The sensitivity of Southern Ocean aerosol concentrations to sea spray and DMS emissions in the HadGEM3-GA7.1 chemistry-climate model

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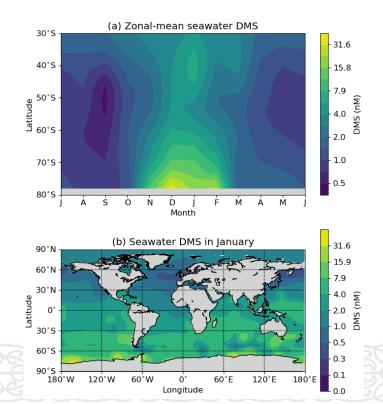


Model simulations

- Simulations performed for the recent past (1989-2008) using the New Zealand Earth System Model AKA
 HadGEM3-GA7.1 (Hadley Centre Global Environmental Model v3, Global Atmosphere 7.1).
 - 85 levels between the surface and 85 km; 1.875°×1.25° horizontal resolution
 - Sea spray aerosol parameterised via Gong (2003)
 i.e. dependent on the 10 m wind speed u^{3.41}₁₀
 - Seawater DMS climatology of Lana et al. (2011)

Gong (2003), Glob. Biogeochem. Cycles Lana et al. (2011), Glob. Biogeochem. Cycles

Seawater DMS maximises at southern high latitudes during austral summer



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Marine aerosols in the model

 Aerosol emission, evolution and deposition are simulated with the Global Model of Aerosol Processes (GLOMAP-mode)

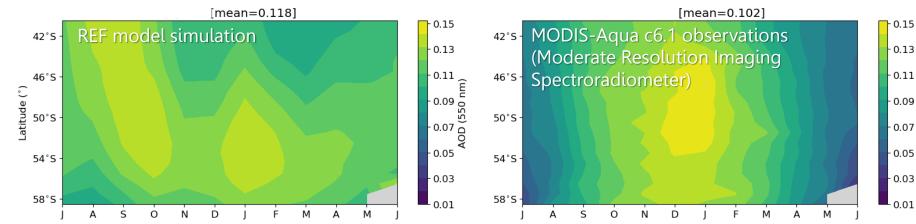
- Aerosols are represented in four log-normal size modes:
 - Nucleation (< 10 nm)
 - Aitken (10 100 nm)
 - Accumulation (100 1000 nm)
 - Coarse (> 1000 nm)



Seasonal biases in total AOD

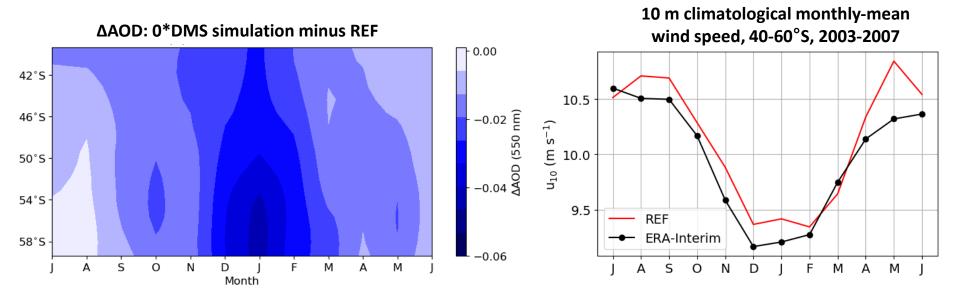


AOD (550 nm)

