

Trace the evolution of organic matter in interplanetary objects using residue analogues

G. Danger^{1*}, A. Fresneau¹, N. Abou Mrad¹, P. de Marcellus², F-R Orthous-Daunay³, P. Modica², V. Vuitton³, F. Duvernay¹, L. Flandinet³, L. Le Sergeant d'Hendecourt², R. Thissen³, T. Chiavassa¹

¹ Aix-Marseille Université, PIIM UMR-CNRS 7345, F-13397 Marseille, France (gregoire.danger@univ-amu.fr / Fax: +33491289194)

² Univ. Paris-Sud, Astrochimie et Origines, Institut d'Astrophysique Spatiale, UMR-CNRS 8617, Orsay, 91405, France

³ UJF-Grenoble 1 / CNRS-INSU, Institut de Planétologie et d'Astrophysique de Grenoble (IPAG) UMR-CNRS 5274, Grenoble, F-38041, France.

Abstract

This contribution focuses on one aspect of our work, which is related to the analysis of refractory residues formed from the UV irradiation and the subsequent warming-up to room temperature of astrophysical ice analogs, the RAHIA project. The understanding of the chemical composition of these refractory residues, commonly called “yellow stuff”, as well as the possible pathways to their formation in astrophysical environments, is an important step to establish what kind of organic matter could be available within interplanetary objects such as comets or asteroids, part of which end up as preserved meteorites on telluric planets.

1. Residue analogues for studying the chemical evolution in astrophysical environments

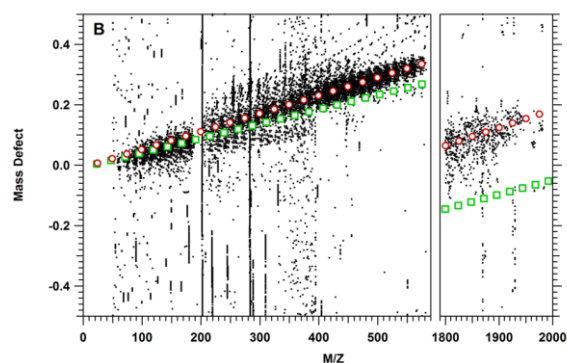


Figure 1. Mass defect vs Exact Mass diagram corresponding to mass spectra of the ¹³C residue analyzed in the negative ESI mode^[1].

We present here the first results obtained by spectrometric analysis with high resolution mass

spectroscopy (LTQ-XL-Orbitrap) of these residues^[1]. These analyzes show that residues are composed of thousands of molecules of high molecular weight ($m/z > 4000$), and present an average elemental composition H/C= 1.6, N/C= 0.4, O/C= 0.4 for an initial ice standard ice mixture, containing H₂O:CH₃OH:NH₃ 3:1:1 (Figure 1).

We also develop specific data representation in order to obtain information on the residue composition^[2]. These representations allow to define that three different groups of molecules are present in these residues, molecules bearing only CHN, CHO or CHNO atoms. These representations also give important information on the family composition of each molecular group. All these developments will be used for the comparison of various residues as well as for the development of more specific analytical methods such as UHPLC-MS or GC-MS. These results demonstrate that from only three simple molecules CH₃OH, H₂O and NH₃, a very complex chemistry occurs when these molecules are subjected to physical processes in the solid state such as those possibly present in the bulk of interstellar grains in the primordial molecular cloud at the time of the Sun formation and possibly then incorporated in comets and/or asteroids.

2. Residue analogue composition vs. meteorites

Furthermore we tentatively compare the abundance of the molecular families constituting our residue to molecules detected from meteorite analyses^[2]. Not so surprisingly, an excellent qualitative and semi-quantitative agreement is obtained between our residues and the soluble organic matter extracted from the Murchison meteorite, demonstrating that such residue can be used for tracing the chemical

history that leads to the formation of the organic matter found in meteorites for instance.

Acknowledgements

This work has been funded by the French national programs “Physique Chimie du Milieu Interstellaire” (P.C.M.I, INSU), “Programme National de Planétologie” (P.N.P, INSU), “Environnements Planétaires et Origines de la Vie” (E.P.O.V, CNRS), the “Centre National d’Etudes Spatiales” (C.N.E.S) from its exobiology program and a PhD grant from the Région Provence Alpes Côte d’Azur (PACA). This work was further supported by the ANR project VAHIA (Grant ANR-12-JS08-0001-01) of the French Agence Nationale de la Recherche.

References

- [1] G. Danger, F. Orthous-Daunay, P. de Marcellus, P. Modica, V. Vuitton, F. Duvernay, L. Flandinet, L. Le Sergeant d’Hendecourt, R. Thissen, T. Chiavassa, *Geochim. Cosmochim. Acta* **2013**, *118*, 184–201.
- [2] G. Danger, A. Fresneau, N. Abou Mrad, P. de Marcellus, F. Duvernay, L. Le Sergeant d’Hendecourt, R. Thissen, T. Chiavassa, *Geochim. Cosmochim. Acta* **2015**, *Submitted*.