Formation Water Characterization of the Shale Reservoir Rocks Using Integrated Workflow

Ekaterina S. Kazak^{1,2}, F. Bilek³, A. V. Kazak²

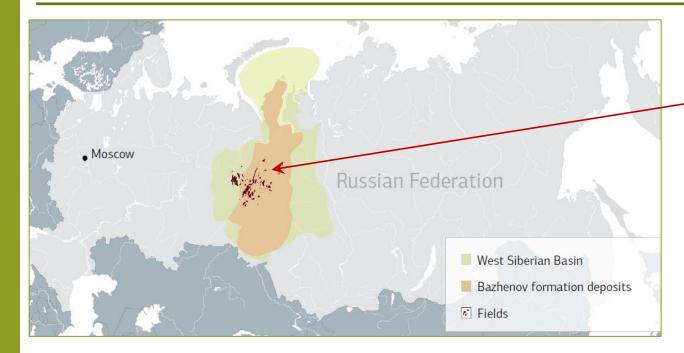
¹ Lomonosov Moscow State University
² Skolkovo Institution of Science and Technology
³ GFI Grundwasser-Consulting-Institut GmbH Dresden, Dresden, Germany







Sample Material



 → Whole core samples of Bazhenov Formation (BF)
(Ø 80÷100 mm) from 5 wells drilled in 5 various oil fields located in the West Siberia (Russia).

- → The distance between the wells ranges from 80 km to 500 km.
- → The coring has been performed within an interval of the BF at a depth of more than 2.5 km.





STATE UNIVERSITY



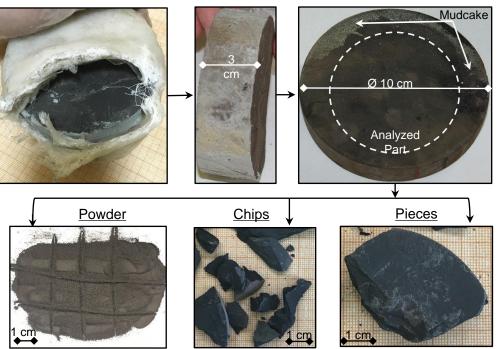
Skolkovo Institute of Science and Technology

EGU General Assembly 2020 | 04.05.2020 | D1473 | EGU2020-20251 | 2

Sample Preparation

Original Core Preservation

Whole Core Fragment



- → Each whole core fragment was preserved from exposure to the environment after opening the core barrel and fragment cutting using polymer (Saran) film and then paraffinsoaked gauze.
- → A rock sample preparation procedure consists of 4 steps: 1) releasing from paraffin sealing shells; 2) crushing 3) weighting the fraction on an analytical balance; 4) distributing according to the requirements and testing.
- → To avoid drilling mud contamination all rock materials were collected from the central part of the core axis.

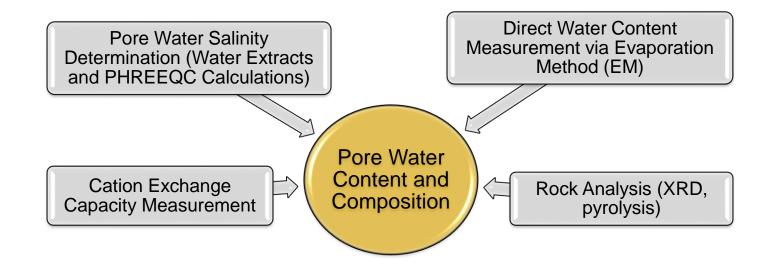






Skolkovo Institute of Science and Technology

Experimental Methods



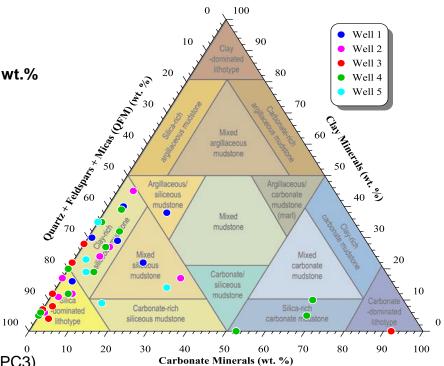
The accurate values of residual pore water content and salinity as a parameter for the organo-mineral model and the interpretation of geophysical data.

