

Horizon 2020 - Marie Sklodowska-Curie Actions Innovative Training Network (ITN) **C**omplex **R**h**E**ologies in **E**arth dynamics and industrial **P**rocesses



## Probing the characteristics of mushmagma transition: insights from laboratory experiments

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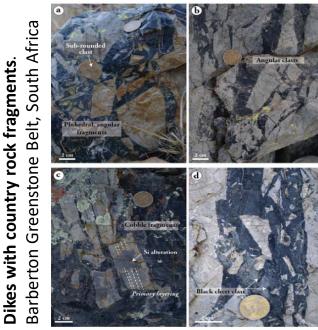


### Context:

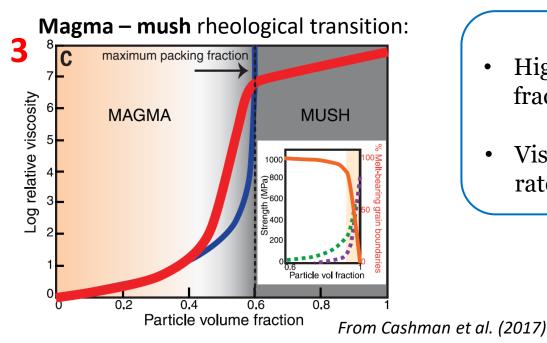
Many geological dynamical processes take place in systems that are heterogeneous in composition, density and mechanical properties:

- settling of crystals and nodules in a magma chamber
- upwelling of magma bodies in heterogeneous lithosphere
- filling of fractures, etc.
- magma-mush transition

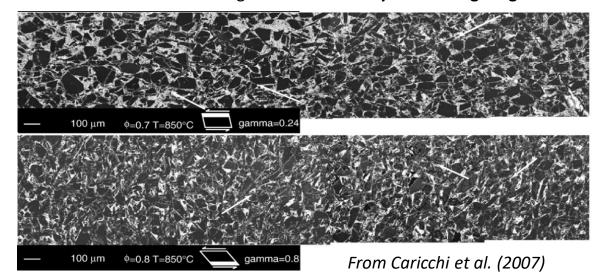
However, description and prediction of these kind of processes often require the definition of an "effective rheology" of the medium.



From Ledevin et al. (2015)



Back scattered electron images of deformed crystal-bearing magma.



### Characteristics of those systems:

• High solid fraction

Jamming =

The local deviatoric stress has to exceed the yield stress ( $\sigma_Y$ ) for the material to flow

Shear-thinning

• Viscosity decreases with increasing shear rate (due to increased organization)

Effective rheology



Examples

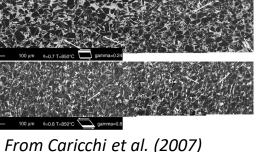
## The experiment: object falling in a fluid

• In **Newtonian** fluid, it's a classical problem and it can be used to measure the viscosity of the fluid.

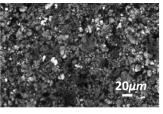
If 
$$Re \ll 1$$
, Terminal velocity:  $v_{stokes} = \frac{2r^2 g \Delta \rho}{9n}$ 

- In a **viscoplastic** fluid:
  - The yield stress (σ<sub>Y</sub>) has to be overcome to unjam the structure
  - $\sigma_Y$  + Shear-thinning + ...

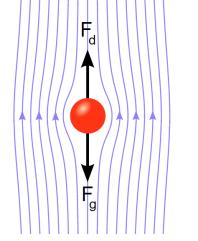
results in a complex rheology Back scattered electron images of deformed crystal-bearing magma.



