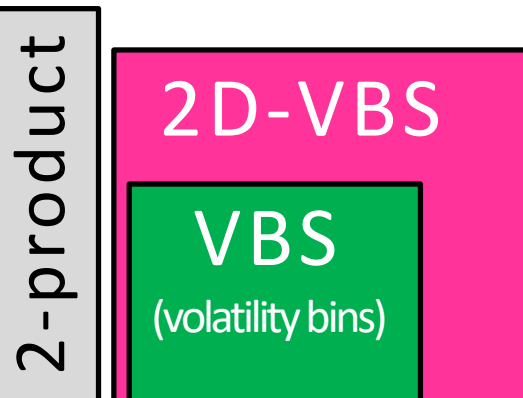
- processing of field measurement data
- operating on limited information



Equilibrium gas-particle partitioning methods

Volatility-based

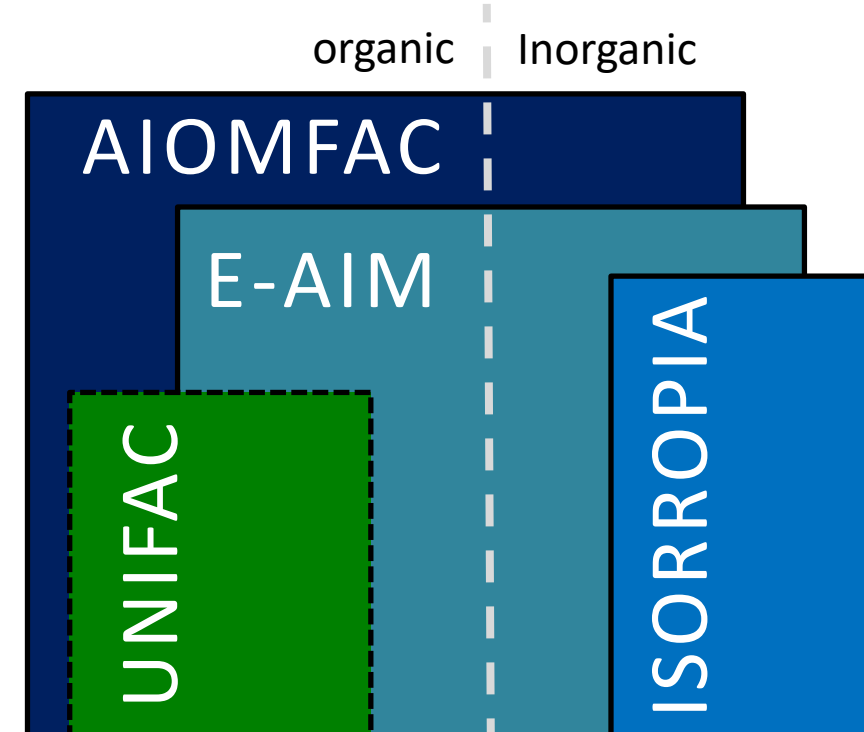
can use commonly
measured (bulk)
properties.



C^* , (K_p) $O:C$

no water content considered
assumes $\gamma \approx 1$ (ideal mixing)

Structure-based models



molecular structure inputs,
water considered

non-ideal liquid mixing

Building an intermediate, reduced-complexity model

Volatility-based

can use commonly measured (bulk) properties.

2-product

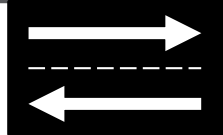
2D-VBS

VBS
(volatility bins)

C^* , (K_p) $O:C$

no water content considered
assumes $\gamma \approx 1$ (ideal mixing)

C^0 , C^* & $O:C$



BAT

γ variable

water-sensitive

fitting data

ONE WAY

Structure-based models

organic | Inorganic

AIOMFAC

E-AIM

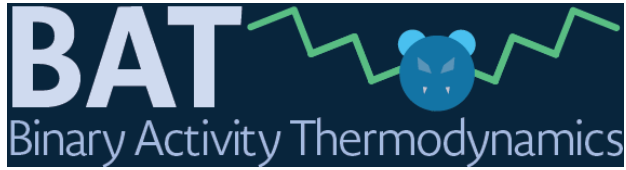
UNIFAC

ISORROPIA

molecular structure inputs,
water considered

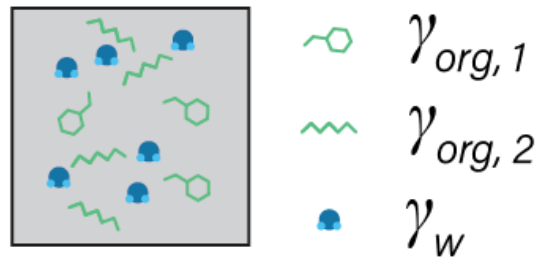
non-ideal liquid mixing

Binary Activity Thermodynamics (BAT) model



- **Organic $\leftrightarrow$ water interactions**
- Generalized Duhem—Margules activity coefficient model

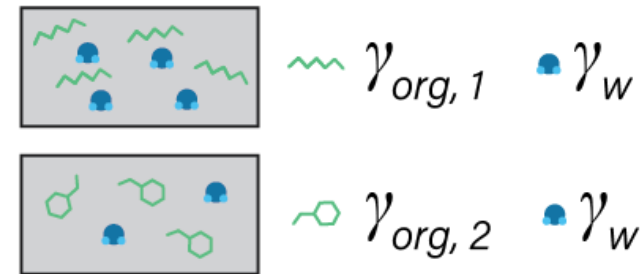
(a) Multi-component mixture



Non-ideal interactions
among all species

(b)

BAT approximation



Non-ideal interactions
with water

BAT model – a general parameterization

Generalized Duhem—Margules binary activity coefficient model of BAT:

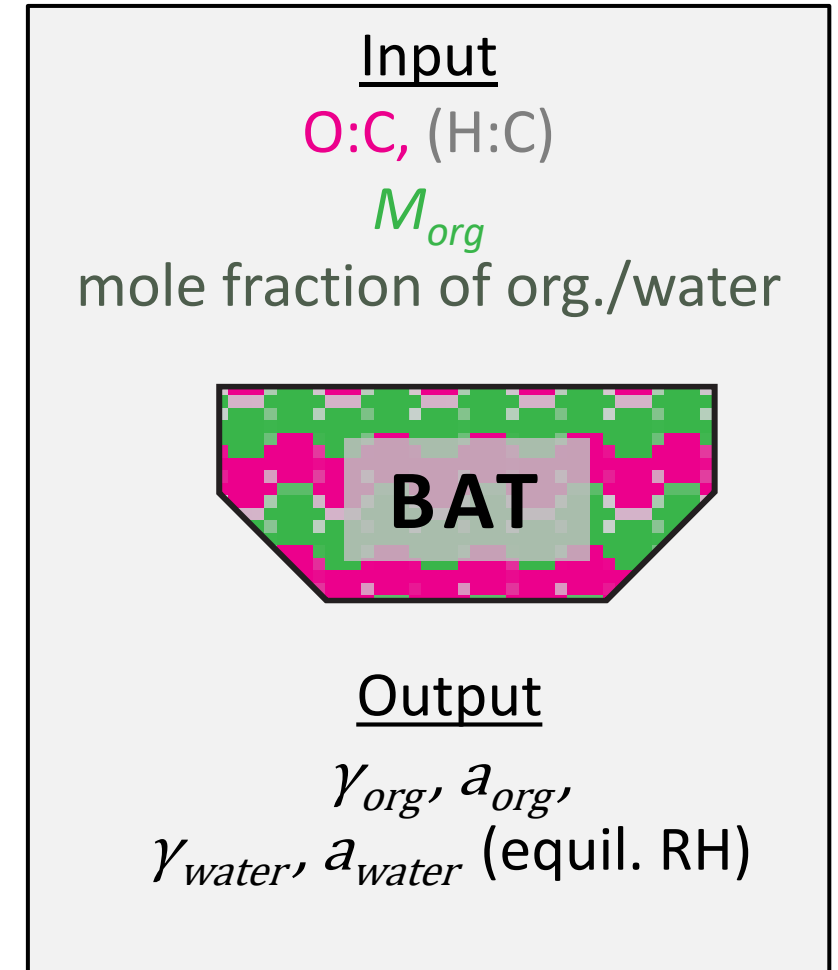
$$g^e/RT = C_1 \varphi_{org}^* (1 - \varphi_{org}^*) + C_2 \varphi_{org}^* (1 - \varphi_{org}^*) (1 - 2\varphi_{org}^*)$$

φ_{org}^* = scaled volume fraction of organic

$$C_1 = a_1 \exp(a_2 \cdot \text{O:C}) + a_3 \exp\left(a_4 \frac{M_{water}}{M_{org}}\right)$$

$$C_2 = a_5 \exp(a_6 \cdot \text{O:C}) + a_7 \exp\left(a_8 \frac{M_{water}}{M_{org}}\right)$$

