

Surface-atmosphere interactions on Triton and Pluto

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1. Introduction

The surfaces of Pluto and Triton are covered by a variety of ices, the most volatile of which being N_2 , CO and CH_4 . Sublimation equilibrium of these ices at surface temperatures of ~ 40 K produces a predominantly N_2 atmosphere, in which CO and CH_4 must be present as well. However, the respective gas mixing ratios depend on the details of the surface-atmosphere interactions. For a long time, information of the gas abundances has been restricted to the single detection of CH_4 in Triton's atmosphere by Voyager [1] and in Pluto's from ground-based observations [2]. A further complication is that the atmospheres of Triton and Pluto, monitored by stellar occultation observations, are known to vary temporally.

2. Observations

We observed Pluto and Triton in 2008 and 2009 respectively, using the CRIRES spectrometer installed on ESO VLT UT1 telescope, using AO mode. We covered the 1.64-1.68 μm and 2.32-2.37 μm ranges at Pluto and Triton respectively, in both cases at a spectral resolution of 60,000. In Pluto's case, the spectral range shows many lines due to the $2\nu_3$ band of gaseous CH_4 . Interpretation is complicated by the fact that Pluto's surface pressure and lower atmosphere temperature structure are uncertain. Nonetheless the data indicate a CH_4 / N_2 mixing ratio of 0.5 ± 0.1 % [3].

At Triton, the 2.32-2.37 μm range contains both signatures due to the $\nu_3 + \nu_4$ band of CH_4 and to the CO(2-0) band. These observations represent the first observations of methane gas since Voyager and the first ever detection in CO in Triton's atmosphere [4]. The CH_4 partial pressure is found to be ~ 10 nbar, i.e. ~ 4 times higher than measured by Voyager. The CO partial pressure is determined to be ~ 24 nbar (to within a factor of 3). Stellar occultations indicate that Triton's surface pressure has increased by a factor of

~ 2 over the 1990s. This is a probable consequence of seasonal evolution, with passage through southern summer solstice in 2000. Assuming $p = 40$ μ bar in 2009, this gives typical mixing ratios of 6×10^{-4} for CO and 2.4×10^{-4} for CH_4 at the surface.

3. Surface-atmosphere interactions

Three scenarios can be a priori envisaged for the nature of surface-atmosphere interactions (i) the "ideal mixture" (solid solution); in this case, the expected partial pressure of each species is the product of its solid mole fraction and its pure vapour pressure. Given the "bulk" abundances of CO and CH_4 ices on Pluto and Triton's surfaces, this scenario implies atmospheric abundances of CO and CH_4 much lower than observed (see Fig. 1 for Triton) and can be excluded (ii) the "pure ice" scenario. This case is relevant if segregated patches of the different volatiles are present on the surface. In this case the atmospheric mixing ratios are in simple proportion of the pure vapour pressures at the relevant ice temperatures, and, except for the main species controlling the pressure, of the fractional area covered by each ice (iii) the "detailed balancing model" [5]. In this case, seasonal evolution of a N_2 -dominated solid solution, with preferential sublimation of N_2 creates a thin surface layer enriched in the less volatile species (CO and CH_4). This "detailed balancing layer" controls the surface-atmosphere exchanges, and in the simplest case where no diffusive separation occurs in the atmosphere, the atmospheric mixing ratios are identical to those in the volatile reservoir below the surface veneer.

We suggest that the detailed balancing model is adequate to explain CO at Triton, whose observed abundance in the atmosphere ($\sim 6 \times 10^{-4}$) and in the surface ($\sim 5 \times 10^{-4}$, [6]) are very similar. Moreover, the miscibility of CO and N_2 ices in all proportions, and the remarkably similar longitudinal distributions of

CO and N₂ ices at Triton [7] strongly argue against the presence of separate CO patches on Triton.

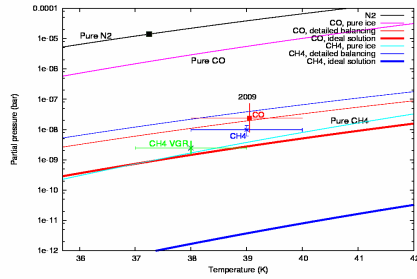


Figure 1: Abundance measurements and equilibrium curves for Triton's volatiles. The black, pink, and light blue curves show the vapour pressure of pure N₂, CO, and CH₄ ices. For CO and CH₄, the thick curves (red and blue, respectively) show the partial pressures for an ideal solid solution with N₂, with CO and CH₄ respective abundances of 5×10^{-4} and 1×10^{-3} in the ice. For CO and CH₄, the thin red and blue lines show the partial pressures expected in the framework of the detailed balancing model. The CO partial pressure we measure is consistent with expectations from the detailed balancing model, while CH₄ is lower. From [4].

On the other hand, applying the detailed balancing model to CH₄ would imply a CH₄ mixing ratio in the surface ice film as high as 50-80 %. This is well beyond the maximal solubility of CH₄ in N₂ (typically 10 %), so that the formation of pure CH₄ grains may be a more likely outcome of the seasonal evolution of a N₂-CO-CH₄ mixture. The presence of pure CH₄ patches is established for Pluto [8], and their ability to elevate the CH₄ atmospheric mixing ratio to ~0.5 % through their enhanced temperatures has been demonstrated [9]. Evidence for such pure CH₄ patches is not yet available on Triton, although the longitudinal distribution of CH₄ there is different from that of N₂ and CO. The increased CH₄ partial pressure in 2009 compared to 1989 is probably the consequence of seasonal evolution. Since the CH₄ vapour pressure is more temperature-dependent than that of N₂, we expect that the atmospheric CH₄ may increase at a higher pace than the pressure.

In summary, we believe that the CO atmospheric abundance reflects the detailed balancing model while that of CH₄ is the consequence of the presence

of pure CH₄ patches on the surface. Additional observations expected in the summer of 2010 may shed further light on this problem. If successful, they will be presented at the meeting.

References

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