

Compound Specific Stable Sulfur Isotope Analysis (δ^{34} S and δ^{33} S) of Organic Compounds Using Gas Chromatography Hyphenated with Multiple Collector Inductively Coupled Plasma Mass Spectrometry (GC-MC-ICPMS)

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Compound Specific Stable Sulfur Isotope Analysis

Isobaric interferences and background

- Isobaric interferences (e.g. ¹⁶O¹⁶O⁺, ³²SH⁺) can falsify signals of ³²S and ³³S
- Normally avoided by separation of sulfur masses from interferences by operating in medium or high resolution; not possible in low resolution \rightarrow thus, investigations concerning isobaric interferences carried out in medium resolution
- GC-MC-ICPMS \rightarrow dry plasma conditions \rightarrow significant reduction of interferences \rightarrow main factor influencing interference intensity and signal sensitivity of ³²S and ³³S is sample gas flow \rightarrow sample gas flow adjusted to maximal signal to noise ratio (Fig. 1A) and not to maximum sensitivity (Fig. 1B)
- Result: ¹⁶O¹⁶O⁺ only a minor abundance, ³²SH⁺ was insignificant



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Figure 1: Mass scans of ³²S (black), ³³S (green) and ³⁴S (red) at medium resolution during inhalation of SF_6 . A: mass scan at a sample gas flow of 1.31 L/min. At this flow the signal-to-noise ratio was highest and interferences at mass 32 are very B: mass scan at a sample gas flow of 1.4 L/min. The signal is close to maximum and interferences are visible

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Compound Specific Stable Sulfur Isotope Analysis Analytical precision

- Tested from 60 to 1500 pmol S injected on-column → signal sizes of 15 to 350 V for mass 32
- Precision (low and medium mass resolution) ranged from <0.25 mUr (³³S) and <0.1 mUr (³⁴S) for 60 pmol S to <0.1 mUr (³³S) and <0.05 mUr (³⁴S) for 400 pmol of S (ca. 100 V signal)
- Precision for ³⁴S similar to previous reports¹
- Precision for ³³S of CSIA method not yet been reported but achieved precision of <0.1 mUr for ³³S only requires 400 pmol S which may be sufficient for multiple applications



Figure 2: Analytical precision for ³³S and ³⁴S for different amounts of sulfur given in pmol. Each data point indicates the standard deviation of five replicate measurements.

Compound Specific Stable Sulfur Isotope Analysis Linearity

- δ³³S and δ³⁴S showed a linear relationship with a slope of about 0.004 and 0.008, respectively → average change of 0.4 mUr (δ³³S) and 0.8 mUr (δ³⁴S) per 100 V measured for mass 32 (Fig. 3A)
- To avoid linearity effects → usage of internal reference compounds SF₆ (gas phase analyses) and thiophene (dissolved organics)
- Internal reference compound was adjusted to analyte → similar signal sizes of internal reference and analyte (maximum difference ±20%)
- Internal reference compound was set to zero at each peak size and analyte of same peak size was referenced against them (Fig. 3B)



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Figure 3: Relationship between signal size and isotopic composition (δ^{34} S versus signal). A: All isotopic ratios of tetrahydrothiophene (THT; signal sizes of 17 to 170V) were referenced against the isotopic ratio of thiophene (internal reference compound) with a 17 V signal size \rightarrow consequence: linear relationship between signal size and δ^{34} S of THT. δ^{34} S shifts by about 1.2 mUr from signal sizes of 17 to 170V. B: THT is referenced against thiophene of similar signal sizes only \rightarrow consequence: $\delta^{34}S$ become independent of signal size within analytical uncertainty.

Compound Specific Stable Sulfur Isotope Analysis Method evaluation and referencing of compounds versus V-CDT

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- Characterization of organics versus V-CDT done by two point normalization using international reference materials IAEA-S2 and IAEA-S3 (Ag₂S) converted to SF₆; quality of normalization was confirmed by analyzing IAEA-S1 as SF₆ (low and medium resolution were tested)
- Needed, because adequate organic sulfur standards not available; additional problem → no δ³³S consensus values of IAEA standards are yet assigned, but there are δ³³S values available in literature²
- Result: no significant difference between δ³³S and δ³⁴S obtained at low and medium resolution and good quality of the normalization (Tab. 1)

Table 1: δ^{33} S and δ^{34} S values of international reference materials (IAEA) and industrially produced organic compounds.

Compound	δ33S [mUr] normalizied	δ33S [mUr] published	δ34S [mUr] normalizied	δ34S [mUr] consensus
IAEA-S1	-0.07±0.18	-0.05 ²	-0.16±0.10	-0.30 ²
IAEA-S2	11.57±0.12*	11.57 ²	22.62±0.08*	22.70 ³
IAEA-S3	-16.61±0.16*	-16.61 ²	-32.49±0.09*	-32.30 ³
SF ₆ (in-house)	1.16±0.15	-	1.87±0.10	-
Dimethyl disulfide (DMDS)	8.60±0.17	-	16.97±0.07	-
Thiophene (THI)	-1.74±0.18	-	-3.35±0.07	-
Diethyl sulfide (DES)	6.36±0.20	-	12.54±0.17	-
Tetrahydro- thiophene (THT)	4.13±0.17	-	7.82±0.15	-

 δ^{33} S and δ^{34} S include values measured at low and medium resolution which were indistinguishable. Isotopic reference materials used as anchors for two point normalization to V-CDT scale are indicated with an asterisk.

3 Mann, J. et al. Rapid Commun. Mass Spectrom, 2009; 23: 1116–1124