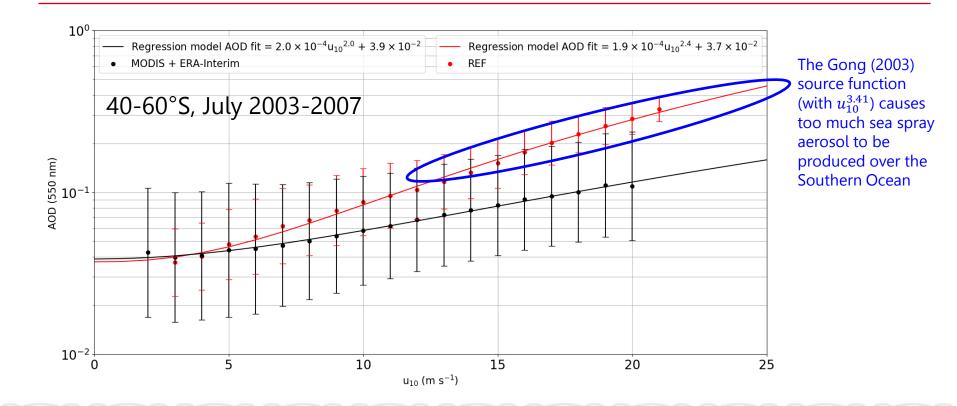


Sea spray aerosol dominates total AOD during JJA





Simulated sea spray aerosol is overestimated at high wind speeds

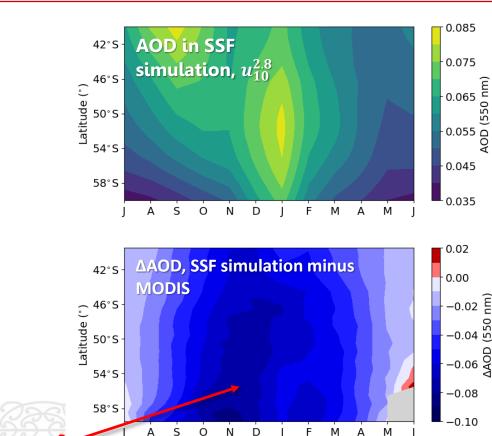


We tested a new sea spray aerosol source function in HadGEM3-GA7.1

Compensating errors

- Hartery *et al.* (2019) developed a powerlaw sea spray aerosol parameterisation similar in form to Gong (2003), but with u^{2.8}₁₀ instead of u^{3.41} → see SSF simulation.
- Based on PCASP-100X optical particle counter measurements during a Tangaroa voyage from Wellington to the Ross Sea in Feb/March 2018.
- At high wind speeds the Hartery *et al.* (2019) sea spray aerosol parameterisation predicts up to 40% less aerosol than Gong (2003).

Hartery et al. (2019), J. Geophys. Res., in review

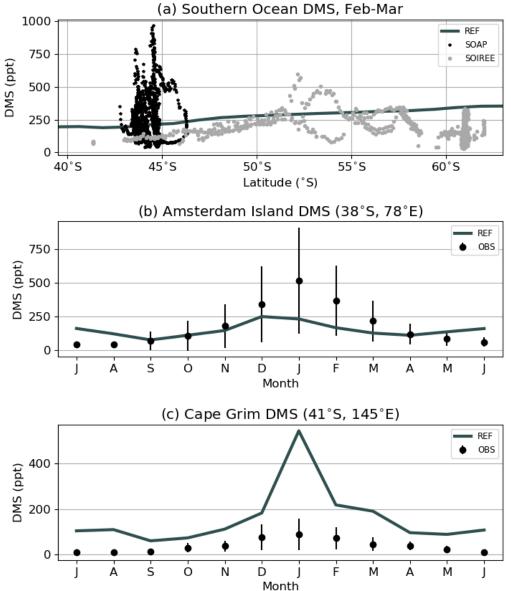


Month



Simulated DMS

- Reasonable agreement between the model and measurements made during the Surface Ocean Aerosol Production campaign (SOAP, Feb-Mar 2012) and the Southern Ocean Iron Release Experiment (SOIREE, Feb 1999)
- Summertime atmospheric DMS concentrations are underestimated at Amsterdam Island but overestimated at Cape Grim.