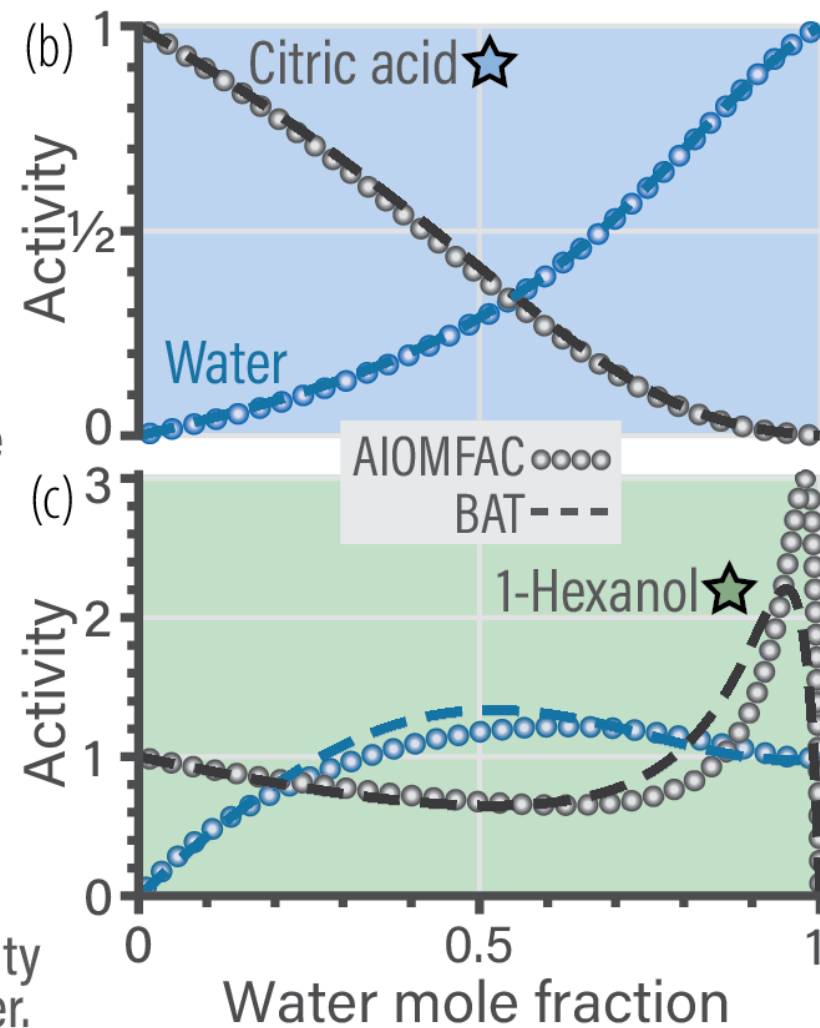
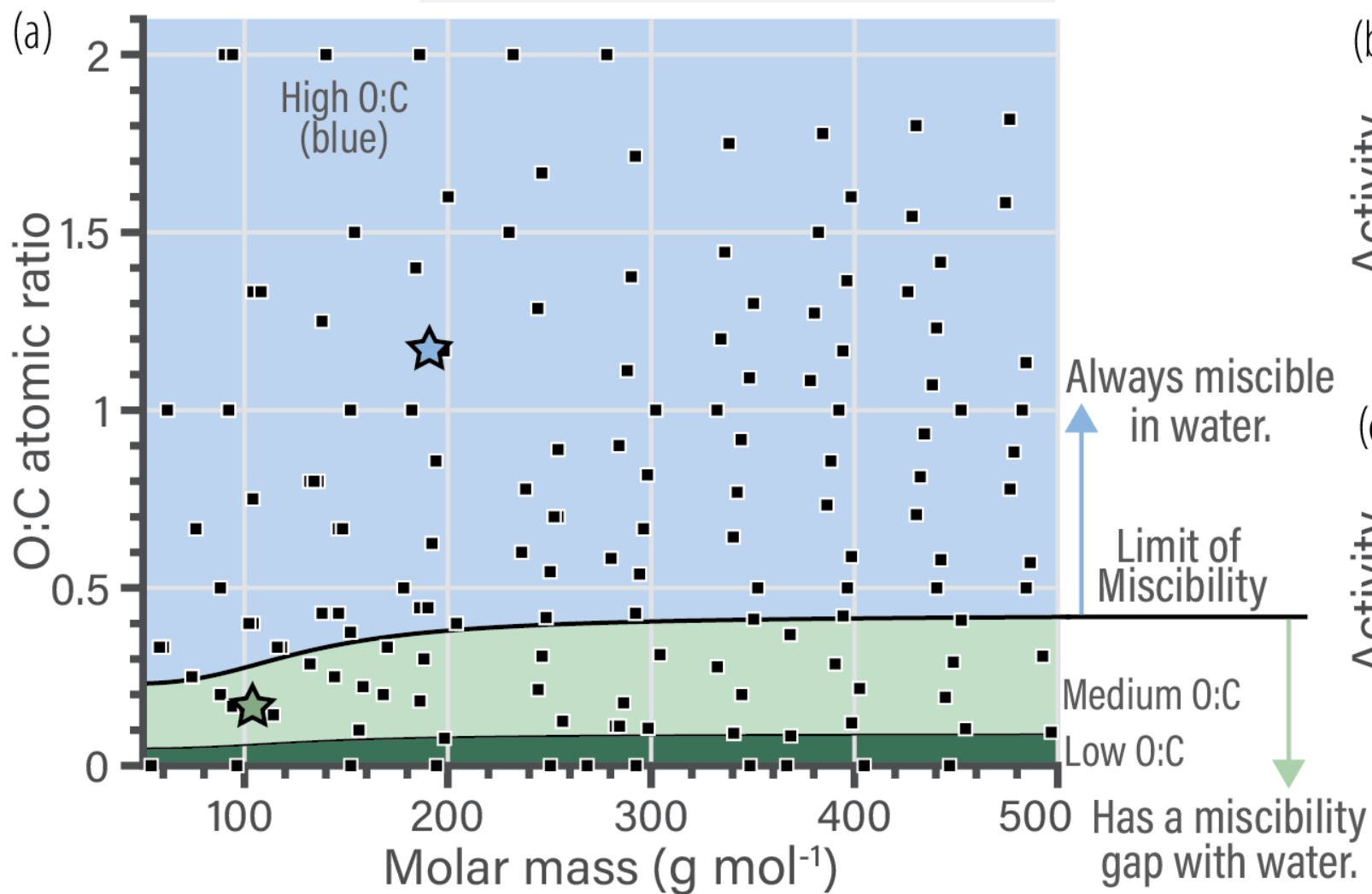
→ Fit parameters $a_1, a_2, \dots$ for a wide class of organics



