# Testing different methods for the extraction and purification of leaf and phloem sugars for oxygen isotope analysis

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# **1. Introduction**

Stable oxygen isotopes are key players in environmental sciences. When analysed in plant sugars, they enable tracing those molecules through different plant tissues. Because carbohydrates exchange oxygen atoms with their environment, they enable invaluable insights into how, when, and where they interact with their environment. Hence follows that it is crucial to understand which steps (during sampling or in the laboratory) might unintentionally alter stable oxygen isotopes.

### Aims

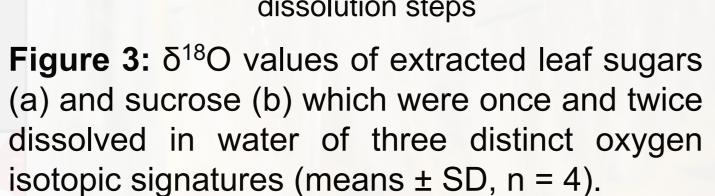
- Highlight potentially problematic work steps when analysing oxygen stable isotopes - from sampling leaf + phloem tissue to obtaining their sugars' final  $\delta^{18}$ O ratios
- Analyse leaf + phloem sugars' δ<sup>18</sup>O and quantify the effect of different techniques used in the laboratory

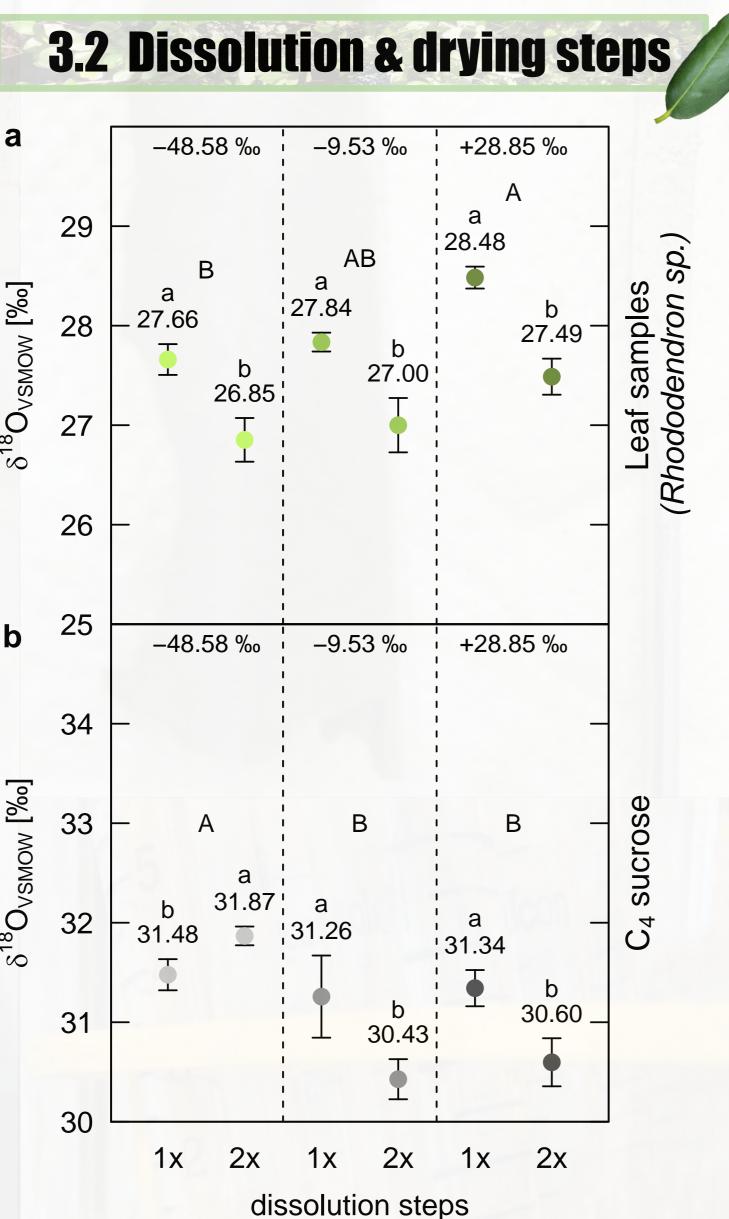
## **3. Results**

**3.1 Drying methods** 

32.2 32.2 C<sub>3</sub> sucrose  $C_4$  sucrose 32.0 32.3 27.8 с 21.0  $\delta^{18}O_{VSMOW}$  [‰] of water

Figure 2: Comparison of oxygen isotope values between oven- (a) vs freeze-dried (b) sucrose samples and phloem sugars from a beech (Fagus sylvatica) twig (means  $\pm$  SD, n = 4). The samples were dissolved in different waters with their own distinct oxygen isotopic composition.