The current DMS oxidation scheme in HadGEM3-GA7.1:



Gas phase

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DMS + OH \rightarrow SO<sub>2</sub>
DMS + OH \rightarrow MSA + SO<sub>2</sub>
                                                                                                 [DMS = dimethyl sulfide, CH_3SCH_3]
                                                                                                 [MSA = methane sulfonic acid, CH_3SO_2OH]
                          DMS + NO<sub>3</sub> \rightarrow SO<sub>2</sub>
                         DMS + O(<sup>3</sup>P) \rightarrow SO<sub>2</sub>
SO<sub>2</sub> + OH \rightarrow SO<sub>3</sub> + HO<sub>2</sub>
                            HSO_3^- + H_2O_{2(aq)} + H^+ \rightarrow SO_4^{2-} + 2H^+ + H_2O_{(aq)}
Aqueous
phase  \begin{array}{c} HSO_3^- + O_{3(aq)} \rightarrow SO_4^{2-} + H^+ + O_{2(aq)} \\ SO_3^{2-} + O_{3(aq)} \rightarrow SO_4^{2-} + O_{2(aq)} \end{array}
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We tested three DMS chemistry schemes



	s reactions	CHEM 1: DMS + OH DMS + OH DMS + NO ₃ DMS + O(³ P) SO ₂ + OH DMS + BrO DMS + Cl	$\frac{\text{CHEM 2:}}{\text{DMS + OH}}$ $\frac{\text{DMS + OH}}{\text{DMS + OH}}$ $\frac{\text{DMS + NO_3}}{\text{DMS + O(^3P)}}$ $\frac{\text{SO}_2 + \text{OH}}{\text{DMS + BrO}}$ $\frac{\text{DMS + BrO}}{\text{DMS + Cl}}$ $\frac{\text{DMSO + OH}}{\text{MSIA + OH}}$ $\frac{\text{MSIA + OH}}{\text{MSIA + OH}}$
queous pl	reactions	$HSO_{3}^{-} + H_{2}O_{2(aq)} + H^{+}$ $HSO_{3}^{-} + O_{3(aq)}$ $SO_{3}^{2-} + O_{3(aq)}$	$HSO_{3}^{-} + H_{2}O_{2(aq)} + H^{+}$ $HSO_{3}^{-} + O_{3(aq)}$ $SO_{3}^{2^{-}} + O_{3(aq)}$
4		CHEM2 + CHEM3: Chen et al., <i>Atmospheric Chem</i>	<i>istry and Physics</i> , 2018

CHEM 3: DMS + OH DMS + OH $DMS + NO_3$ $DMS + O(^{3}P)$ $SO_2 + OH$ DMS + BrO DMS + ClDMSO + OH MSIA + OH $MSIA + O_3$ $SO_2 + OH$ $DMS_{(aq)} + O_{3(aq)}$ $MSIA_{(aq)} + O_{3(aq)}$ $MSI^- + O_{3(aq)}$ $HSO_{3}^{-} + HOBr_{(aq)}$ $SO_3^{2-} + HOBr_{(aq)}$ $HSO_{3}^{-} + H_{2}O_{2(aq)}$ $HSO_{3^{-}} + O_{3(aq)}$ $SO_{3}^{2^{-}} + O_{3(aq)}$

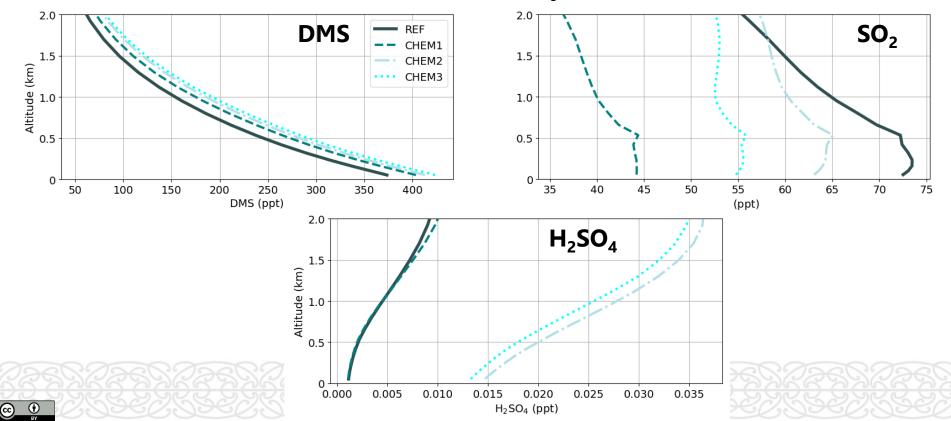
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Changing the chemistry scheme alters the yield of DMS, SO₂ and H₂SO₄