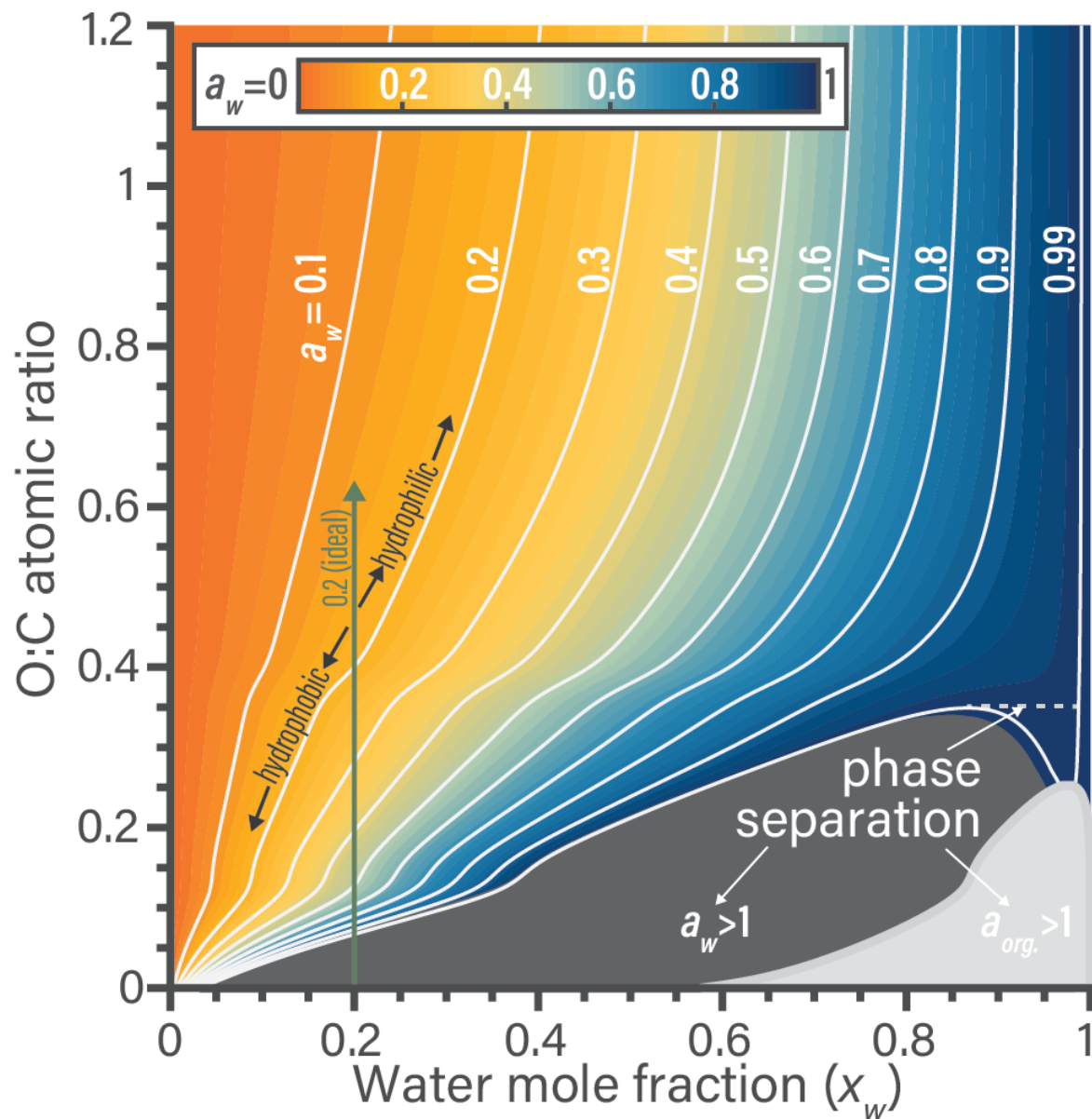
Training a **BAT** – 3 distinct domains

Binary Activity Thermodynamics

Training dataset of 160 molecules



BAT predictions for non-ideal mixing behavior



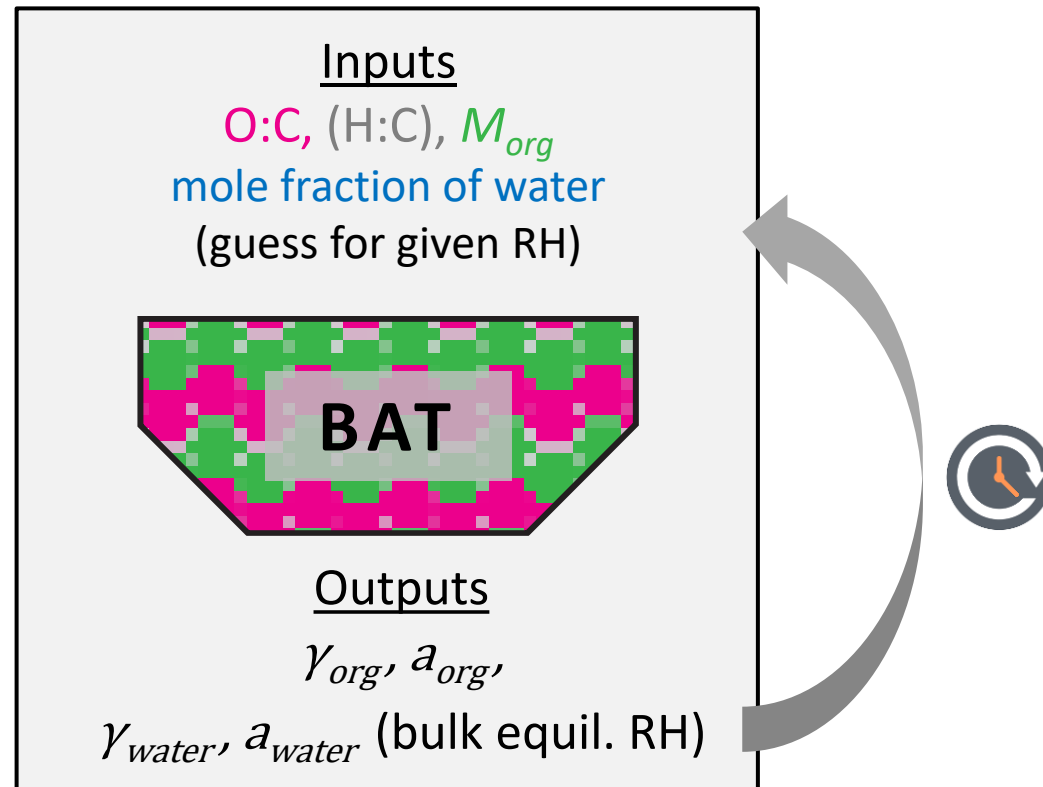
BAT  **prediction of water activity**
Binary Activity Thermodynamics

- here $M_{org} = 200 \text{ g/mol} \rightarrow$ scan O:C, x_w
- $a_w = x_w \times \gamma_w$ (= bulk equil. RH)
- BAT reveals range of phase separation via a simplified LLPS estimation method (for high efficiency)

