

Computer modeling of the structure and spectral characteristics of rondonfite series minerals

M. Dulski (1), R. Wrzalik (1), and E. V. Galuskin (2)

(1) University of Silesia, Katowice, Poland (mdulski@us.edu.pl), (2) University of Silesia, Sosnowiec, Poland

Our researches are focused on the study of crystal structure and spectral characteristic of rondonfite with empirical formula $\text{Ca}_8\text{Mg}(\text{SiO}_4)_4\text{Cl}_2$ and a new phases possessing the similar structure. Rondonfite was first described from the Belleberg volcano in Germany. Rondonfite analyzed by us comes from the Upper Chegem volcanic structure, Northern Caucasus, Russia. Rondonfite forms isometric, rounded grains up to 0.3mm in size, it crystallizes in a cubic system (Fd-3 space group). The main structural unit is magnesiosilicate Si_4MgO_16 , pentamer, in which the atomic arrangement is characterized by four isolated SiO_4 tetrahedra combined by four-coordinated Mg^{2+} cation and sometimes by Al^{3+} and $\text{Fe}^{2+}/\text{3+}$ cations. The structural configuration, where Mg^{2+} is in tetrahedral position, is very atypical for silica-minerals due to the geometry restrictions. The magnesiosilicate Si_4MgO_16 pentamers is complemented by 12 CaO_6 octahedra. To balance the charge, big structural cages are occupied by Cl or OH groups. Wide isomorphism between Mg and Al^{3+} or $\text{Fe}^{2+}/\text{3+}$ and also substitution of Cl by OH and H_2O allow us to expect a finding new mineral species in the rondonfite group. Theoretical calculation of structures supposed new mineral phases based on rondonfite structure and checking a stability of these phases make easier the procedure of new mineral species definition.

The structural data obtained for rondonfite crystal using powder and single crystal X-ray analyses were the starting data of our calculations. Modeling of the new phases in the rondonfite group using the isomorphic substitutions ($\text{Ca}^{2+} + \text{Mg}^{2+} \rightarrow \text{Al}^{3+} + \text{Na}^+$, $\text{Mg}^{2+} \rightarrow \text{Fe}^{2+}$ or $\text{Ca}^{2+} + \text{Mg}^{2+} \rightarrow \text{Fe}^{3+} + \text{Li}^+$, $\text{Cl} \rightarrow \text{OH}^-$), checking a stability of these crystal structures and determination of its spectral parameters were performed. Molecular dynamics calculation shows how different type of atoms affects to the reconstruction of the crystal structure of initial minerals. Molecular dynamics allows us also to carry out the calculation under various temperatures and pressures. This calculation allows to predict spectral parameters, i.e infrared (IR), Raman (RS), X-ray photoelectron spectra (XPS) or K- and L1-edge X-ray Absorption Spectra (XAS). These results will be helpful to recognize the new mineral species, processes and conditions, in which these minerals could be formed.

The calculations has been performing by QUANTUM ESPRESSO software using primitive unit cell. This program is based on density-functional theory, plane waves, and pseudopotentials (norm-conserving, ultrasoft, and projector-augmented wave).