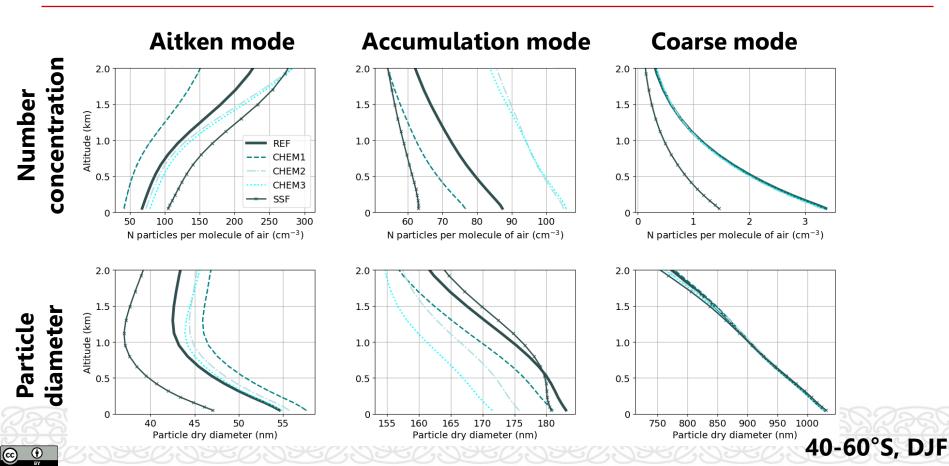


Concentrations in the CHEM sensitivity simulations, 40-60°S, DJF

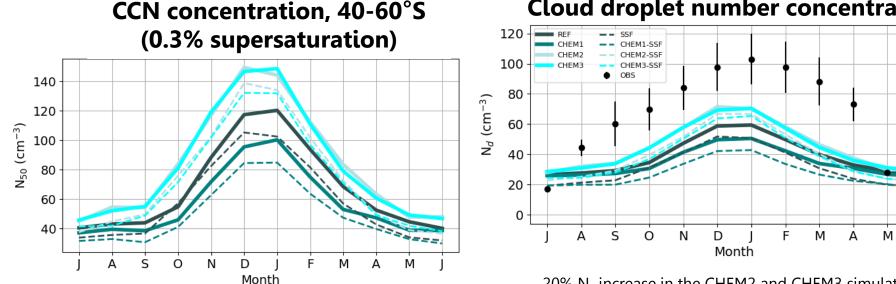


Changes in aerosol mode number concentrations and particle diameters





Improvement in cloud droplet number concentrations in the CHEM2 and CHEM3 simulations

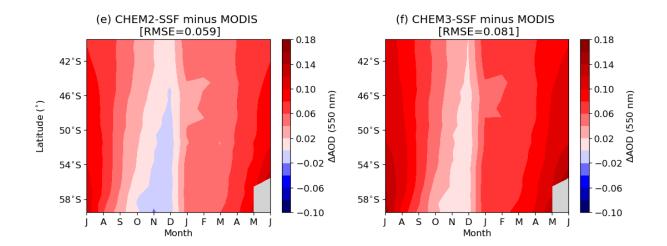


Cloud droplet number concentration

20% N_d increase in the CHEM2 and CHEM3 simulations compared with REF.

N_d observations: Grosvenor & Wood, ACP, 2014

Effects on AOD from combining the new sea salt aerosol parameterization and chemistry schemes



Te Whare Wānanga o Waitaha CHRISTCHURCH NEW ZEALAND

Conclusions

- Reducing the production of sea salt aerosol at high wind speeds improves the simulation of total AOD over the Southern Ocean during JJA.
- New DMS chemistry schemes improve the simulation of cloud droplet number concentrations relative to observations. Summertime AOD agrees better with MODIS observations but otherwise AOD is too high in the model.
- Seasonality must be taken into account in evaluating aerosols in climate models.
- For more detail, see: *Revell et al. (2019), Atmospheric Chemistry and Physics, 19, 15447-15466,* <u>https://doi.org/10.5194/acp-19-15447-2019</u>

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Challenges

THE DEEP SOUTH

Te Kômata o Te Tonga