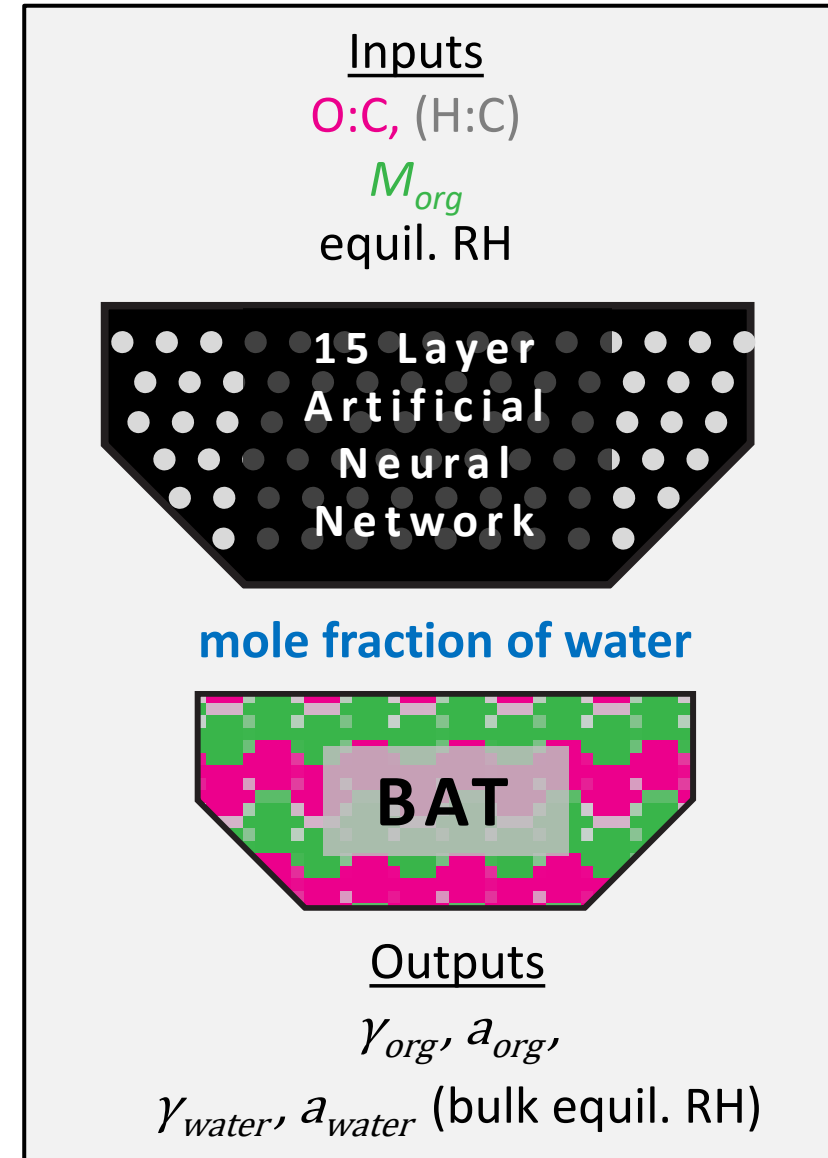
Organic water uptake: iteration vs. neural network

In SOA experiments and chemical transport models the **RH**, **O:C**, (H:C) and M_{org} are typically provided.

Iteration: find aerosol water content **for given RH**



iterative → time consuming



with trained NN, non-iterative → substantially faster

VBS + BAT: including water, γ_j , and phase separation

VBS + BAT
Binary Activity Thermodynamics

sums include water

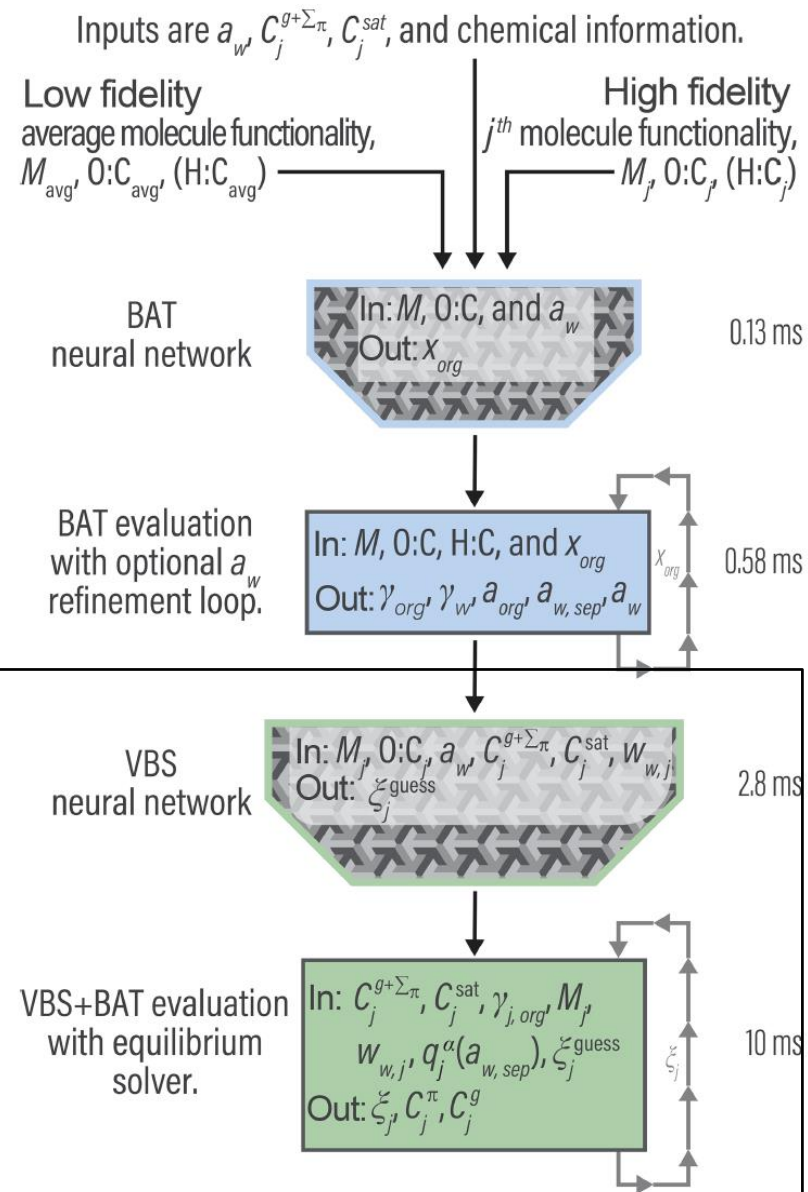
$$C_j^* = C_j^{sat} \gamma_j^\alpha q_j^\alpha \frac{\sum_k \frac{C_k^\alpha}{q_k^\alpha}}{M_j \sum_k \frac{C_k^\alpha}{M_k}}$$

activity coeff.
(in phase α)