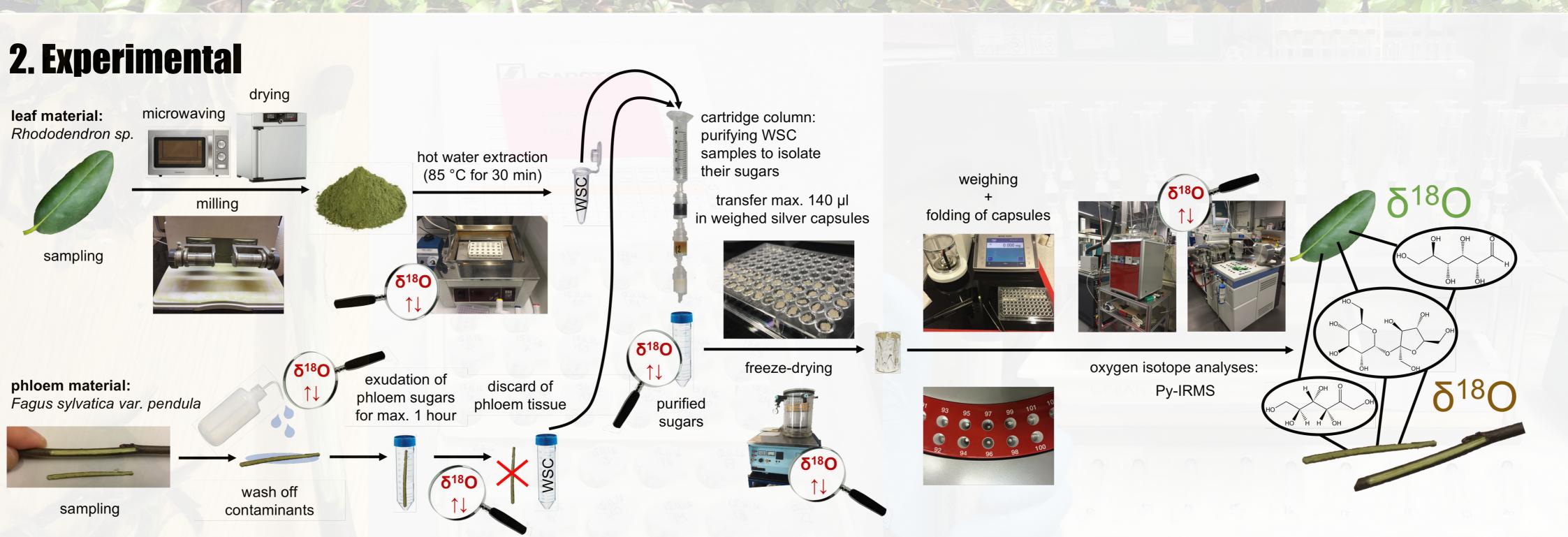


Figure 1: Schematic illustrating the process of oxygen isotope analysis for leaf and twig phloem sugars. Potential 'pitfalls' which might unintentionally offset the sugars' oxygen isotopic composition are highlighted by the magnifying glasses and investigated further in the figures 2-7 below. WSC stands for water-soluble content which consists of amino acids, organic acids, phenolic compounds, and sugars. Purification of the sugars is achieved using SPE (solid phase extraction) whereby three consecutive cartridges with sorbents retain phenolic and ionic compounds.

