



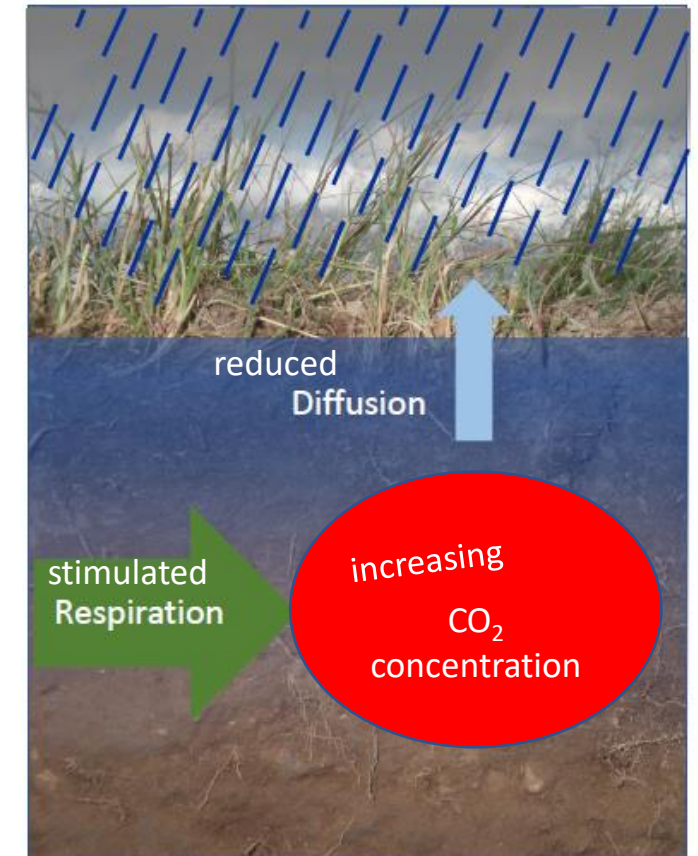
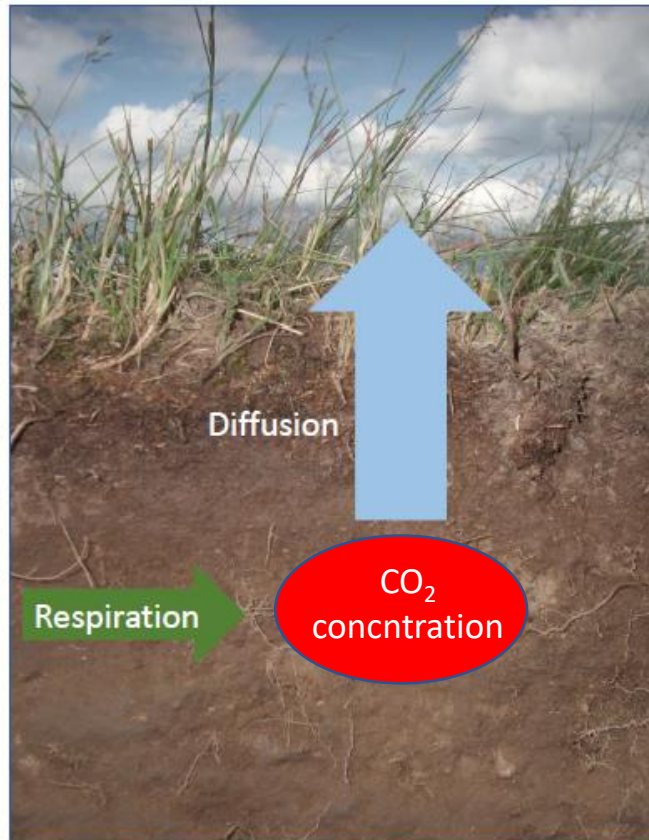
Forstliche Versuchs-
und Forschungsanstalt
Baden-Württemberg

Dynamics and effects of soil CO₂ on carbonate dissolution and transport in response to precipitation events

Martin Maier¹, Laurin Osterholt¹, and Andreas Hartmann²

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²Chair of Hydrological Modeling and Water Resources, Albert-Ludwigs-University of Freiburg, Freiburg, Germany



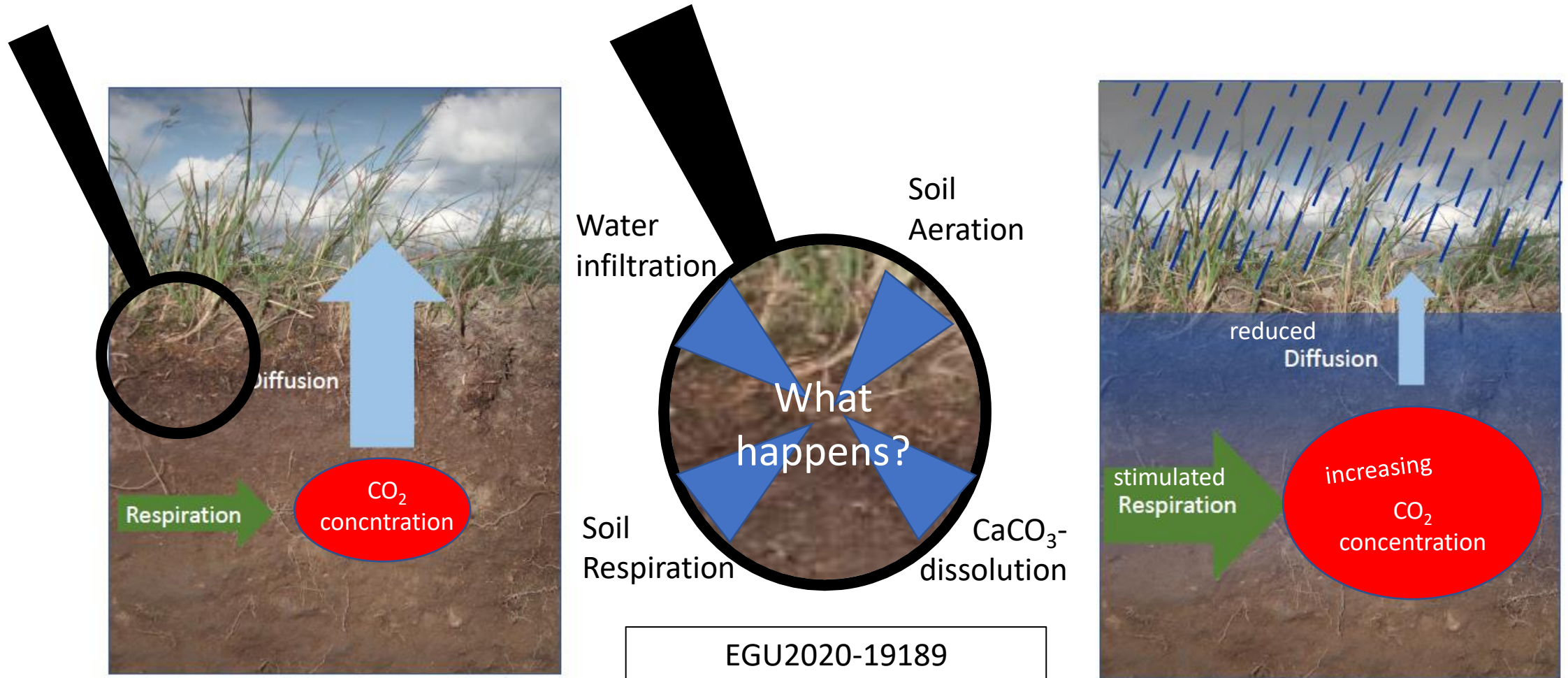
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Dynamics and effects of soil CO₂ on carbonate dissolution and transport in response to precipitation events