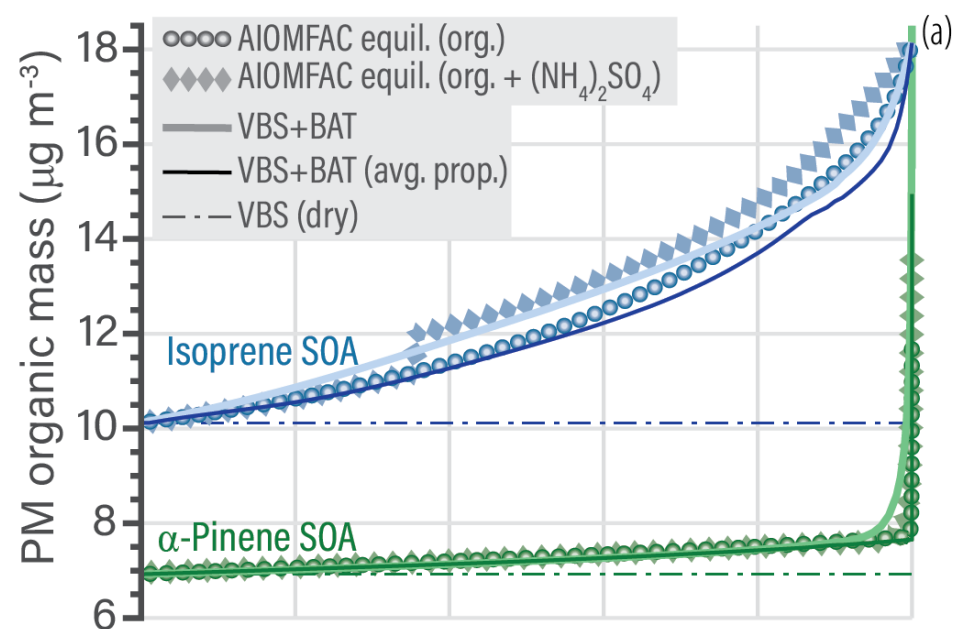
Liquid-liquid phase
partitioning (0 to 1)

Organic + water PM mass:

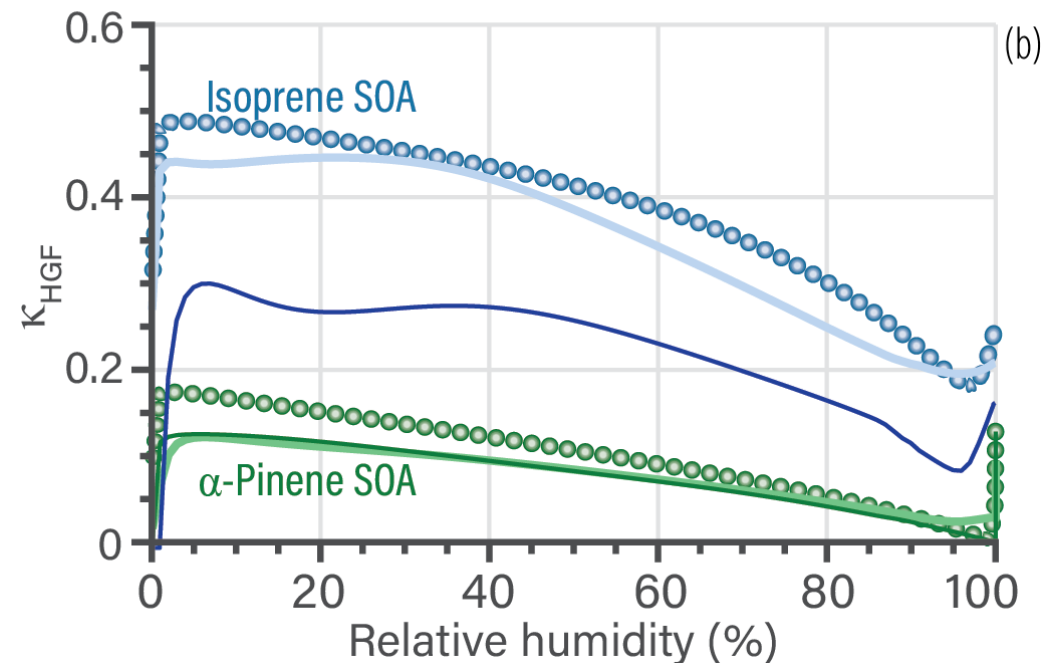
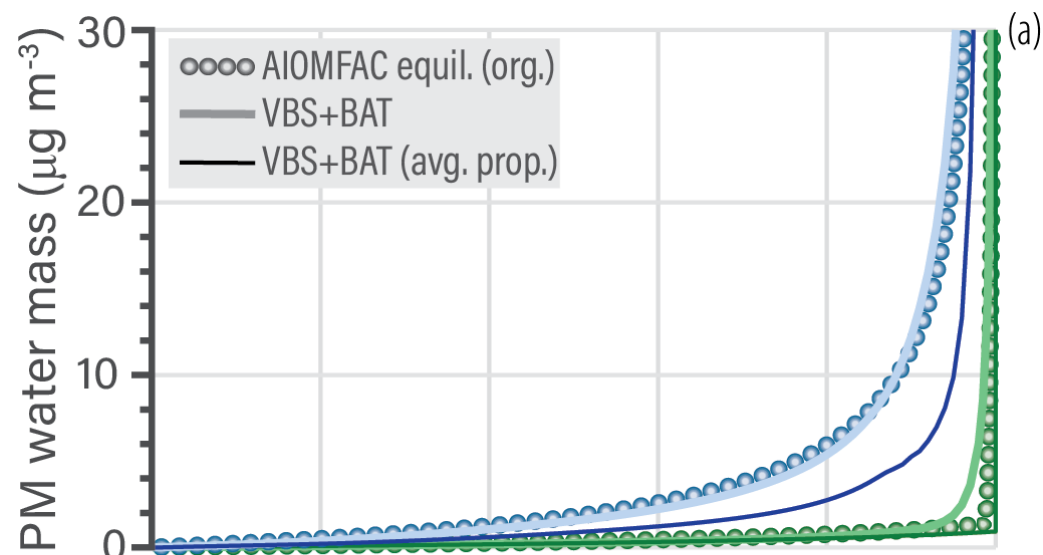
$$C_{OA,aq} = \sum_k C_k^\alpha + C_k^\beta ; \quad \xi_j = \frac{1}{1 + \frac{C_j^*}{C_{OA,aq}}}$$