Carbopol

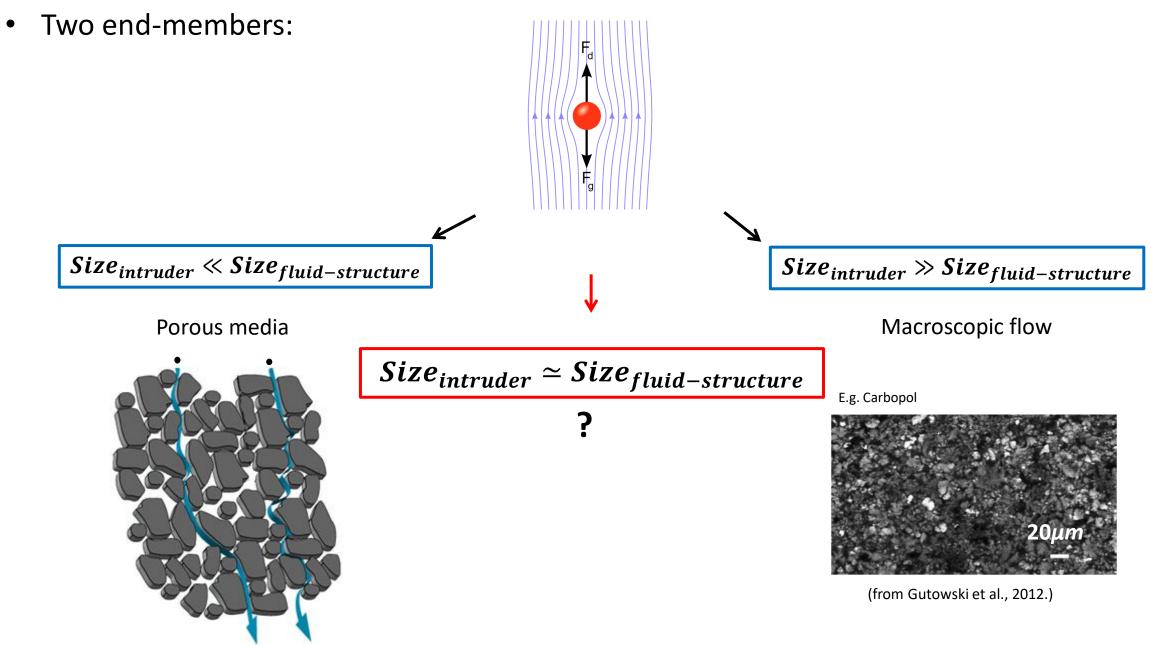


(from Gutowski et al., 2012.)









(From http://hercules.gcsuedu/~sdatta.html)



## The fluid under scrutiny

- The gel is a superabsorbent polymer (**SAP**). It is polyacrylamide, made by copolymerization of acrylic acid and acrylamide
- In water, these polymer powder grains swell up to 200 times (1g of this SAP can absorb up to 200g of water) and form gel grains whose size  $(d_g)$  can be controlled by controlling the size of the initial powder.

