



Limestone reservoirs: are they good for CO₂ geological storage?

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A promising method that could drastically reduce the effects of anthropogenic carbon-dioxide emissions is the capture of CO₂ and its storage in geological formations (CCS technology). The processes that can take place in saline aquifers got under the spotlight in the last decades and the most promising options are sandstone reservoirs. However, natural CO₂ trapped in carbonate (limestone) reservoirs are not well studied. The general assumption is that CO₂ aggressively dissolves the limestone (matrix, grains, and cement), which would cause drastic changes in the reservoir properties (e.g., porosity, permeability).

To better understand the processes that CO₂ injection can cause in a carbonate reservoir, a natural CO₂ subsurface occurrence in Ölbő (Hungary) was investigated, where CO₂ has been trapped safely in the limestone on a geological timescale. Core samples of the reservoir from 1700-1900 m depth were studied with various methods like petrography (carbonate facies analysis, nannoplankton determination), scanning electron microscopy, cathodoluminescence microscopy, X-ray diffraction and infrared spectroscopy. Microdrilling of the carbonates was also carried out to determine the C and O isotope composition of different constituents in order to reveal possible dissolution/recrystallization processes which may occur in the CO₂ reservoir.

Two types of cement were found in the samples, a blocky, drusy cement and a syntaxial cement on the echinoderms (early cement). Contrary to the assumption, dissolution features, may be related to the CO₂ inflow, were not observed in the rocks.

The average mineral composition of the samples is the following: 79 m/m% calcite; 6 m/m% dolomite; 3 m/m% ankerite, mica and quartz; 1 m/m% kaolinite, minor feldspar and pyrite. Dawsonite, the indicator mineral of CO₂ flooding in siliciclastic sandstones, was not identified in the samples.

Carbonate components of the rock are Red algae, Foraminifera, Bryozoa, Bivalves, Echinoderms and Brachiopods. Nearly all were originally calcitic. Based on nannoplankton biostratigraphy and literature, the age of the host rock is Upper Badenian (Serrevallian), Middle Miocene.

The stable C and O isotope data of microfossils shows a narrow range, $\delta^{13}\text{C}$ is ranging from -1.55‰ to 2.05‰ (average: -0.23‰), $\delta^{18}\text{O}$ is between -7.98‰ to -0.25‰ (average: -4.54‰),

expressed on the V-PDB scale. These data do not indicate the effect of magmatic CO₂, which may reside in the Ölbő reservoir (Cseresznyés et al., 2021), in agreement with the petrography. According to our preliminary results, CO₂ inflow did not affect the Ölbő limestone reservoir, i.e., did not imply significant dissolution, neither was involved in cement precipitation. Limestone thus could be an excellent physical trap for CO₂. However, due to limited mineral reactions, our results indicate that limestone reservoirs may not be the best for mineral trapping which is the safest storage mechanism of CO₂ on geological timescale. Further analyses will be carried out with geochemical modeling, to study the water-CO₂-limestone reactions based on the Ölbő CO₂ field.

Reference:

Cseresznyés et al 2021. ChemGeol. <https://doi.org/10.1016/j.chemgeo.2021.120536>