

The microphysical formation process of mesospheric clouds on Earth and Mars

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Abstract

Mesospheric water ice clouds are a phenomenon frequently occurring on Earth at mid to high latitudes during summer. Surprisingly, similar clouds consisting of CO₂ ice were also detected at low latitudes in the mesosphere of Mars. Both types of clouds are believed to form by heterogeneous nucleation on nanometer size Meteoric Smoke Particles (MSPs). However, describing cloud formation is still flawed by high uncertainties due to a lack of experimental data regarding the nucleation process. We present measurements on heterogeneous nucleation and subsequent growth of H₂O and CO₂ on sub 3nm silica, iron oxide and mixed iron silicates as analogues for MSPs. The experiments were performed at conditions which are relevant to the mesosphere of Earth and Mars. We show measured desorption energies and critical saturations of H₂O and CO₂ for the different particle materials. Using Classical Nucleation Theory (CNT), we infer the contact parameter resulting in a full parameterization of the nucleation process. We conclude that the same parameterization may also be valid for Martian Dust Particles [MDPs] and predict conditions at which H₂O and CO₂ clouds form in the mesospheres of Earth and Mars. The results are discussed in comparison with atmospheric observations.

1. Introduction

Mesospheric water ice clouds, often referred to as Noctilucent Clouds (NLCs), are a phenomenon frequently occurring on Earth at mid to high latitudes during summer [e.g. 5, 10]. Similar clouds were also detected at low latitudes in the mesosphere of Mars. These clouds mainly occur during post- and pre-aphelion season [e.g. 2, 7, 8]. In contrast to Earth, the majority of the martian mesospheric clouds consist of CO₂ ice. Both types of clouds have in common that they are believed to form by heterogeneous

nucleation on nanometer size Meteoric Smoke Particles (MSPs) during gravity wave induced supersaturated conditions [e.g. 1, 4, 11]. If the conditions needed for cloud formation would be well known, cloud observations could be used as a tracer for the thermodynamic conditions of the observed region. Unfortunately, predictions for cloud formation still struggle with high uncertainties due to a lack of experimental data regarding the nucleation process at realistic mesospheric conditions of Earth and Mars and regarding realistic nuclei materials.

Recently, we introduced a novel experimental setup [3, 6] specifically designed for the investigation of adsorption, nucleation and subsequent depositional growth of H₂O and CO₂ on nanoparticles with MSP material compositions. With the new setup we are able to determine the desorption energy and contact parameter which are key parameters describing the nucleation ability of the aerosol particles at realistic mesospheric conditions of Earth and Mars [3, 9].

2. Experimental Method

Singly charged nanometer size MSP analogues are produced in a microwave plasma particle source. The particles enter into the TRAPS vacuum system [6] in which we are able to select particles with radii between 1 and 3 nm. The size selected particles are then accumulated in the novel MICE-ion trap [3]. In MICE, the particles are exposed to realistic mesospheric conditions in terms of background pressure and temperature as well as CO₂ and H₂O vapor concentrations. From MICE we extract a small amount of the trapped particle population at periodic time steps and analyze their mass distribution with a time-of-flight mass spectrometer. As a result, we observe the adsorption process of CO₂ and H₂O on the MSP analogues. Depending on the conditions in MICE, adsorption is followed by nucleation and subsequent growth of CO₂ and H₂O ice particles.

3. Results and Discussion

Figure 1 shows an exemplary series of H₂O deposition on 2.6nm Silica particles. Here, the H₂O concentration has been kept constant to 5E15 m⁻³ and the particle mass is shown as a function of trapping time for 4 different particle temperatures.

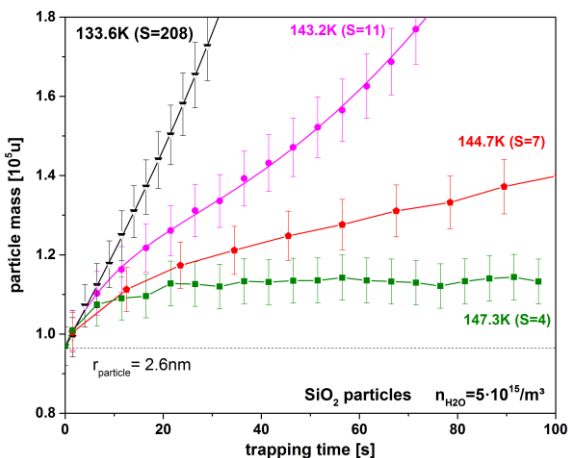


Figure 1: Exemplary dataset of an H₂O deposition experiment on 2.6nm SiO₂ particles used for desorption energy (green curve), critical saturation (red curve) and growth rate (black curve) analysis.

The green curve at a particle temperature of 147.3K and a Saturation of S=4 represents a typical measurement below the nucleation threshold. The increase in mass reflects the adsorption/desorption equilibrium process of H₂O molecules on the surface of the particles under these conditions. From measurements like this we determine the desorption energies for the investigated vapors and nuclei materials. By decreasing the particle temperature and thereby increasing S we reach conditions at which nucleation and further mass aggregation takes place (red curve, T=144.7, S=7). By observing this transition, we are able to experimentally determine the critical saturations needed for cloud formation. Fitting nucleation theory to the determined critical saturations gives us the contact parameter of the particle material. In addition, measurements at high S values (black curve T=133.6K, S=208) allow us to investigate particle growth rates.

Overall, we present a full parameterization of the nucleation process of CO₂ ice between 64K and 74K and of H₂O ice between 128K and 155K on pure iron oxide, silica and mixed iron silicates as analogues for MSPs. We will discuss the differences in the nucleation ability of the materials in respect to CO₂

and H₂O ice and will show that the same parameterization may also be valid for the nucleation on Martian Dust Particles. Using this parameterization we will deduce the Saturations at which H₂O and CO₂ cloud formation is activated at conditions relevant to the mesosphere of Earth and Mars and compare the results to observations.

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