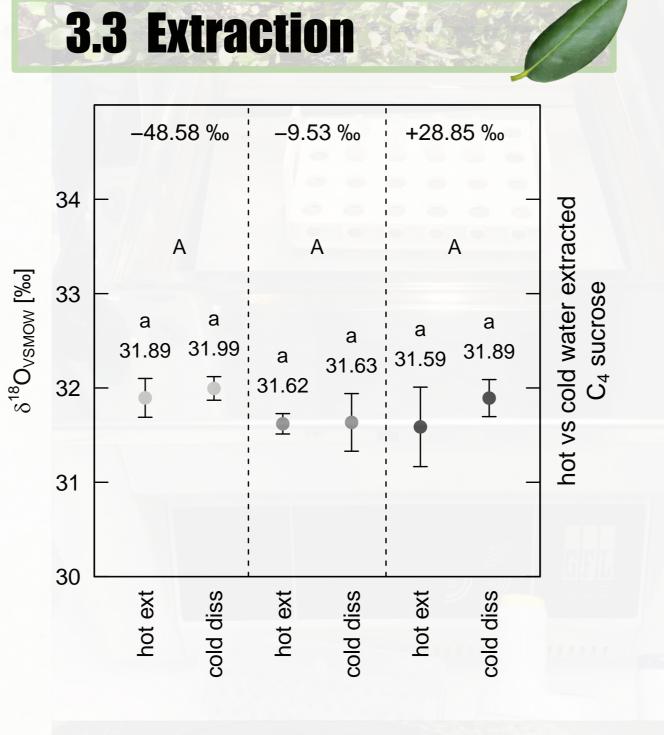


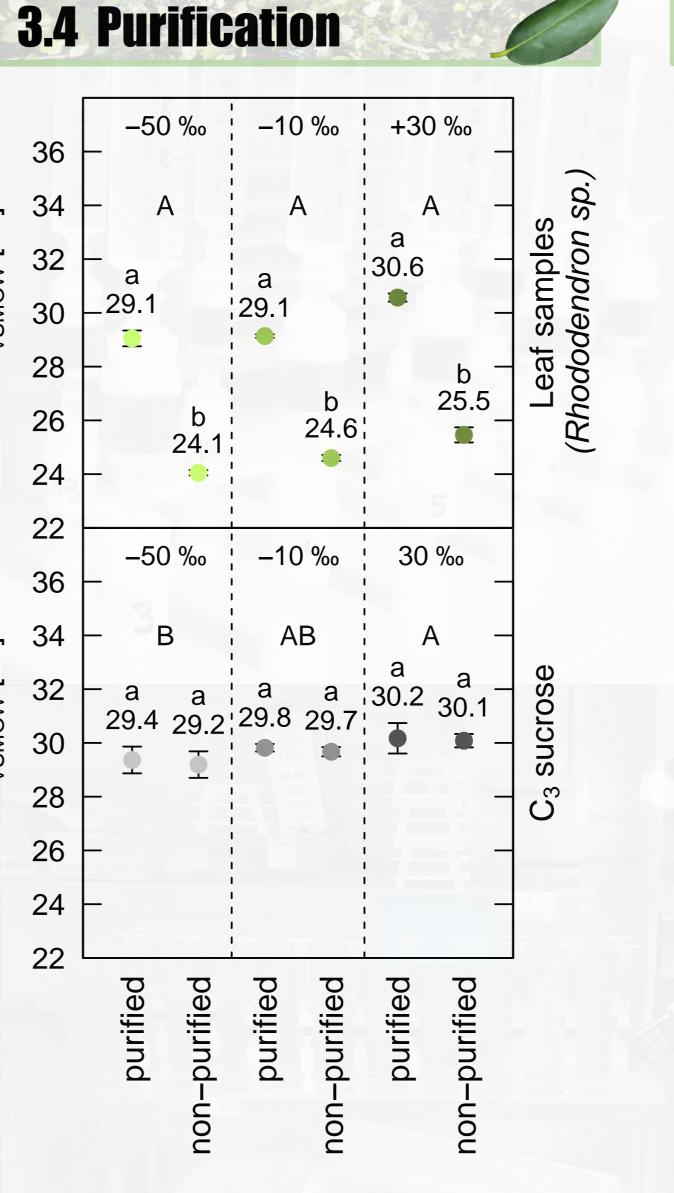
Figure 4: Oxygen isotopic signatures of sucrose which was extracted in hot water for 30 min at 85°C (hot ext), and its counterpart that was only dissolved in cold water (cold diss), (means ± SD, n = 4). The samples were extracted / dissolved in water with different oxygen isotopic signature.