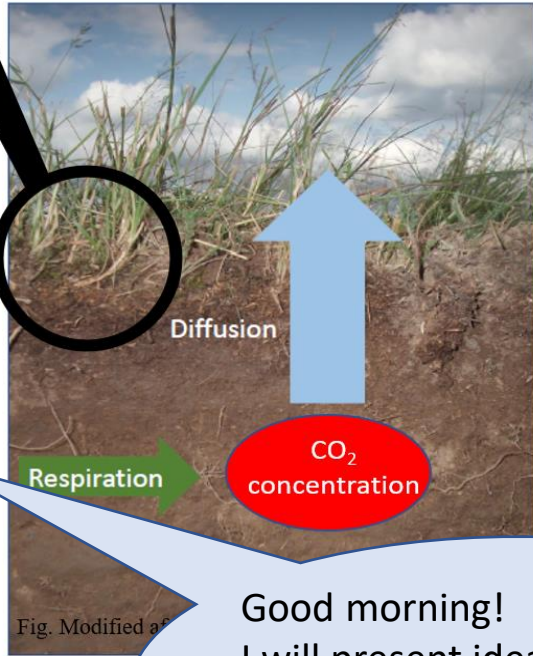
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M.Maier: Soil CO₂ effects on carbonate dissolution



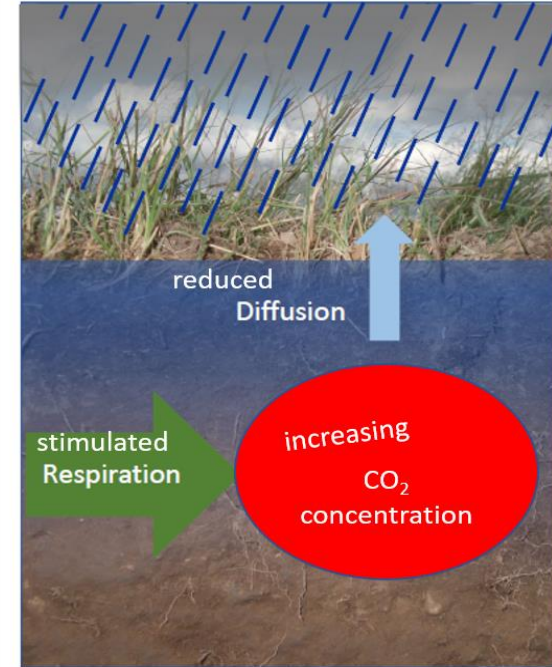
Water infiltration

Soil Aeration



Soil Respiration

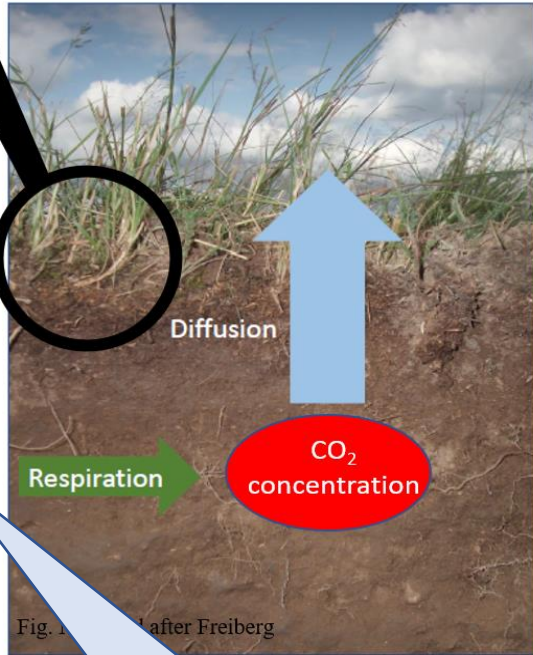
CaCO₃-dissolution



Good morning!
I will present ideas and experimental results about soil CO₂ effects on carbonate dissolution. Let's start!



M.Maier: Soil CO₂ effects on carbonate dissolution



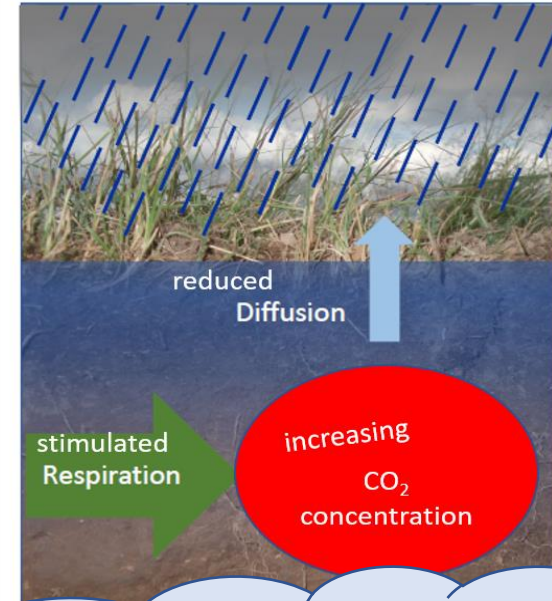
Water infiltration

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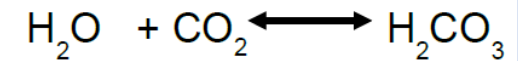
Soil Respiration

CaCO₃-dissolution

What happens?



