

Outgassing of icy bodies in the Solar System - The influence of surface dust layers on the sublimation of H_2O ice

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1. Introduction

The absorption of solar radiation by surface layers of icy bodies in the Solar System (such as icy satellites, Mars and comets) leads to a sublimation of the icy constituents and alters the surface properties. While the energy absorption and its distribution inside the surface layers are not fully understood in detail, there is observational evidence of the formation of nonvolatile surface dust layers and sublimation beneath the residuals of the surface layers (Prialnik and Bar-Nun, 1988; Kömle, 1992; Sunshine et al., 2007). The formation of a non-volatile surface dust layer further decreases the sublimation rate of the icy constituents. Gas molecules escaping through the residuals are scattered by the dust particles, which can lead to a collimation of the gas flux. As a consequence, loose dust particles can be ejected by the gas drag of the outflowing gas.

2. Experimental Setup

We perform laboratory experiments in order to study the influence of dry dust layers on the sublimation of $\rm H_2O$ ice (see Fig. 1; Gundlach et al., 2011), which is the most abundant volatile material in the Solar System. Therefore, cylindrical ice-dust samples are positioned on a cold plate, which is cooled by liquid nitrogen. The experimental chamber can be evacuated to pressures below $\sim 10^{-5} \, \rm mbar$. The gas flux and the temperature of the ice-dust samples are continuously sampled during the measurements. To heat the ice-dust sample's surface by insolation, the pressure gauge can be replaced by a halogen lamp.

3. Results

The first measurements with this experimental setup were performed to investigate the reduction of the gas production rate of $\rm H_2O$ ice surfaces covered by a dry

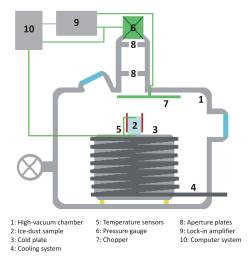


Figure 1: Schematic diagram of the experimental setup (Gundlach et al., 2011). Sublimated H_2O molecules from the sample's surface are detected using a pressure gauge together with a rotating chopper wheel.

dust layer (see Fig. 2; Gundlach et al., 2011). A comparison of the obtained results with a theoretical model of the gas diffusion in random porous media (Skorov et al., 2011) yield a clear difference between theoretical and experimental data.

Furthermore, the so-called sublimation coefficient, defined as the ratio between experimental and maximal sublimation rate (Isono and Iwai, 1969; Kossacki et al., 1999), was derived from these experiments (see Fig. 3). We conducted a systematic evaluation of the temperature dependence of the sublimation coefficient, which has a tremendous influence on the energy balance of sublimating ice surfaces and, therefore, on the thermal modeling of icy bodies in the Solar System.

Recent investigations of insolated ice-dust mixtures have shown that dust particle ejection can occur due

to ${\rm H_2O}$ ice sublimation. The modification of the sample's surface by melting small channels into the ice leads to an increase of the activity at the location of the modifications.

During this conference, we will present our experimental results and their impact on current cometary research.

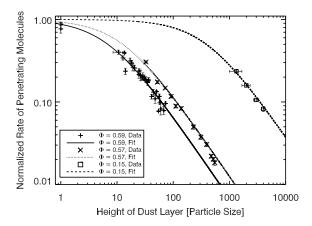


Figure 2: Reduction of the gas production rate of $\rm H_2O$ ice surfaces covered by a dry dust layer. ϕ is the volume filling factor of the dust layer. See Gundlach et al. (2011) for details.

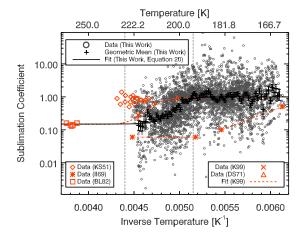


Figure 3: Temperature dependence of the sublimation coefficient (Gundlach et al., 2011). For comparison, the results of the following works are also visualized: Kramers and Stemerding (1951, diamonds), Isono and Iwai (1969, asterisks, red dash dotted curve), Beckmann and Lacmann (1982, squares), Davy and Samarjai (1971, triangle) and Kossacki et al. (1999, cross, dashed curve).

Acknowledgements

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References

- Beckmann, W., Lacmann, R., 1982. Interface kinetics of the growth and evaporation of ice single crystals from the vapour phase II. Measurements in a pure water vapour environment. Journal of Crystal Growth 58, 433–442.
- Davy, J.G., Samarjai, G.A., 1971. Studies of the vaporization mechanism of ice single crystals. Journal of Chemical Physics 55, 3624–3636.
- Gundlach, B., Skorov, Y.V., Blum, J., 2011. Outgassing of icy bodies in the solar system I. The sublimation of hexagonal water ice through dust layers. Icarus 213, 710–719.
- Isono, K., Iwai, K., 1969. Growth mode of ice crystals in air at low pressure. Nature 223, 1149–1150.
- Kömle, N.I., 1992. Temperature evolution of porous ice samples covered by a dust mantle. Icarus 96, 204–212.
- Kossacki, K.J., Markiewicz, W.J., Skorov, Y.V., Kömle, N.I., 1999. Sublimation coefficient of water ice under simulated cometary-like conditions. Planetary and Space Science 47, 1521–1530.
- Kramers, H., Stemerding, S., 1951. The sublimation of ice in vacuum. Applied Science Research A3, 73–82.
- Prialnik, D., Bar-Nun, A., 1988. The formation of a permanent dust mantle and its effect on cometary activity. Icarus 74, 318 323.
- Skorov, Y.V., v. Lieshout, R., Blum, J., Keller, H.U., 2011. Activity of comets: Gas transport in the near-surface porous layers of a cometary nucleus. Icarus 212, 867 876.
- Sunshine, J.M., Groussin, O., Schultz, P.H., A'Hearn, M.F., Feage, L.M., Farnham, T.L., Klaasen, K.P., 2007. The distribution of water ice in the interior of Comet Tempel 1. Icarus 190, 284–294.