Proper resource evaluation and certainty estimation of hydrocarbon reserves for shale oil fields



Mineral Composition, Geochemistry and Rock Typing

- → Gross Mineral Composition
 - → silica up to 86 wt.%
 - → clays (mixed-layer minerals, hydromica, kaolinite) up to 45 wt.%
 - → carbonate minerals
 - → remains: plagioclase, pyrite etc.
 - → No smectite or montmorillonite
- → Geochemistry
 - → TOC 1÷28 wt.%
 - \rightarrow S₁ 0.9÷9.22 mg HC/g rock
 - \rightarrow S₂ 6.37 \div 229.67 mg HC/g rock
 - → T_{max} 428÷437°C
- → gradual variation regarding catagenesis stage or kerogen
- → thermal maturity: from protocatagenesis or immature kerogen (PC3) to metacatagenesis or main oil window (MC3)





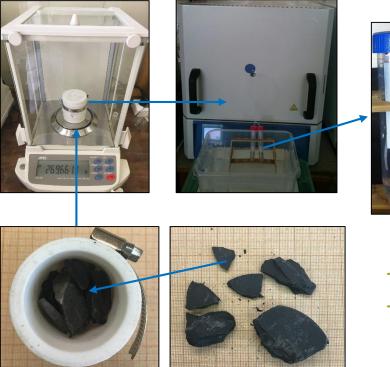


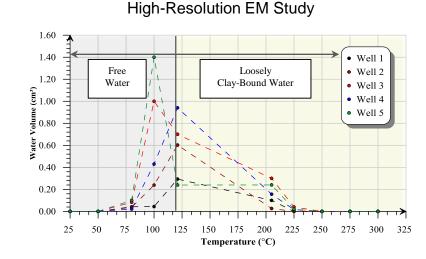
STATE UNIVERSITY



Skolkovo Institute of Science and Technolog

Direct Water Content Measurement via Evaporation Method (EM)





- \rightarrow Up to 121° C free water release
- \rightarrow Up to 250° C residual water (total content of the free and loosely clay-bound water) release
- \rightarrow Accuracy of the Evaporation Method is 0.2 \div 6.8 rel.%



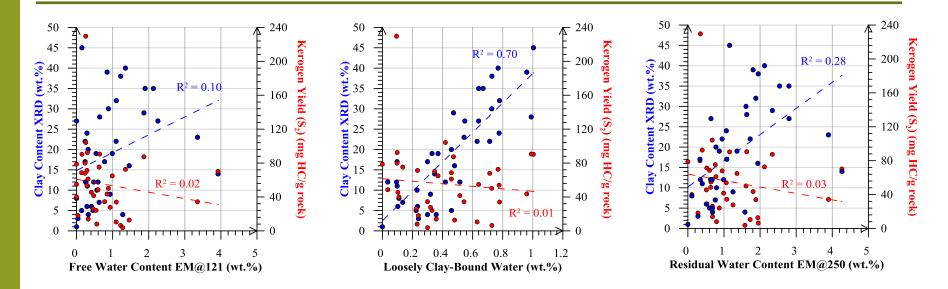


STATE UNIVERSITY



EGU General Assembly 2020 | 04.05.2020 | D1473 | EGU2020-20251 | 6

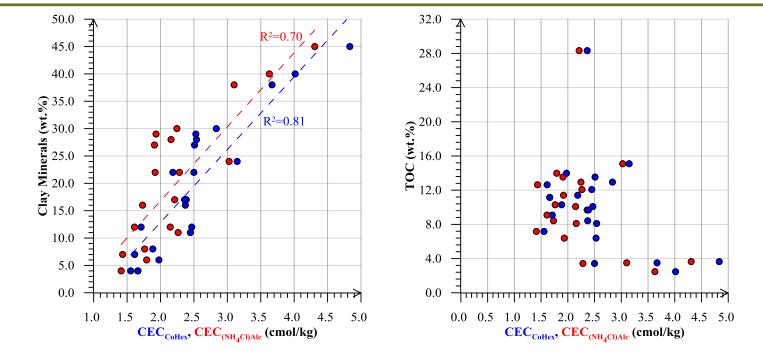
EM Water Content versus Clay Mineral Content and Pyrolysis S₂



- → The target rock samples contain the residual formation water 0.11÷4.27 wt.%, including free 0.04÷3.92 wt.% and loosely clay-bound water 0.09÷0.96 wt.%.
- → The free water content, does not depend on the rock mineral composition and kerogen content.
- → The loosely bound water content correlates well to the clay mineral fraction.



Cation Exchange Capacity (CEC) Measurement

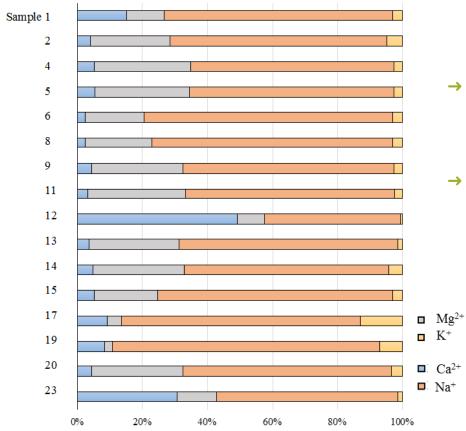


- → 2 methods of cation exchange capacity (CEC) measurement were used alcoholic NH₄Cl (NH₄Cl)Alc and hexaamminecobalt(III) chloride (CoHex).
- → Both CEC methods delivered consistent results.
- → CEC depends on the clay content.

EGU General Assembly 2020 | 04.05.2020 | D1473 | EGU2020-20251 | 8



Cation Exchange Composition



- → Ca, Na, Mg, K form the exchange complex of all studied core samples.
- → According to interrelation (rNa+rK)>rCa, the exchange complex type is marine and was inherited from the composition of the paleobasin seawater (Bazhevov sea).