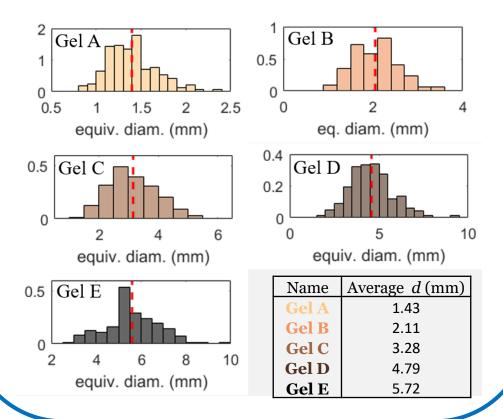
Dry gel

After water absorption



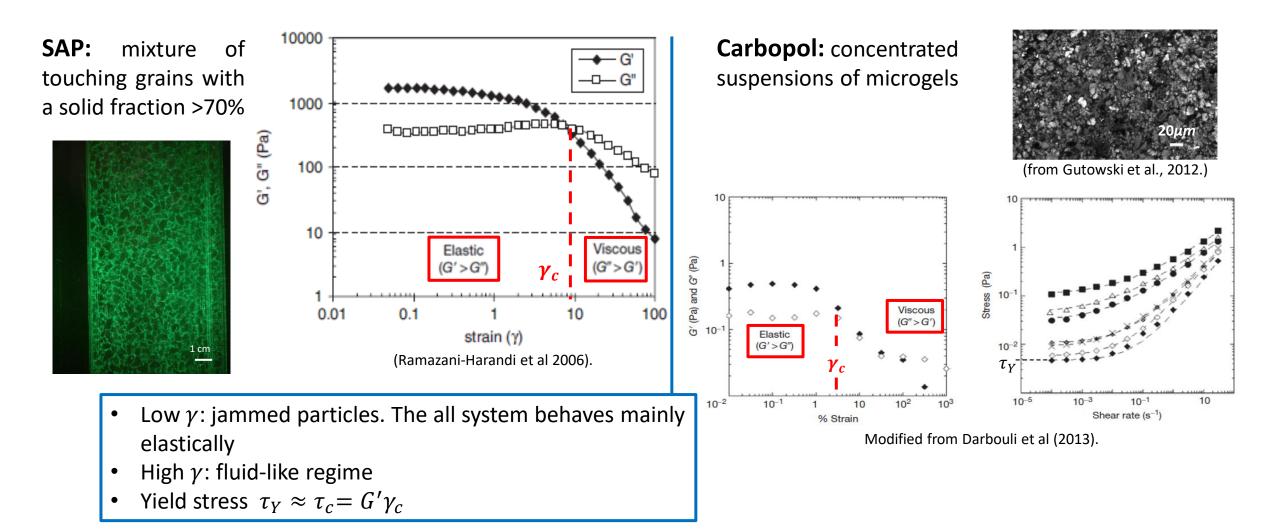
We performed experiments with **5** different grain size gels.

The grain size distributions (by imaging analysis) and average equivalent grain sizes are:



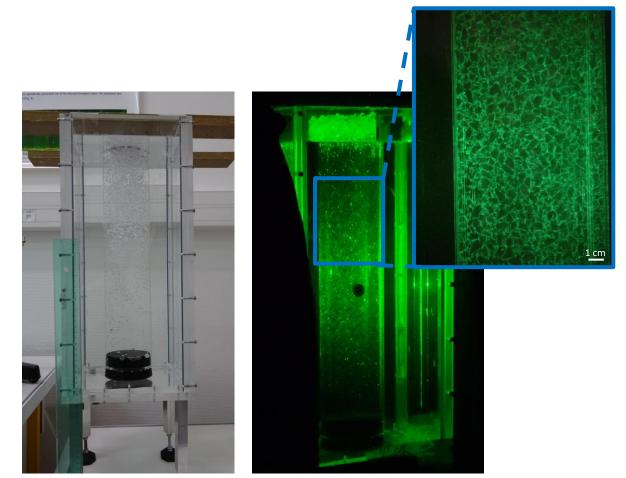


# Rheological behaviour of SAPs and Carbopol gels



### **Experimental setup**

- Sphere was dropped in the center of a cylindrical vessel (100 mm wide and 500 mm deep) previously filled by the gel.
- The diameter of the spheres  $(d_s)$  ranges between 3 and 30 mm. We use spheres of various materials (steel, tungsten, nickel alloy, glass) in order to cover a wide range of densities (from 2200 to 15000 kg/m<sup>3</sup>).
- Gels have been gently stirred for 2-3 days in order to eliminate air bubbles and to avoid preferential paths between different ball releases.

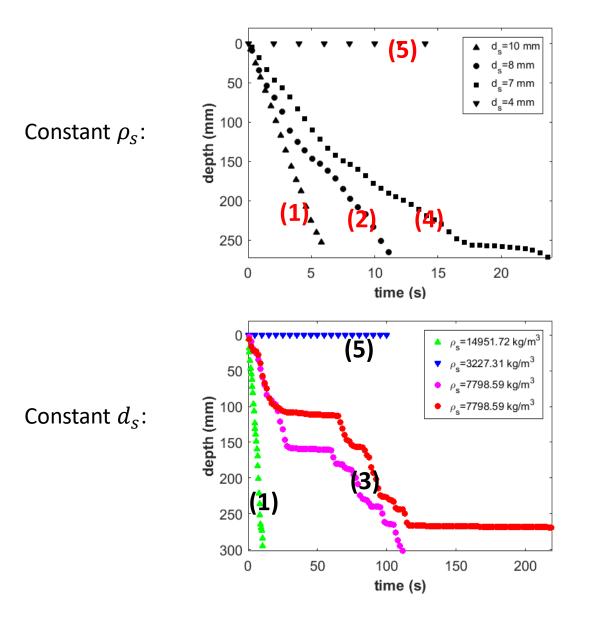




Illuminated cross-section



• **5 motion regimes**. For a given gel we get:



- 1) Linear regime: rapid and straight fall at constant terminal velocity
- 2) Irregular regime superimposed to linear: spheres never stop during their way down ( $v_y > 0$ ) but their velocity varies.
- 3) Stop&go regime: periods of no-motion ( $v_y = 0$ ) and periods of irregular falls follow one another
- 4) Logarithmic regime: a slow fall at a progressively decreasing velocity
- 5) No-motion



### • Scalings:

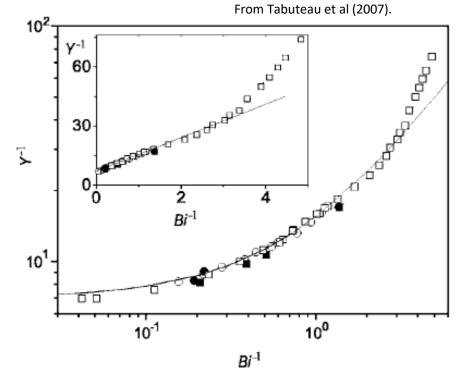
Taking Carbopol as end member (that is,  $d_{sphere} \gg d_{grain}$ ), the slow motion of a sphere (i.e. not considering the stoppage cases) is parameterized by two key dimensionless numbers (Bingham and Yield number) and a master curve:

- Two dimensionless numbers:
  - The yield number:  $Y = \frac{3\sigma_Y}{gd_s\Delta\rho}$
  - The Bingham number:  $Bi = \frac{\sigma_Y}{K_v (v_V/d_S)^n}$
- A master curve:

$$\frac{1}{Y} = 6kX(n) + \frac{6X(n)}{Bi} = 7 + \frac{8.52}{Bi}$$

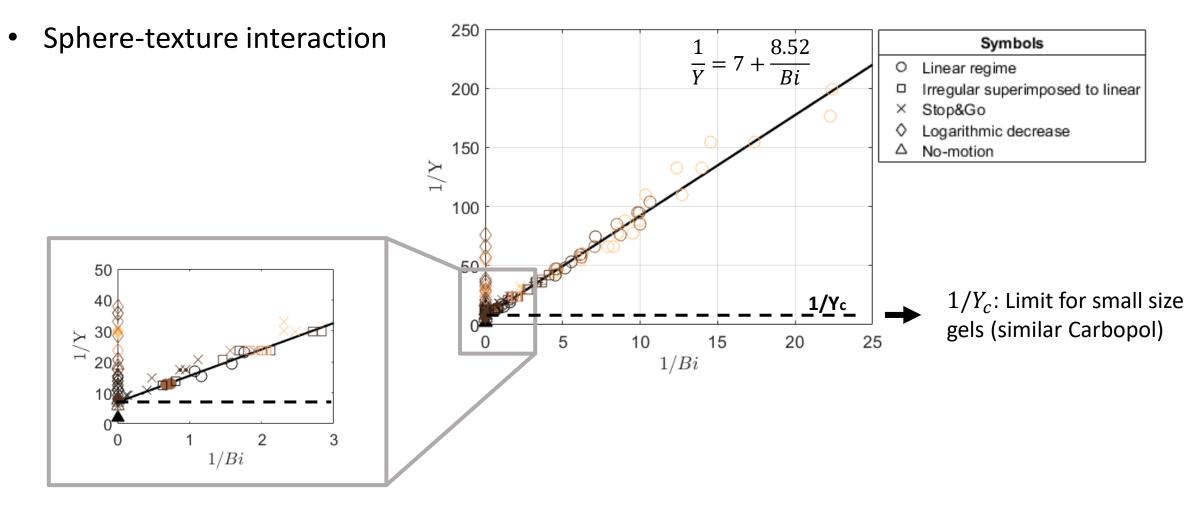
Where:

- > n = 0.5, fitting parameter
- ➢ k = 0.823, found by numerical simulation (Beaulne et al 1997)
- X(n) = 1.42, drag-correction factor for power-law fluids (Gu&Tanner.1985)



From *Tabuteau et al. 2007*, the critical value  $Y_c$  above which there is no motion of the sphere (for  $d_{sphere} \gg d_{grain}$ ) is  $Y_c = 0.145$ .