VBS + BAT vs. AIOMFAC predictions: RH-dependent aerosol mass



Notes: organic PM mass concentration increases with RH due to hygroscopic growth and associated feedback on gas—particle partitioning of SVOCs; α -pinene SOA mass conc. increase is rel. abrupt at high RH due to step-like switch from low to high water content; Both AIOMFAC and VBS+BAT capture this behavior...

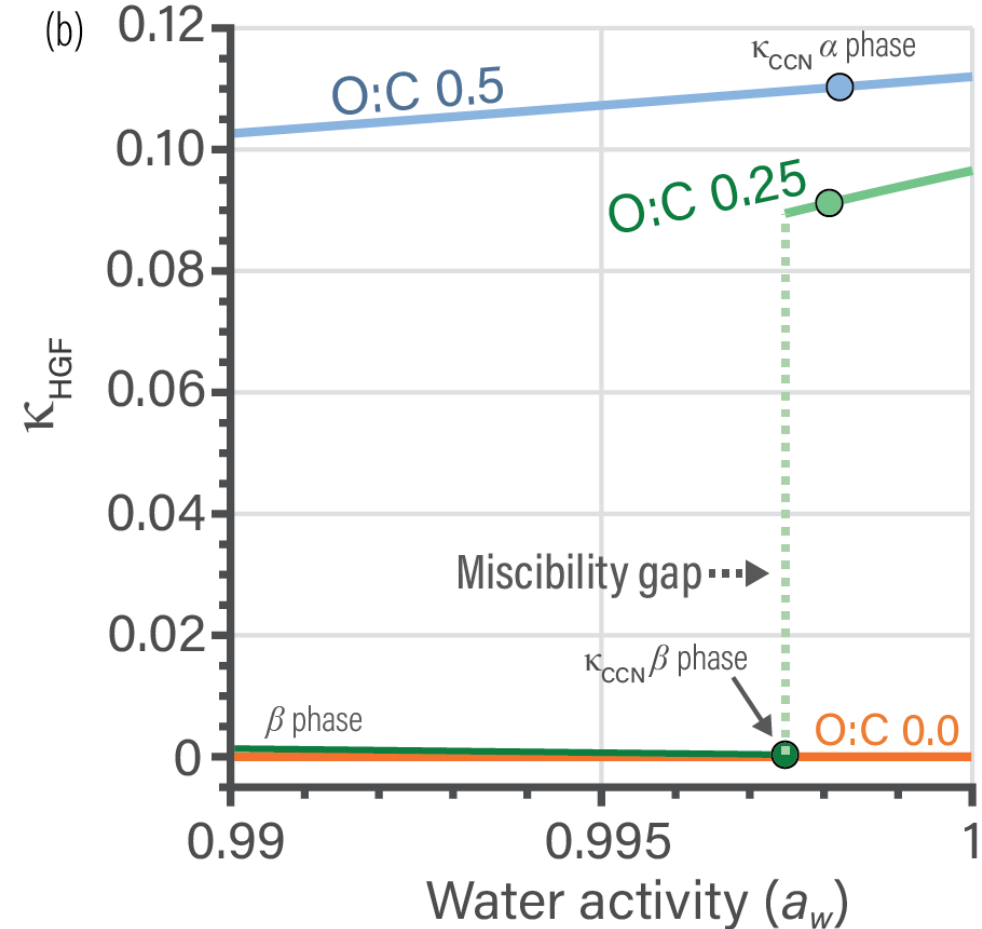
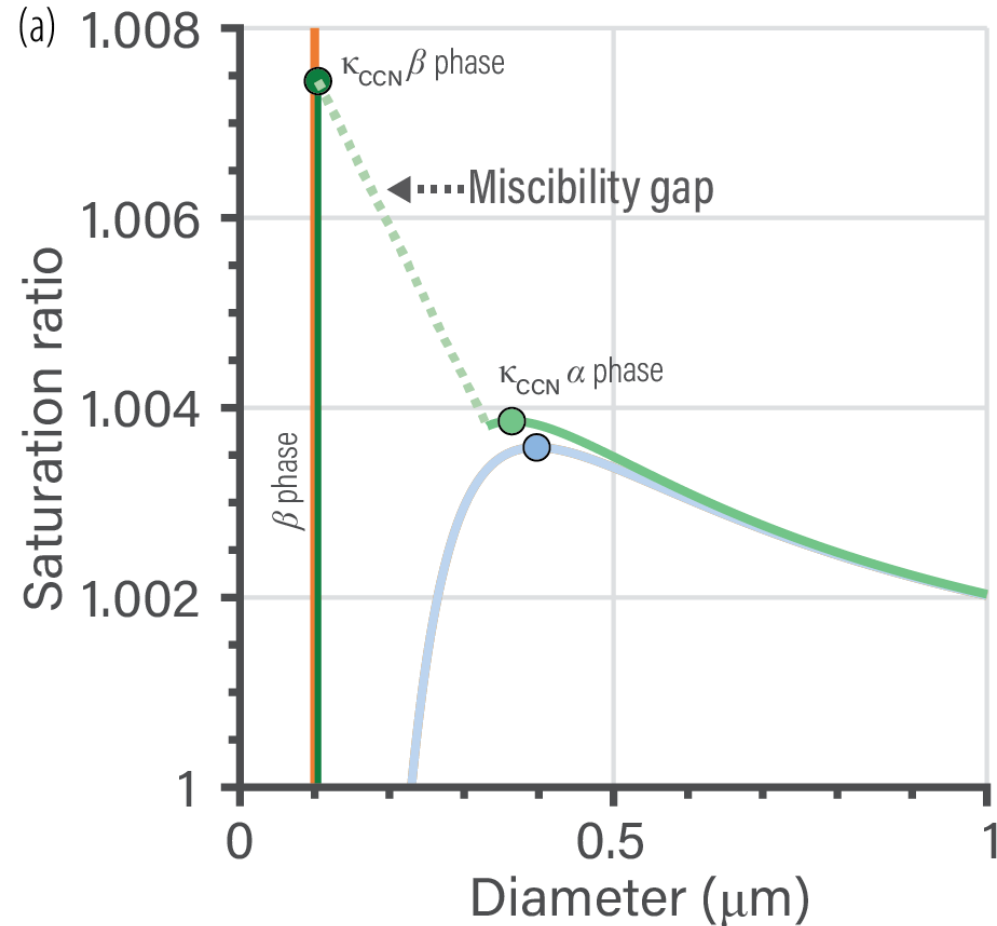