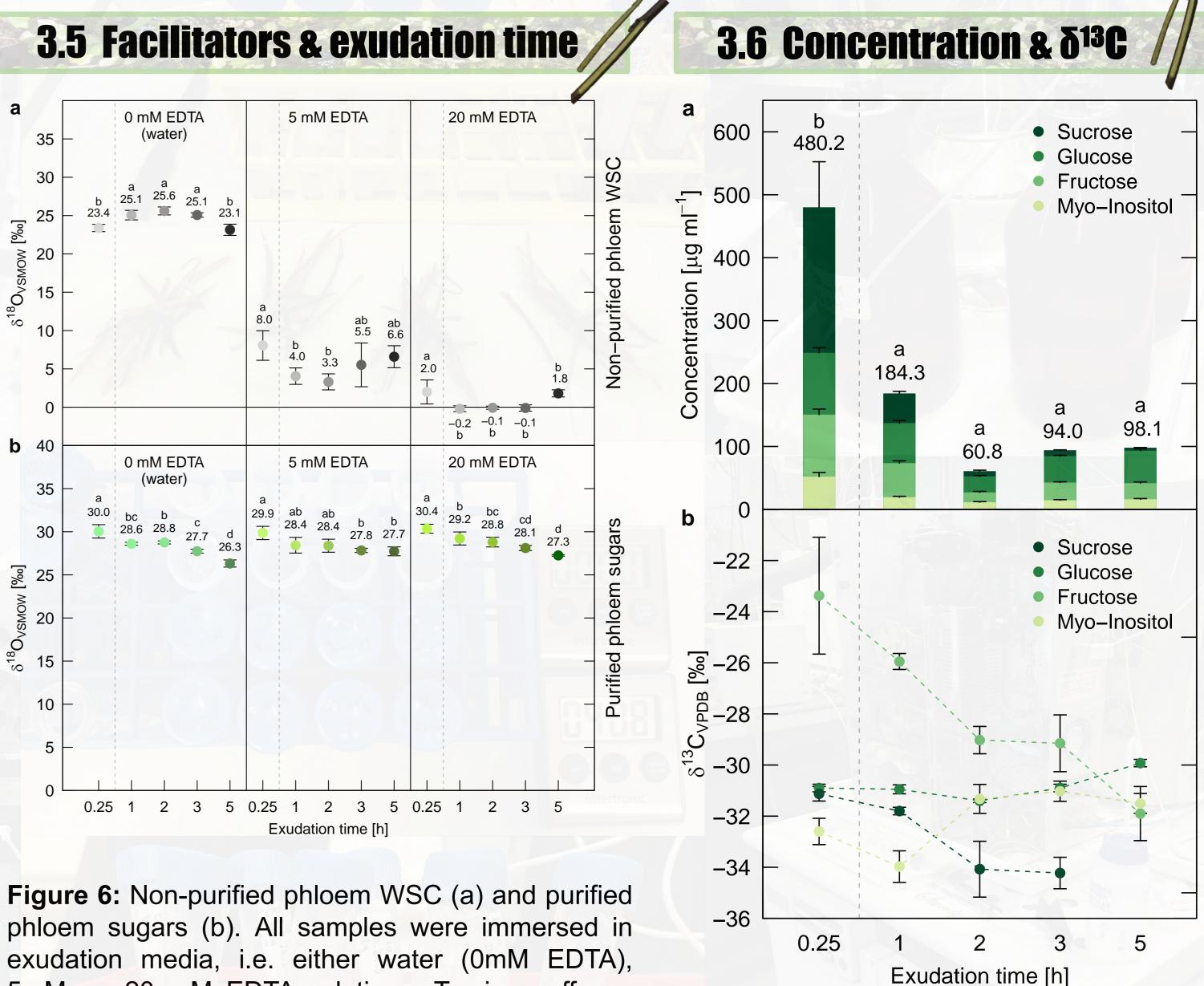
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Figure 5: Oxygen isotope composition of purified and non-purified leaf samples (a) and sucrose (b). Samples were tested using waters of three distinct  $\delta^{18}$ O values (means  $\pm$  SD, n = 4).

# **4. Conclusions**

- water.
- ✓ With





5mM, or 20 mM EDTA solutions. To rinse off any contaminants prior to the sugar exudation, the phloem tissue was immersed for 15 min before they were transferred in fresh media of the same concentrations.



Freeze-drying seems more reliable than oven-drying as it provided more consistent oxygen isotope signatures  $(\delta^{18}O)$  in sucrose + phloem sugars.

Verall, less dissolution + drying steps are advisable, otherwise <sup>18</sup>O-depletion in leaf sugars might occur.

For the extraction of WSC (= water-soluble content), temperature (85 °C) + duration (30 min) appear to not affect the δ<sup>18</sup>O values of sucrose.

/ Purification by SPE cartridges is essential to remove unwanted compounds (incl. EDTA to increase sugar yields). Leaf + phloem sugars were successfully isolated - purified phloem sugars' + pure sucrose's δ<sup>18</sup>O values seem unaltered.

The most consistent δ<sup>18</sup>O values for bulk phloem sugars were received after 1 hour exudations in pure

increasing time, the sugar concentration decreased, while the individual compounds' carbon isotopic compositions were inconsistent.

**Figure 7:** Concentration (a) and  $\delta^{13}$ C values (b) of individual sugars and a sugar alcohol present in the phloem samples (n = 4). An initial immersion lasted 15 min (to wash off contaminants), followed by a transfer in fresh media for 1, 2, 3, and 5 hours (for sugar exudation).

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