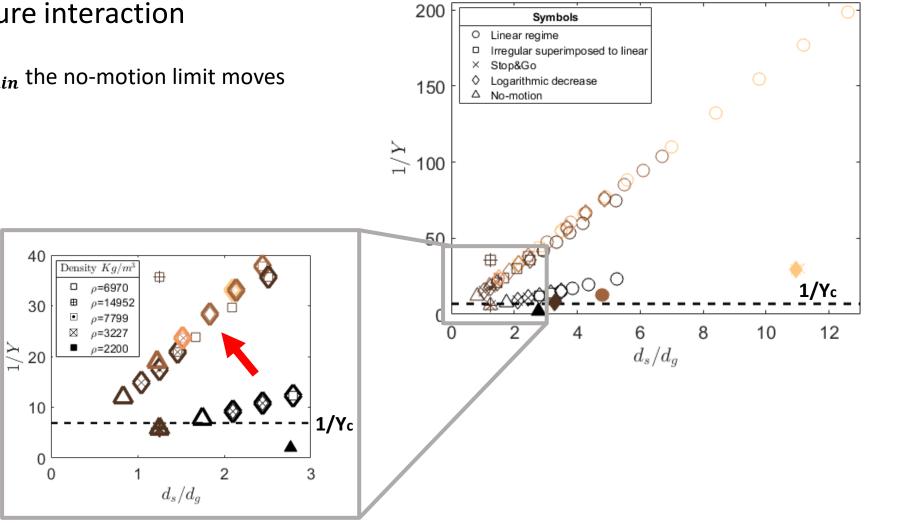




In SAP, no-motion ( $\Delta$ ) and the logarithmic ( $\Diamond$ ) regime are also observed for  $1/Y > 1/Y_c$ . This is due to the interaction between spheres and the gel structure.

Sphere-texture interaction •

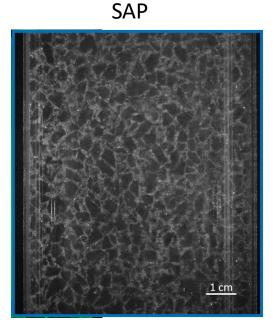
When  $d_{sphere} 
ightarrow d_{grain}$  the no-motion limit moves to higher 1/Y



When  $d_{sphere}$  approachs  $d_{grain}$ , spheres "see" obstacles on their way and the effective rheology breaks down.