Part	Reaction equations
Dissolution/precipitation of CaCO ₃	$\text{CaCO}_3 (\text{s}) \leftrightarrow \text{Ca}^{2+} + \text{CO}_3^{2-}$
Dissociation of carbon dioxide	$\text{CO}_2 + \text{H}_2\text{O} \leftrightarrow \text{HCO}_3^- + \text{H}^+$
$\text{CO}_2 + \text{H}_2\text{O} = \text{H}_2\text{CO}_3$	$\text{HCO}_3^- \leftrightarrow \text{CO}_3^{2-} + \text{H}^+$



Dissolution of gaseous CO₂ according to Henry-Dalton law

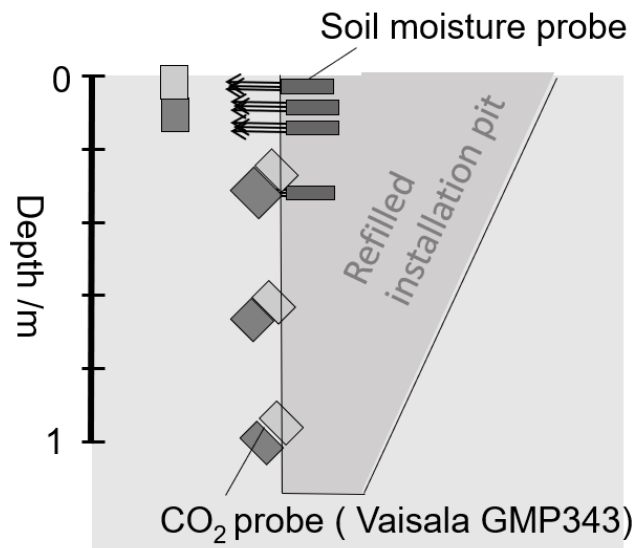
Carbonate dissolution depends on soil pH which depends on soil CO₂. Soil air CO₂ can easily dissolve in water at pH>7 as described by Henry's law.



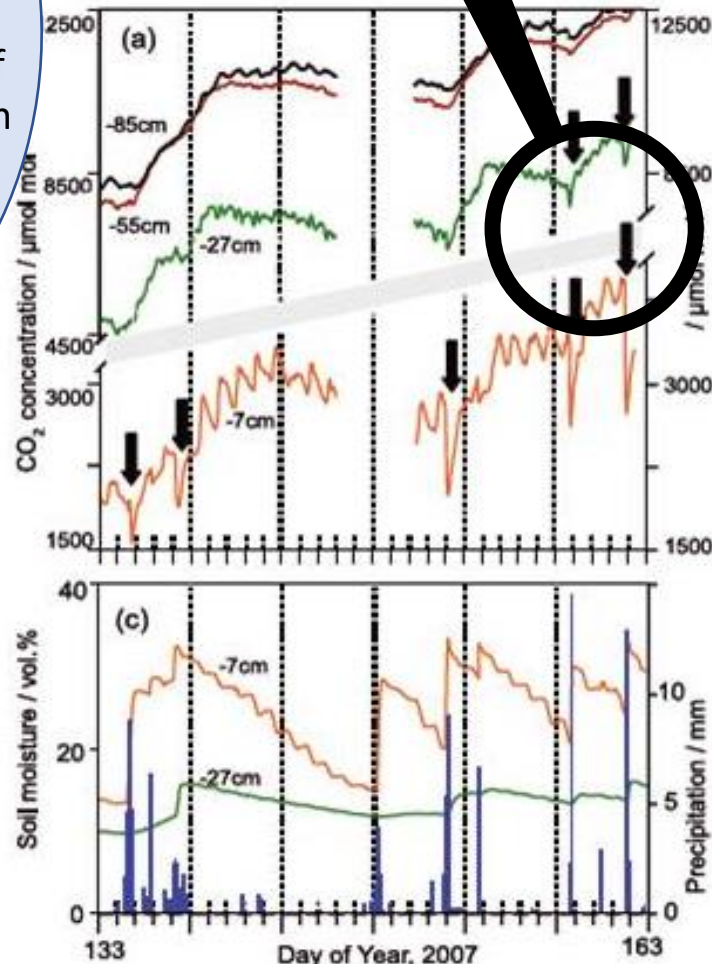


Some time ago we had a field experiment about soil gas fluxes where we installed CO_2 probes in the soil. We measured soil CO_2 concentration at every minute and observed two different types of reaction to rain in the soil air. An immediate effect after intense rain, and a slower reaction after every rain

