Monitoring Ambient Nitrate Radical (NO₃) by Open Path **Cavity Enhanced Absorption Spectroscopy (CEAS)**

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Introduction

Nitrate radical (NO₃) is one of the most important reactive nitrogen species and oxidants in the atmosphere. NO_3 contributes to the nighttime oxidation of volatile organic compounds (VOCs) and produces organic nitrate. The key role of NO₃ chemistry in regulating the NOx and VOC budget generates great interest in measuring NO₃ in the atmosphere. The closed-type CEAS suffers from the change of the transmission efficiency in field measurement, while the open path (OP) CEAS provides an alternative idea to avoid this uncertainty in measuring NO₃. Here we developed a new light emitting diode (LED)-based OP-CEAS instrument for measuring NO₃. The detailed setup of our instrument and the data processes will be introduced as well as its first field application in the Pearl River Delta, China.



Reflectivity	0.99985
Mirror distance	0.84 m
Effective Length	0.68 m
Enhanced Length	~ 5 km
Time resolution	30 s
Limit of detection	3 pptv (2σ)
Uncertainty	11-15%

Figure 1. Reference spectrum (RS) selection. (a) Spectra of the pure N₂ and ambient samples during the daytime and nighttime (a high and a low NO₃ case). (b) Numerical calculation of the dependence of the absorption coefficient on the ratio of I to IO (I/IO) at 662 nm. We selected daytime spectrum as RS due to the low absorption coefficient (leads to small residual spectrum) and free of retrieving H_2O absorption.

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25
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Field Application



Figure 5. (a) Time series of the NO_3 production rate and measured NO₃ on the night of October 30, 2017 in a field campaign in PRD, China. (b) Time series of the measured NO_3 concentration by OP-CEAS and calculated NO₃ by the observed temperature, NO₂, N_2O_5 (with 20% uncertainty).

Figure 6. The functional dependence of the ratio of the spectral intensity to the N₂ spectrum on the ambient $PM_{2.5}$ in field measurement at 662 nm. To ensure the sensitivity of the instrument, our OP-CEAS configuration is applicable when the $PM_{2.5}$ below 50 µg m⁻³. The aerosol mass loading of 50 µg m⁻³ is relatively common in the region of China where these experiments were performed, in polluted regions of Europe or in the U.S., which typically have much lower aerosol mass loadings, and OP-CEAS might be a useful alternative to closed path systems.

Reference Spectrum and Reflectivity





Figure 2. Reflectivity calibration. The $R(\lambda)$ is determined from the difference between pure N₂ and He Rayleigh scattering extinction. The cavity is enclosed by a detection cell with an outlet when calibrating the mirror reflectivity.

Figure 3. Spectral fitting. The measured spectral data is processed by using the DOAS Intelligent System (DOASIS) spectral fitting software. Here is an example of the spectral retrieval of low (a) and high (b) NO_3 in field measurements.



Applicable Environment



Figure 4. Limit of detection. Time series (a) and the histogram (b) of 500 ambient spectral fitting results during the daytime measurement showing that the detection limit of OP-CEAS is 3.0 pptv (2σ) in 30 s with an offset of -0.38 pptv Take home message

Acknowledgement. This work was supported by National Natural Science Foundation of China (Grants No. 91544225, 91844301) and the National Key R&D Program of China (2017YFC0209400). The project is funded by the China Postdoctoral Science Foundation (Grants No. 2018M641095, 2019T120023).





Spectral Fitting and LOD





Free of the sampling loss of the NO₃ radical by opening the optical detection path, the OP-CEAS instrument is engineered to provide sensitive and accurate measurements in field application.

The field measurement of NO₃ by OP-CEAS is feasible and applicable in clean or slightly polluted regions, its current configuration is better suited for a region that does not suffer from heavy PM pollution.

For more details of this work please visit Anal. Chem. 2019, 91, 16, 10687-10693, or feel free to contact with H.W or K.L.