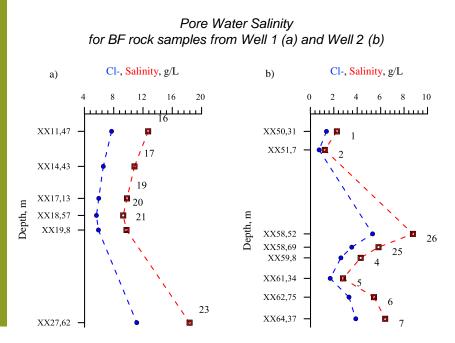
STATE UNIVERSITY



Skolkovo Institute of Science and Technology

Pore Water Composition

→ Using the thermodynamic calculations with PHREEQC, the ratio of cations in pore water was estimated - Na (up to 91%), Mg (up to 5.6%), Ca (up to 2.6%) and K (up to 0.8%).



80 % percentage 30 0.0 0.2 0.4 0.6 0.8 10 1.2 1.4 1.6 1.8 Ionic stength in mol/L m NaX m KX

Cation Exchange composition versus ionic strength

- \rightarrow The water extracts analysis shows that the pore water salinity as NaCl is 1.23÷21.96 g/L.
- → The δ^2 H (-64.5÷-63.8‰), δ^{18} O (-2.0÷1.4‰) are in a good agreement with typical values for formation waters and brines associated with petroleum systems reported for many sedimentary basins.





STATE UNIVERSITY



Conclusions

- The target rock samples contain the residual formation water (0.11÷4.27 wt.%), including free 0.04÷3.92 wt.% and loosely clay-bound water 0.09÷0.96 wt.%. The loosely bound water content correlates well to the clay mineral fraction. The amount of chemically bound water fell in a range of 0÷6.40 wt.% and exceeds that of free and loosely bound water.
- 2) CEC varies from 2.87 to 5.82 meq/kg by (NH₄CI)Alc method and from 2.87 to 6.38 cmol/kg by CoHex method. Both methods could be used for CEC determination in BF rocks. CEC depends on the clay content.
- Na⁺ is a dominant cation in the exchange complex of all investigated samples, which means that all of them belong to the marine type.
- 4) The pore water of BF rocks has mainly Na-CI composition and salinity $1.23 \div 21.96$ g/L.
- 5) The presented research sheds more light on the presence and distribution of the free and loosely clay-bound water and pore water composition in the reservoir rocks of the Bazhenov formation.



Acknowledgements

- → Gazpromneft Scientific-Technology Center for providing core material and accompanying geological and geophysical data, as well as Dr. Aleksey D. Alekseev for setting the goal for the research.
- → Dr. Mikhail Yu. Spasennykh and Dr. Elena V. Kozlova of Skoltech's Center for Hydrocarbon Recovery for valuable input on the geochemical part.
- → Dr. Andrey Yu. Bychkov, R. Khamidulin, Yana. V. Sorokoumova, Ilya Bugaev and Vladimir A. Lekhov of the Faculty of Geology, Lomonosov Moscow State University for valuable help with conducting the experiments.





References on the Topic

Journal of Petroleum Science and Engineering 183 (2019) 106301



A novel laboratory method for reliable water content determination of shale reservoir rocks

```
Ekaterina S. Kazak (Ph.D.)<sup>a,b,*</sup>, Andrey V. Kazak (Ph.D.)<sup>b</sup>
```

^a Faculty of Geology, Lomonosov Moscow State University, GSP-1, Leninskie Gory, Moscow, 119991, Russia ^b Skolkovo Institute of Science and Technology (Skoltech), Sikorskogo 11, Moscow, 121205, Russia

- 1) Kazak E. S., Kazak A. V. Experimental Features of Cation Exchange Capacity Determination in Organic-Rich Mudstones // Journal of Natural Gas Science and Engineering (*currently under peer review*)
- 2) Kazak E. S., Kazak A. V. A Novel Laboratory Method for Reliable Water Content Determination of Shale Reservoir Rocks // Journal of Petroleum Science and Engineering. 2019. vol. 183. pp. 106301.
- 3) Kazak, E. S., Sorokoumova, Y. V. Pore Water Characterization of the Tight Rocks of Bazhenov Formation Using the Water and Salt Extracts // IOP Conference Series: Earth and Environmental Science. 2020. V. 459. – pp. 052075. <u>http://dx.doi.org/10.1088/1755-1315/459/5/052075</u>.
- 4) Kazak E. S., Kazak A. V., Spasennykh M., Voropaev A., Quantity and Composition of Residual Pore Water Extracted from Samples of the Bazhenov Source Rock of West Siberia, Russian Federation, SGEM 2017, 29 June 5 July, Albena, Bulgaria, 2017, pp. 829–841.





OMONOSOV MOSCOV

Skolkovo Institute of Science and Technolog