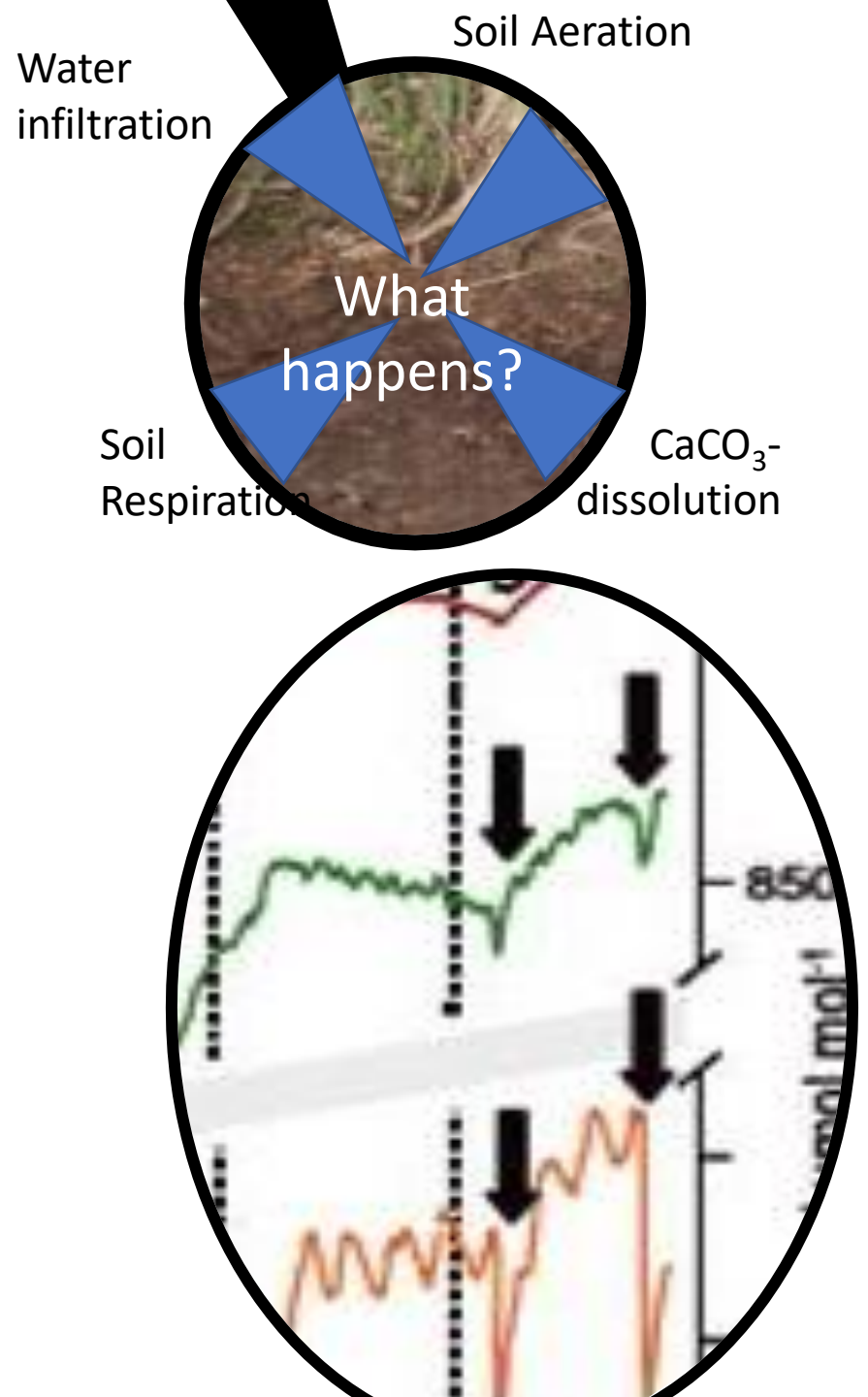
Soil profile measurements



Field observation: behavior of soil CO_2 after rain



(shown examples from Maier et al. 2010, Eur. J. Soil Science 61; more data available)

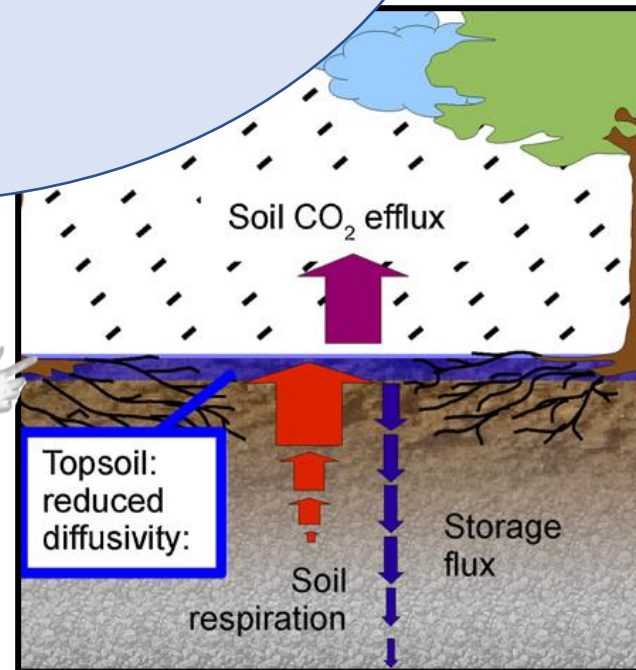


So, what happens after a „normal“ „slow rain event in a calcarous soil?

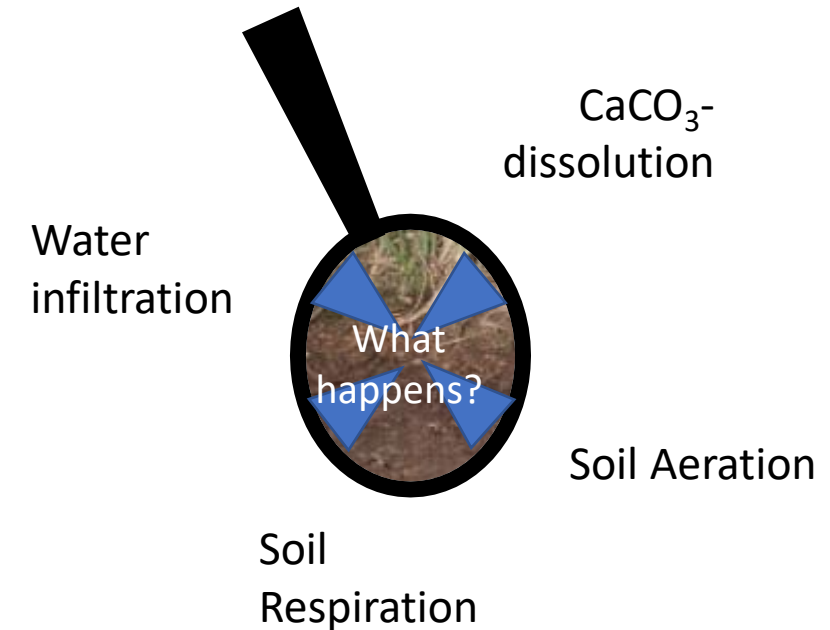
1. The top soil gets wet, formerly areated soil pores get blocked.

„Slow“ rain event

1. Slow infiltration= reduced aeration
2. Increase in soil CO₂
3. Increase in soil respiration & increase in CaCO₃-dissolution



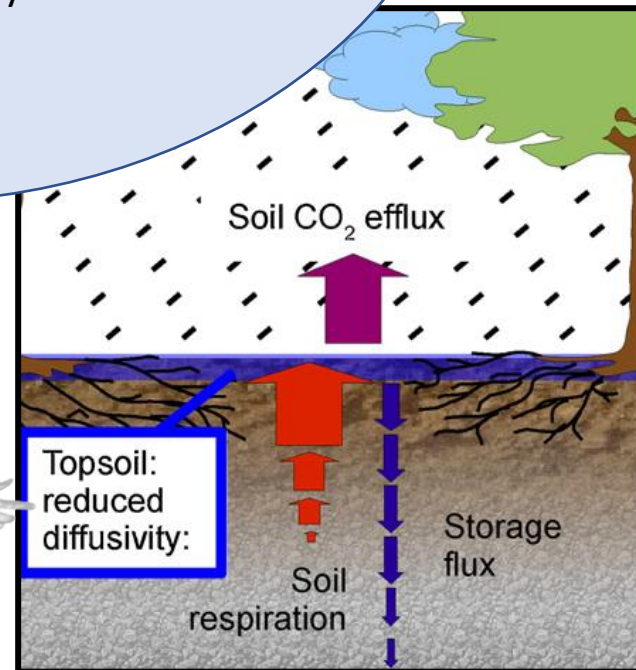
(Maier et al. 2011, Agric. For. Met 151)



