



## Atomic-scale mixing between MgO and H<sub>2</sub>O in the deep interiors of water-rich planets

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Astrophysical surveys so far have suggested that water-rich planets could be common [1] (including Uranus and Neptune in our Solar System). In the conventional interior model of water-rich planets, it has been assumed to have separate layers of atmosphere, ice/fluid, rocky mantle and metallic core [2]. However, recent studies have proposed the existence of heavy elements in the ice/fluid layer of Uranus, challenging the conventional view [3]. In addition, chemical interaction and thermodynamic processes of major rock-forming minerals at the H<sub>2</sub>O-rock interface conditions of the water-rich planetary interiors have been scarcely explored.

We have performed laser-heated diamond-anvil cell experiments on two rock-forming minerals, olivine ((Mg<sub>0.9</sub>,Fe<sub>0.1</sub>)<sub>2</sub>SiO<sub>4</sub>) and ferropericlase ((Mg<sub>0.9</sub>,Fe<sub>0.1</sub>)O), in water at the pressure and temperature conditions expected for the water-rich planets. During laser-heating, we collected X-ray diffraction (XRD) data at beamlines 13-IDD of GSECARS at APS and P02.2, the ECB of PETRA III at DESY. Our dataset covers pressures between 20 and 80 GPa. After high-pressure and high-temperature experiments, we conducted chemical and textural analysis using focused ion beam (FIB) and scanning electron microscope (SEM) at Yonsei University.

During laser-heating, Si-rich high-pressure phases were formed, such as bridgmanite ((Mg,Fe)SiO<sub>3</sub>) and stishovite (SiO<sub>2</sub>), from the high Mg/Si ratio of starting composition (olivine). The formation of Si-rich phases from Mg-rich starting composition suggests dissolve of MgO into H<sub>2</sub>O-liquid during laser-heating at high-pressures. This was also found for (Mg<sub>0.9</sub>,Fe<sub>0.1</sub>)O ferropericlase starting material. The intensity of the diffraction peak of ferropericlase was dramatically decreased at high-pressure and high-temperature conditions, which indicates that (Mg<sub>0.9</sub>,Fe<sub>0.1</sub>)O is soluble in H<sub>2</sub>O-liquid. From chemical analysis, we found the dome-like structures, which showed that domes are Mg-rich and below the domes is Si-rich. Between Mg-rich and Si-rich regions, porous structures (almost empty) were positioned, meaning that MgO-rich fluid existed at high-pressure and high-temperature conditions. In summary, the textural and chemical analysis combined with XRD data indicates a selective leaching of MgO preferentially from silicate during laser heating, making MgO-dissolved in high-temperature fluid, which peaks between 20 and 40 GPa and above 1,500 K [4].

For water-rich planets with 1–6 Earth masses, the chemical reaction at the deep H<sub>2</sub>O–rock interface would lead to high concentrations of MgO in the H<sub>2</sub>O layer. For Uranus and Neptune, our experiments indicate that the top ~3% of the H<sub>2</sub>O layer of them, the pressure and temperature conditions of which have been achieved in this study, would have a large storage capacity for MgO. If an early dynamic process enables the H<sub>2</sub>O–rock reaction, the topmost H<sub>2</sub>O layer may be rich in MgO, possibly affecting the thermal history of the planet.

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