• The Mush-Magma transition



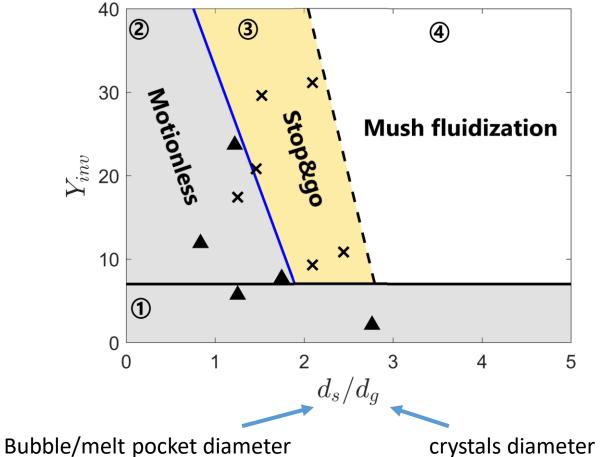
Crystal-rich magma



Melt-rich mush



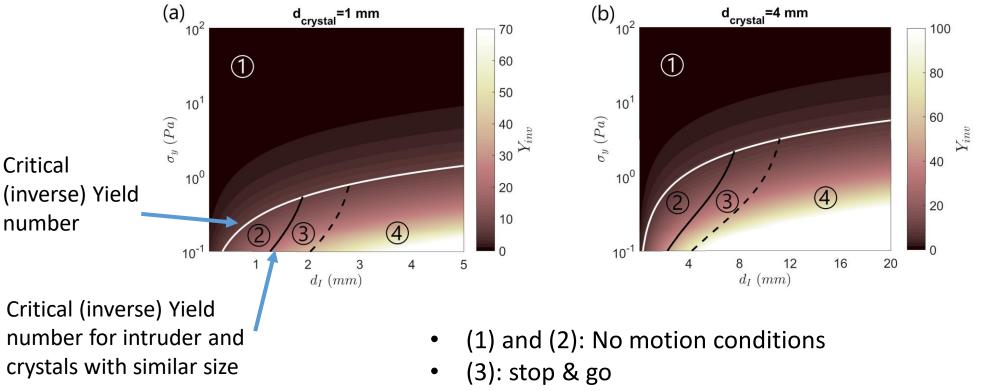
Based on the observed motion regimes in our experiments, we can derive the following regime diagram for an "object" that interacts with a magmatic mush:



• Motion/ no-motion conditions for a buoyant melt pocket (of diameter dI) in a crystal-rich Hershel-Bulkley magmatic reservoir:

(†)

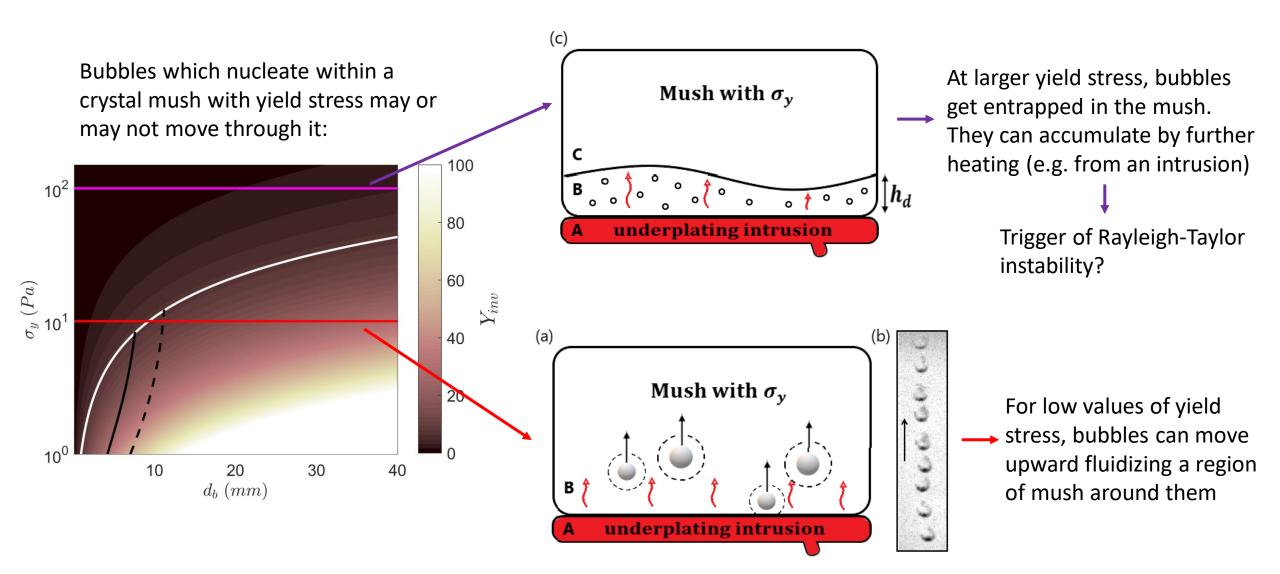
CC



• (4): motion conditions



• Motion/ no-motion conditions for a bubble (of diameter db) in a crystal-rich Hershel-Bulkley magmatic reservoir:





## Conclusion

- Beside the classical steady-state motion and no-motion regimes, typical of viscoplastic fluids, the interaction between moving objects and fluid structure results in two additional regimes where motion becomes more chaotic.
- Considering the mush as a jamming material, large (and buoyant) melt intrusions or bubbles can unjam (i.e. fluidize) the mush around them and move slowly upward.
- Bubbles which nucleate in a crystal mush with yield stress may or may not move through it. In the latter case, the accumulation of entrapped bubbles can form a less dense layer. Further heating can make the layer unstable and a Rayleigh-Taylor instability might develop remobilizing the entire mush.



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