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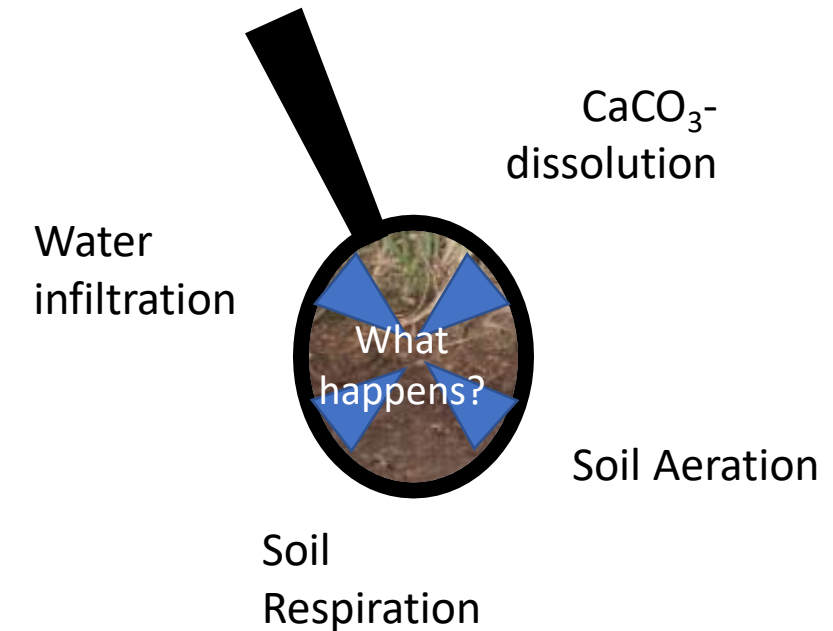
1. The top soil gets wet, formerly areated soil pores get blocked.
2. Soil CO₂ accumulates until the reduced diffusivity and increased soil CO₂ concentration reach a new steady state.
3. Soil air CO₂ equilibrates with CO₂ dissolved in soil water, and this means the pH decreases, yet only slightly since it is buffered by the dissolution of CaCO₃, which increases

„Slow“ rain event



(Maier et al. 2011, Agric. For. Met 151)

1. Slow infiltration= reduced aeration
2. Increase in soil CO₂
3. Increase in soil respiration & increase in CaCO₃-dissolution



And after a „fast“ rain event?

1. The Top soil gets wet quickly, but there is also preferential flow, so that rain water infiltrates quickly into deeper layers.



„Fast“ rain events

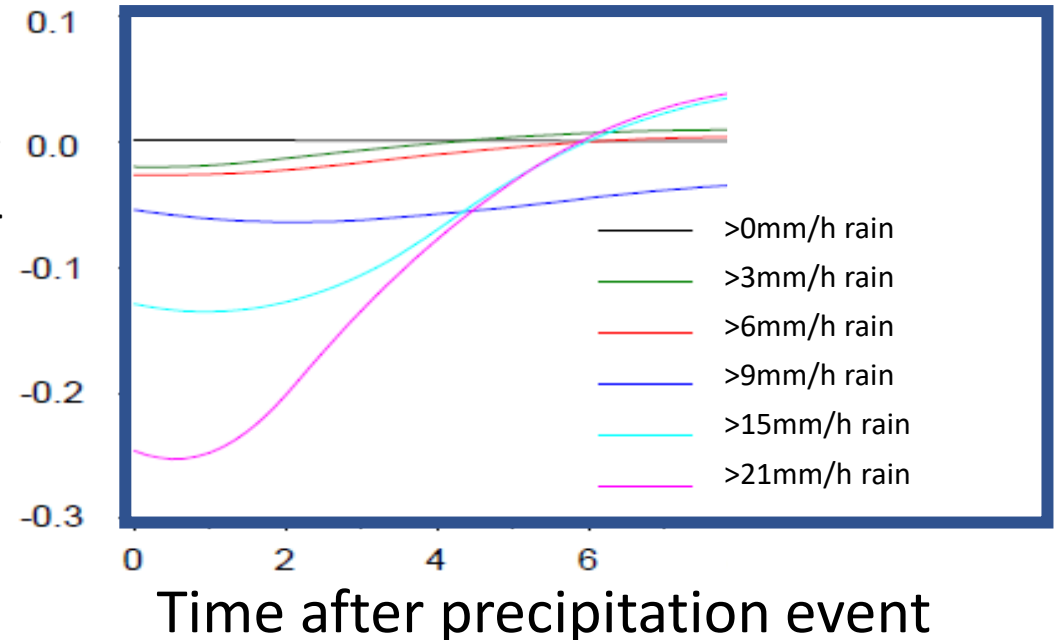
First few hours:

1. Fast infiltration & preferential flow= less effect on aeration
2. Mixing of rainwater low in CO_2 with soil water enriched in CO_2
-> relative undersaturation of new soil water
3. Dips in soil CO_2
4. Increasing CaCO_3 -dissolution

Relative drop in CO_2

concentration

of initial concentration
at 7 cm depth / %



And after a „fast“ rain event?

1. The Top soil gets wet quickly, but there is also preferential flow, so that rain water infiltrates quickly into deeper layers.
2. The percolating rainwater is CO_2 depleted relatively to the soil water and will thus „draw“ CO_2 molecules from the surrounding soil air.
3. Dips in soil CO_2 concentration can be observed due to this effect. The stronger the rain, the more preferentially percolating water, the stronger dips in CO_2
4. And this means that more CaCO_3 is dissolved.



„Fast“ rain events

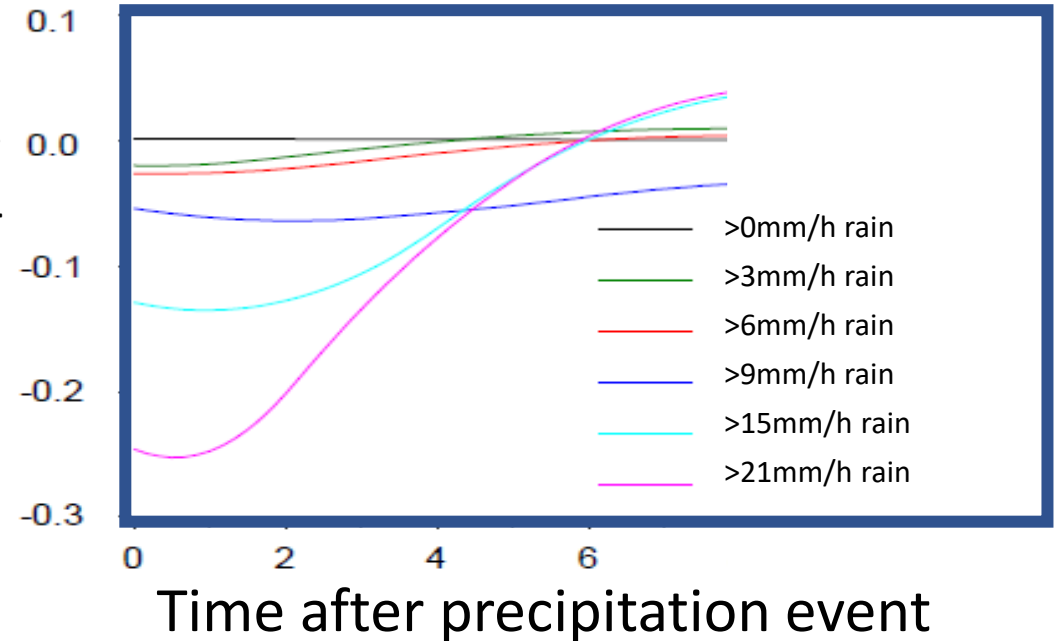
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Relative drop in CO_2

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The total amount of Ca^{2+} mobilized due to CaCO_3 dissolution is not affected by slow or fast rain event, but the localization where exactly the Carbonate is dissolved is affected.



