







Night-time vertical profiles of nitrate radical concentrations in urban environment (Paris, France)

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The nitrate radical



Simplified schematic of nocturnal NOy chemical processes, Benton et al., 2010

- NO₃ plays an important role in night-time chemistry
- Non-negligible impact on night-time polluted area and daytime chemistry in low sunlight conditions (*Geyer et al, 2003; Brown et al, 2005; Forkel et al, 2006*)
- NO₃-oxidation of VOCs produces large amount of organic nitrates (*eg. 50% of isoprene nitrates (Horowitz et al, 2007)*) and secondary organic aerosols (*Brown and Stutz, 2012*) → impact on the air quality and climate



The nitrate radical in urban environment

- NO₃ radical is important in the suburban area far from NOx sources
- Urban atmosphere is less favorable for NO₃ radical occurrence because of the presence of NO₃ sinks (specially NO)

 $NO_3 + NO \rightarrow 2 NO_2$ → Local consumption of NO_3 $NO + O_3 \rightarrow NO_2 + O_2$ $NO_2 + O_3 \rightarrow NO_3 + O_2$ → Slow formation of NO_3

 N_2O_5 losses on surfaces





But: What about vertically ?

- Atmospheric nitrogen chemistry near the earth surface is strongly linked to the dynamics of the boundary layer
- In summer, in altitude:
 - Levels of O₃ increase
 - Rapid conversion of NO in NO₂ (NO rapidly depleted by ozone)
 - NO₃ and N₂O₅ formation

Large variability of NO₃ mixing ratios as a function of height expected and NO₃ peak concentrations may exists (Brown et al., 2005)



What about verticaly ? Using « Ballon de Paris » to vertically explore the night-time urban atmosphere

Balloon de Paris Generali - high payload touristic tethered balloon located within the André Citroën park, on the banks of the Seine, in inner Paris.







What about verticaly ? Using « Ballon de Paris » to vertically explore the night-time urban atmosphere

Possibility to go up to 300 m high above Paris, far away from surfaces and NOx emissions \rightarrow Far away from NO₃ sinks

Objectives

✓ NO₃ vertical profils in Paris urban area
✓ Role of NO₃ radical chemistry above Paris





What about vertically ?

Using "Ballon de Paris" to vertically explore the night-time urban atmosphere

- Measurements during nights with clear whether conditions, no wind, high O₃ concentrations and low Relative Humidity (RH) : July 6, 7 and 13, 2018
- Vertical profiles up to a height of 150 m, and up to 300 m in very calm wind conditions:
 - 6 profiles up to 300 m
 - 6 profiles up to 150 m

Speed: 6 meters.min⁻¹ to 13 meters.min⁻¹

 Scientific equipment was installed every evening after the last tourist flight → time of installation ~1 hour



What about vertically ?

Using Ballon de Paris to vertically explore the night-time urban atmosphere

- The instruments deployed on board:
- ✓ IBBCEAS (Incoherent Broad Band Cavity Enhanced Absorption Spectroscopy) for the concentration measurement of the NO₃ radical (LISA)
- ✓ Mo Environnement SA analyzer for NOx measurement (LISA)
- \checkmark UV Environment SA absorption analyzer for O₃ measurement (LISA)
- ✓ CAPS Environnement SA analyzer for NO₂ measurement (ICARE)
- ✓ OPC GrimmTM for the measurement of particle number concentrations (LISA)
- $\checkmark\,$ RH probe and temperature

• Additional measurements at the ground:

O_{3,} NOx, Particle counter Meteo parameter (RH, P, T, Wind speed and direction), boundary layer height







LIDAR (boundary layer height)

MILEAGE station

The IBBCEAS instrument for NO₃ radical concentration measurements

Incoherent Broad Band Cavity Enhanced Absorption Spectroscopy technique

→ Broad band emission lamp (LED) centered on the strong absorption band of NO₃ at 662 nm and allowing detection of several gas absorbing in the emission region of the LED (NO₃, NO₂, H₂O)

→ Enhance of the optical path of the light in the cavity up to 4,5 km by 2 high reflective mirrors ($R(\lambda) = 99.98\%$)

 \rightarrow The absorption coefficient of the sample is determine as follows:

$$\alpha(\lambda) = [X]\sigma(\lambda) = \frac{1}{L}(\frac{I_0}{I} - 1)(1 - R)$$

 \rightarrow Quantification of absorbing species:

 $\alpha(\lambda) = [NO_3]\sigma_{NO3}(\lambda) + [NO_2]\sigma_{NO2}(\lambda) + [H_2O]\sigma_{H2O}(\lambda) + p(\lambda)$

Performances:

- Detection limit: up to 3 4 ppt/10 seconds
- High optical stability → easy and quick to install



Adapted from Ventrillard-Courtillot et al., 2010

 $\alpha(\lambda)$ = absorption coefficient of the sample in the cavity (cm⁻¹) L = effective length between the two mirrors of the cavity I, I₀ = the intensity of the light in the presence and absence of absorbing sample respectively

 $R(\lambda)$ = mirrors reflectivity

 $\sigma(\lambda)$ = cross section of the absorbing gaz n(cm².molecule⁻¹)

[X] = concentration (molecule. cm⁻³)

 $p(\lambda)$ = cubic polynomial to correct baseline deformations due to small LED intensity variations **EGU** General 2020

The IBBCEAS instrument for NO₃ radical concentration measurements – Balloon deployment

Characteristics:

- Calibration with an accurate known concentration of NO₂ (AirLiquide cylinder, 600 ppb) for reflectivity determination
- Protection of the mirrors to avoid reflectivity degradation:
 - N₂ flush on the mirror surfaces (50 sccm)
 - PTFE filter for particles on the sample line
- PFA coated cavity and line to minimize NO₃ wall losses

Balloon deployment

- ightarrow Installed every evening after the last tourist flight
- ightarrow Calibration at ground level every evening before measure
- \rightarrow I₀(λ) recorded at ground level and every 20 to 30 minutes during flights
- \rightarrow I(λ) recorded every minute





The IBBCEAS instrument for NO₃ radical concentration measurements



Technical characteristics:

Dimension = 1350 x 220 x 220 cm Weight = 30 kg





Vertical profiles of gas species **6/07/2018 22:48 - 23:37** Preliminary results

In altitude:

 \rightarrow NO₃ radical and N₂O₅ formation

- → Strong correlation between vertical profiles of [NO₃] radical and precursors [O₃] and [NO₂]
- \rightarrow [NO₃] radical maximal concentration at 200 m of altitude

Near ground level :

 \rightarrow Competition between sources and sinks of NO $_{\rm 3}$ (NO and surfaces)

 $NO_3 + NO \rightarrow 2 NO_2$ $N_2O_5 + H_2O \rightarrow HNO_3$ $NO + O_3 \rightarrow NO_2 + O_2$ $NO_2 + O_3 \rightarrow NO_3 + O_2$ → NO_3 radical in ppt range



Overview

- ✓ Successful deployment of an IBBCEAS instrument for NO₃ radical measurement on a tethered balloon possible due to the high optical stability of the instrument;
- ✓ First results show a vertical profile of NO₃ radical concentrations over Paris with a peak concentration at 200 m of altitude
- ✓ IBBCEAS data treatment still in progress:
 - loss corrections
- more precise treatment of the strong absorption by water vapor that spectrally overlaps with NO_3 absorption in the 662 nm region,
 - NO₂ cross section in the 655 665 nm region source of uncertainties

