

Crystal chemistry and hydrogen bonding of rustumite $\text{Ca}_{10}(\text{Si}_2\text{O}_7)_2(\text{SiO}_4)(\text{OH})_2\text{Cl}_2$ with variable OH, Cl, F

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Rustumite $\text{Ca}_{10}(\text{Si}_2\text{O}_7)_2(\text{SiO}_4)(\text{OH})_2\text{Cl}_2$, space group $C2/c$, $a \approx 7.6$, $b \approx 18.5$, $c \approx 15.5 \text{ \AA}$, $\beta \approx 104^\circ$ has first been reported from contact-metamorphosed Jurassic limestone at Kilchoan, on the Northwest coast of Scotland (Agrell 1965). Together with dellaite $\text{Ca}_6(\text{Si}_2\text{O}_7)(\text{SiO}_4)(\text{OH})_2$, it is one of two rare ortho- di-silicates, sharing the same Ca/Si ratio and variable Cl content, which can not be routinely distinguished by electron microprobe investigations (Armbruster et al. 2011). Cl contents are between 0 – 5.2 wt.% for dellaite (Armbruster et al. 2011) and 3.22 – 7.38 wt.% for rustumite (this study).

Three samples of the skarn mineral rustumite with variable OH, Cl, F content were investigated by electron microprobe, single-crystal X-ray structure refinements, and Raman spectroscopy: (1) low chlorine rustumite $(\text{Ca}_{10}(\text{Si}_2\text{O}_7)_2(\text{SiO}_4)(\text{OH})_{2.60}\text{F}_{0.12}\text{Cl}_{1.28})$ from skarns associated with the Rize batholith near İkizdere, Turkey; (2) F-bearing rustumite $(\text{Ca}_{10}(\text{Si}_2\text{O}_7)_2(\text{SiO}_4)(\text{OH})_{1.17}\text{F}_{0.87}\text{Cl}_{1.96})$ from xenoliths in ignimbrites of the Upper Chegem Caldera, Northern Caucasus, Russia, and (3) low-Cl, F-bearing rustumite $(\text{Ca}_{10}(\text{Si}_2\text{O}_7)_2[(\text{SiO}_4)_{0.88}(\text{H}_4\text{O}_4)_{0.12}](\text{OH})_{2.09}\text{F}_{1.07}\text{Cl}_{0.84})$ from altered merwinite skarns of the Birkhin massif, Baikal Lake area, Eastern Siberia, Russia. Rustumite from Birkhin massif is characterized by a significant hydrogarnet-like substitution at the orthosilicate group, leading to specific atomic displacements. The crystal structures including hydrogen positions have been refined from single-crystal X-ray data. Depletion in Cl and replacement by OH is associated with smaller unit cell dimensions. The substitution of OH by F leads to shorter hydrogen bonds $\text{HO-H}\cdots\text{F}$ instead of $\text{HO-H}\cdots\text{OH}$. Raman spectra for all samples have been measured and confirm slight strengthening of the hydrogen bonds with uptake of F.

Rustumite is occasionally associated with dellaite. However, a paragenesis of two minerals with the same Ca/Si ratio and similar Cl content is rather unlikely.

Agrell, S.O. (1965) *Min. Mag.*, 34, 1-15, Armbruster, T. et al. (2011) *Min. Mag.*, 75, 379–394.