„Fast“ rain events

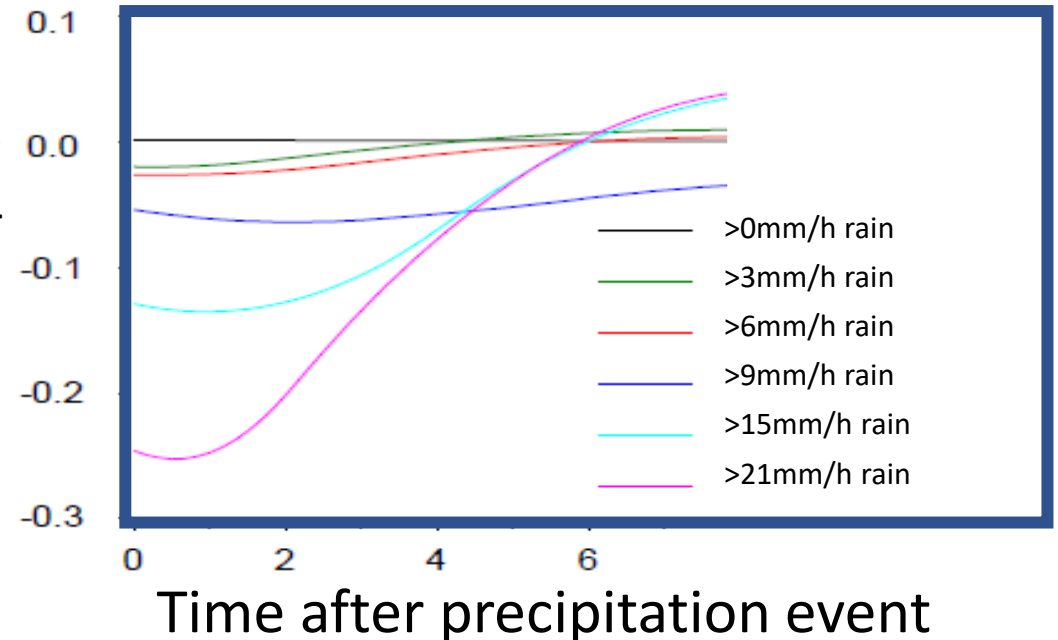
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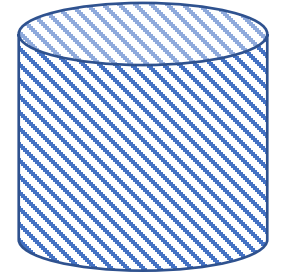
If this interpretation was true,
we should be able to reproduce this effect als in
the lab. So we set up an experiment with soil
mesocosms. One with intact soil structure, and
one repacked mesocosm.



Intact Mescosm



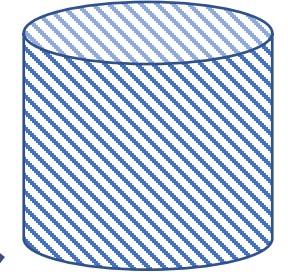
Re-Packed Mescosm



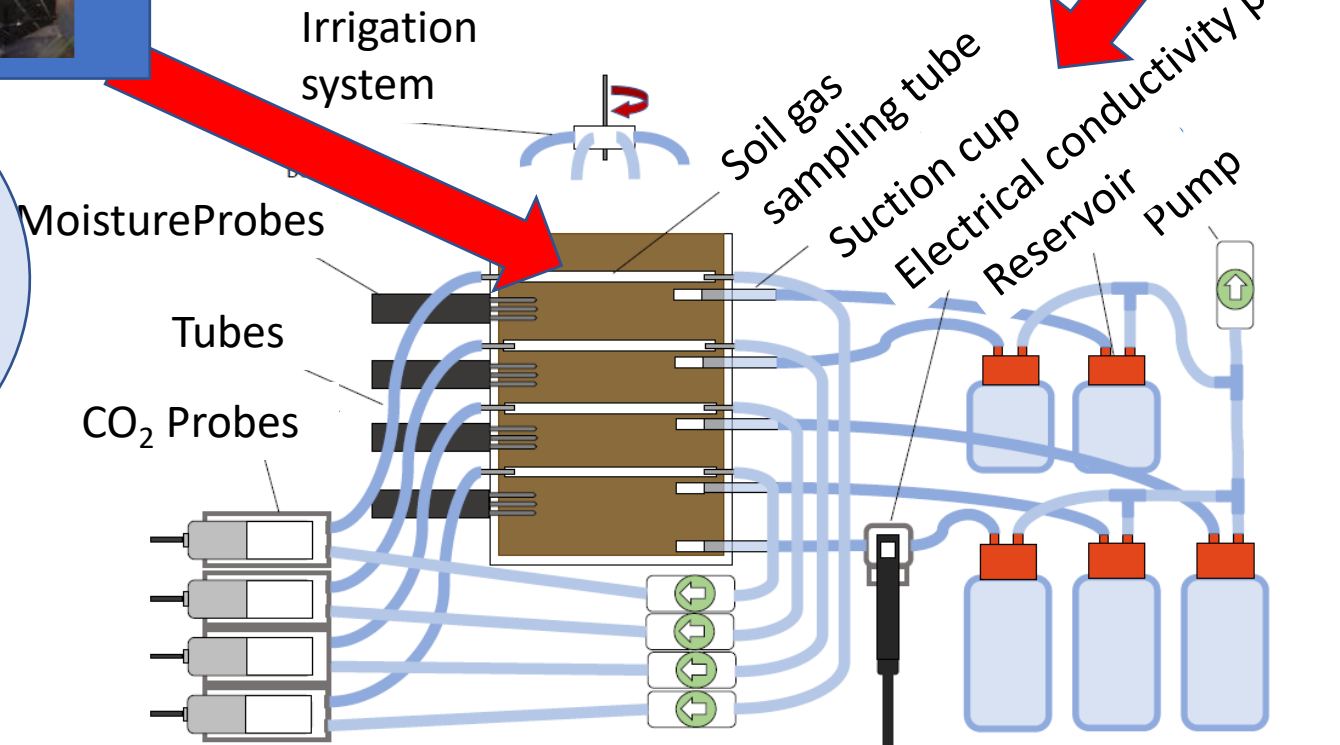
Intact Mescosm



Re-Packed Mescosm



And we ran experiments similar to those in the field, measuring soil CO_2 concentrations, soil moisture, water from suction cups, an electrical conductivity of the draining water.