[Gorkowski et al., 2019, ACP]

BAT predictions: O:C-dependent CCN activation

Particle of 100 nm dry diameter and $M_{\text{org}} = 200 \text{ g/mol}$

[Gorkowski et al., 2019, ACP]

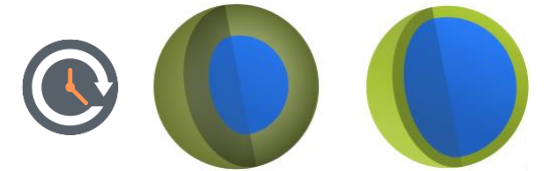
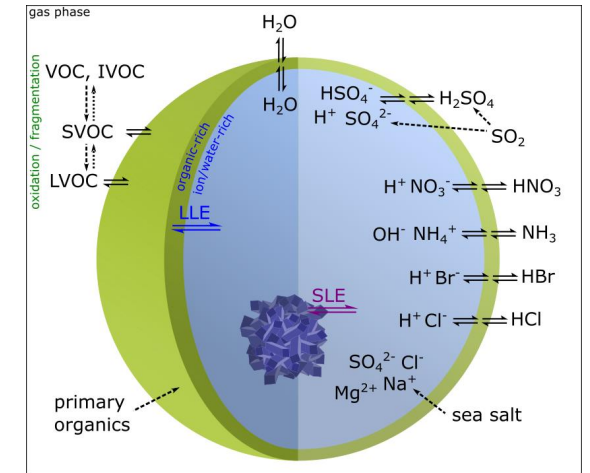


Notes: BAT model allows for predictions of Köhler curves (left) and associated effective kappa values (right) in the context of scanning the O:C / M_{org} parameter space;
→ High-RH behavior and CCN activation point can be predicted for (generic) organic aerosols.

Summary

Prediction of water uptake and related effects on aerosol and CCN properties with different tools:

- AIOMFAC & AIOMFAC-based thermodynamic equilibrium framework
- Organic aerosol viscosity prediction (AIOMFAC-VISC)
- Binary Activity Thermodynamics (BAT) reduced-complexity model
- VBS + BAT
 - organic PM water content (hygroscopicity)
 - organic co-condensation effects with increasing RH
 - can account for liquid–liquid phase separation.



Acknowledgements

Funding Support

- Environment and Climate Change Canada (ECCC), G&C grant

This project was undertaken with the financial support of:
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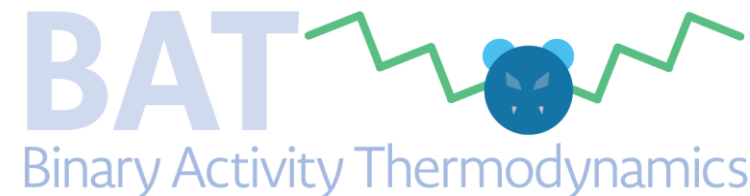
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- Natural Sciences and Engineering Research Council of Canada (NSERC)



References



- **Gorkowski, Preston & Zuend (2019):** Relative-humidity-dependent organic aerosol thermodynamics via an efficient reduced-complexity model, Atmos. Chem. Phys., 19, 13383–13407, <https://doi.org/10.5194/acp-19-13383-2019>, 2019.

AIOMFAC-VISC

- **Gervasi, Topping & Zuend (2020):** A predictive group-contribution model for the viscosity of aqueous organic aerosol, Atmos. Chem. Phys., 20, 2987–3008, <https://doi.org/10.5194/acp-20-2987-2020>, 2020.
- <https://aiomfac.lab.mcgill.ca>

Contact

- Andi Zuend: andreas.zuend@mcgill.ca