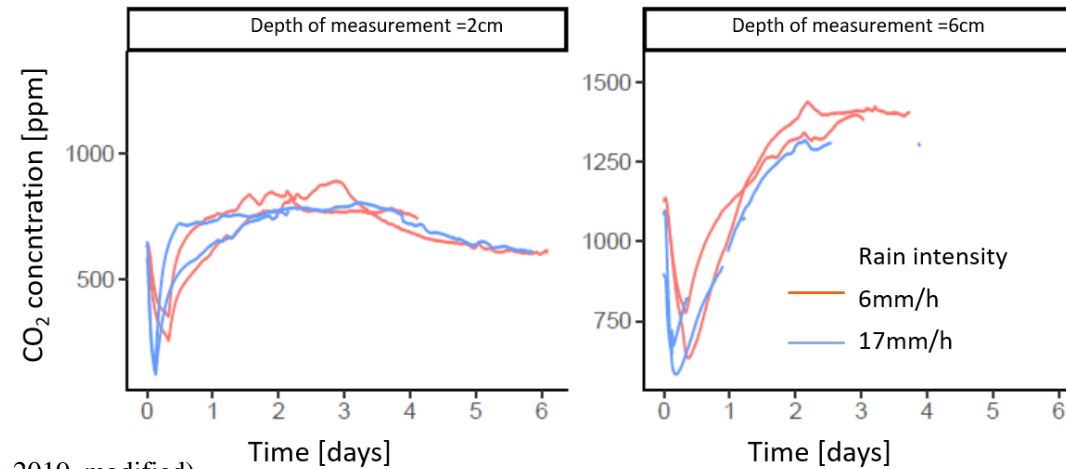
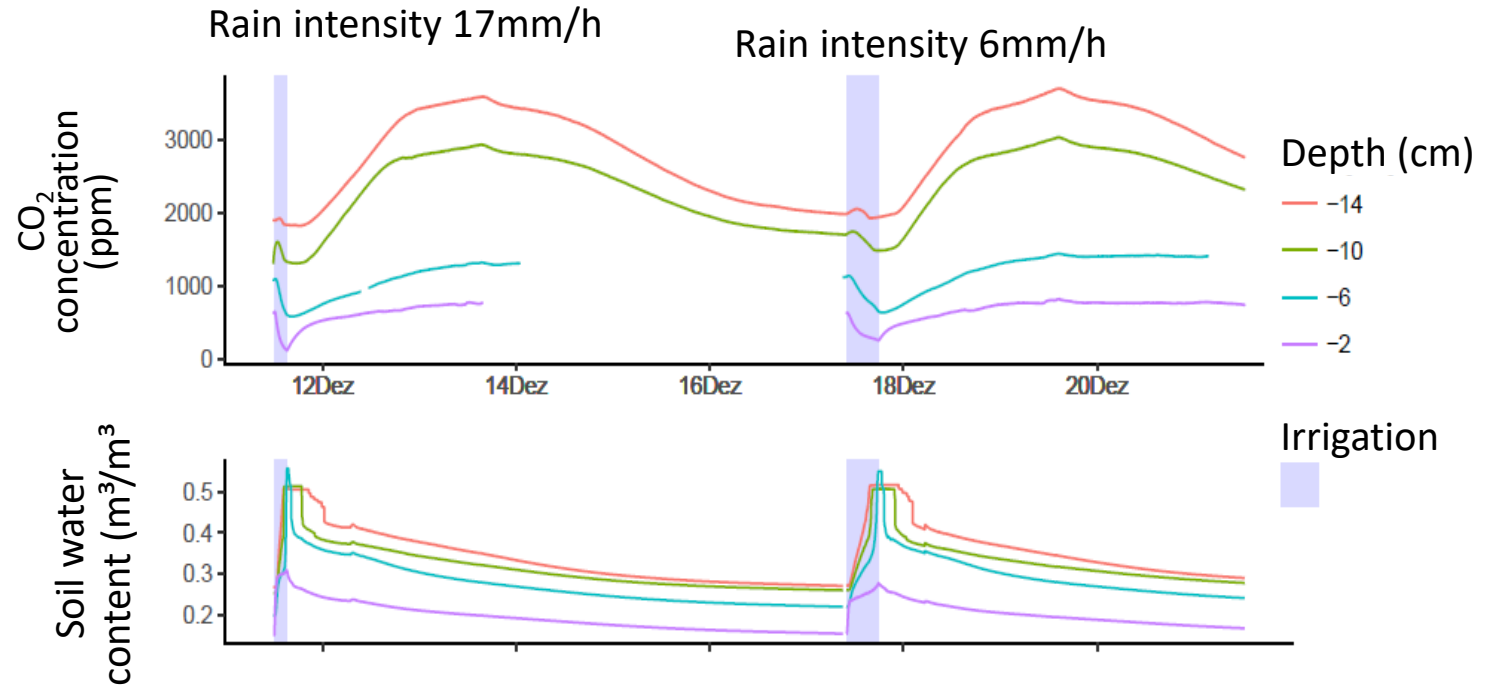


(after Freiberg, 2019, M.Sc. Thesis , University of Freiburg, modified)



And we observed the same dips in soil CO₂ concentrations as in the field, with stronger dips during intense rainfall events.

Examples from the re-packed mesocosm

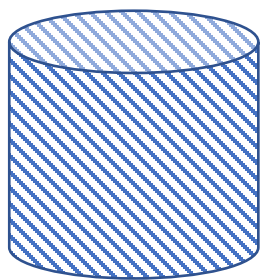
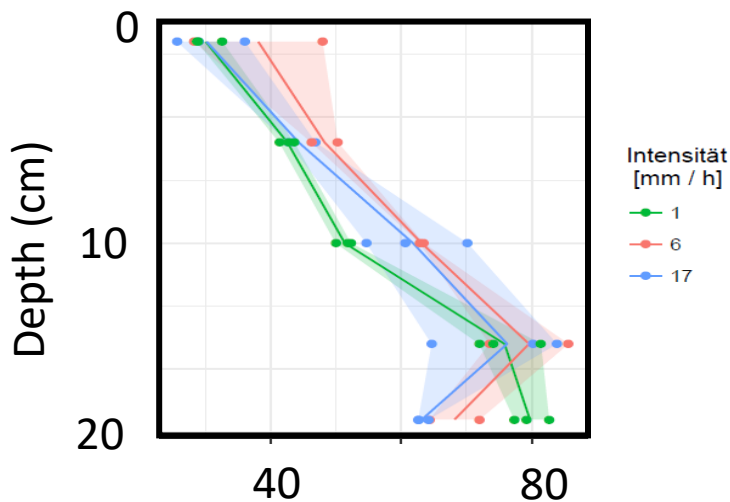


(Freiberg, 2019, modified)

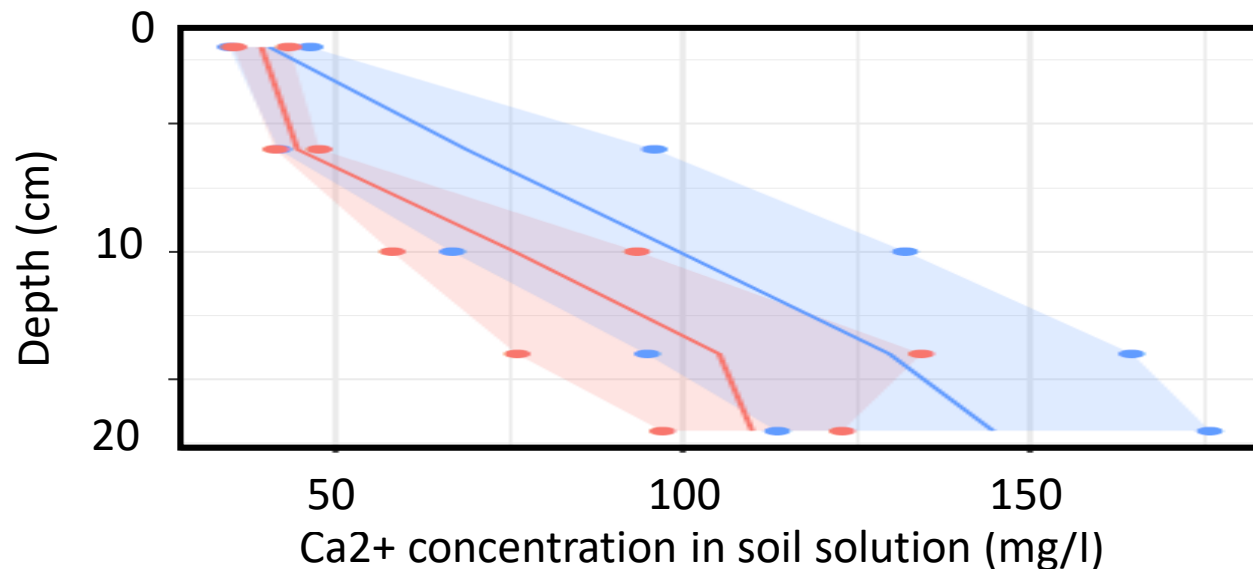
And we also saw that much more Ca^{2+} was washed from the re-packed soil column, especially during the first experiments. The surfaces of the preferential flow paths seem thus less susceptible to mobilize Ca^{2+} , maybe, because most Ca^{2+} has been washed from this pore surfaces already.



Intact Mescosm



Re-Packed Mescosm



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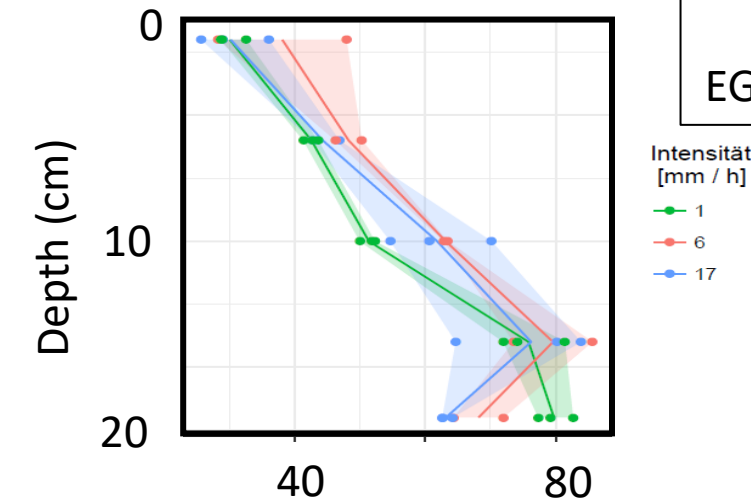
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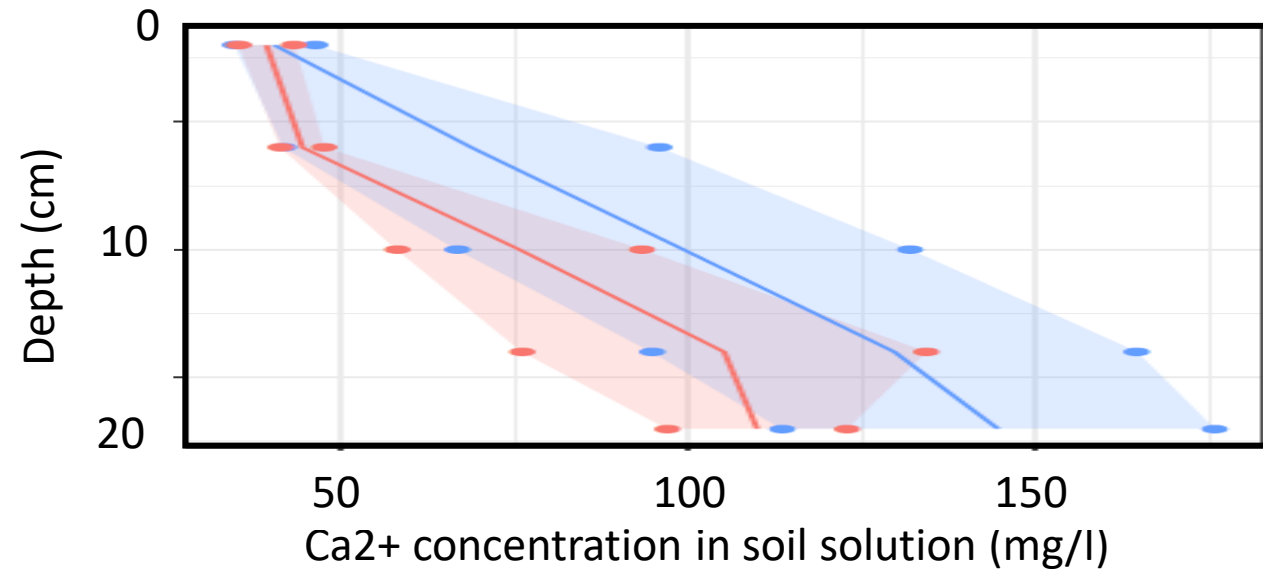
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Intact Mesocosm



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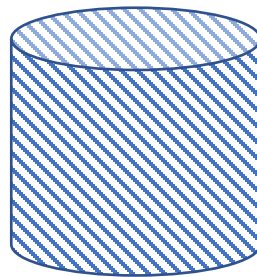


(Freiburg, 2019, modified)



Thank you for watching!

If you have questions or comments you'll find me in the chat on Friday, 8 May 2020, 10:45-12:00, or please find my e-mail via the EGU!



Re-Packed Mesocosm